INTRODUCTION

The Fischer-Tropsch synthesis may be defined as a hydrogenation of carbon monoxide producing higher hydrocarbons and oxygenated organic molecules that have predominantly straight carbon chains, at least in the range C₄ to C₇. This definition eliminates related processes, such as the higher alcohol and iso syntheses.

The Fischer-Tropsch synthesis is a composite of many reactions producing a variety of moleof many reactions producing a variety of more-cules of different types and carbon numbers. The catalyst and its environment during syn-thesis are no less complicated, for the pores are at least partly filled with high-molecular-weight hydrocarbons that are liquid at synthesis temperatures. In addition, carbides, oxides, elemental carbon, and carbonyls are formed to varying degrees, resulting not only in changes in the phases present but also in the physical structure of the catalyst.

This paper considers several physicochemical aspects of the synthesis. It is based in part on three recent reviews. 4-6 For more details on studies of the Fischer-Tropsch synthesis by

the Bureau of Mines, the reader is referred to various references, 7-13 as well as to papers cited in the text.

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