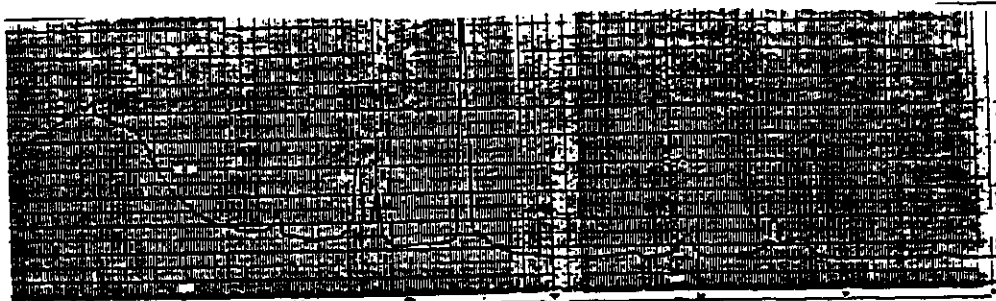
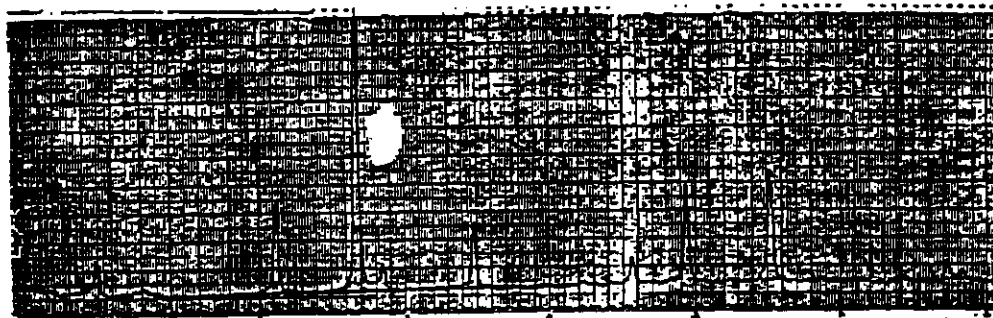


APPENDIX E

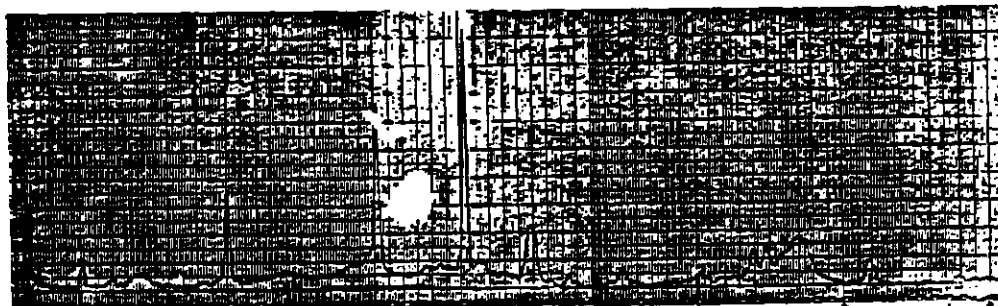
SAMPLE XRD EXPERIMENT
AND
LINE BROADENING CALCULATION



a. $\text{Co}_3\text{O}_4/\text{Al}_2\text{O}_3$ Catalyst Used in This Study.



b. Baker Cobalt Oxide/MCB Activated Alumina.



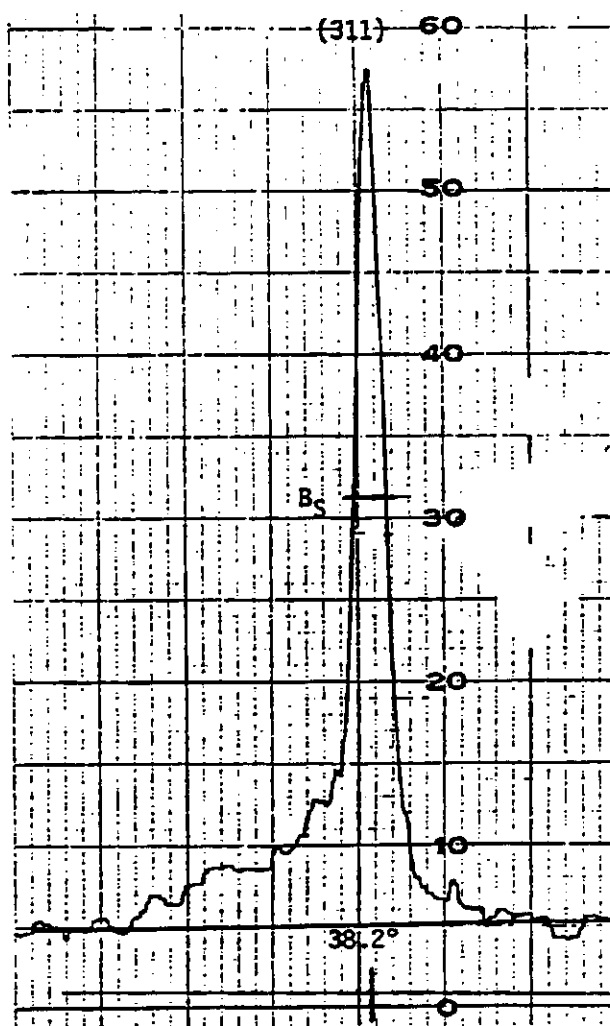
c. ALCOA F-1 Activated Alumina (150 mesh).

Figure E.1. XRD Analysis Patterns of Catalyst Samples.

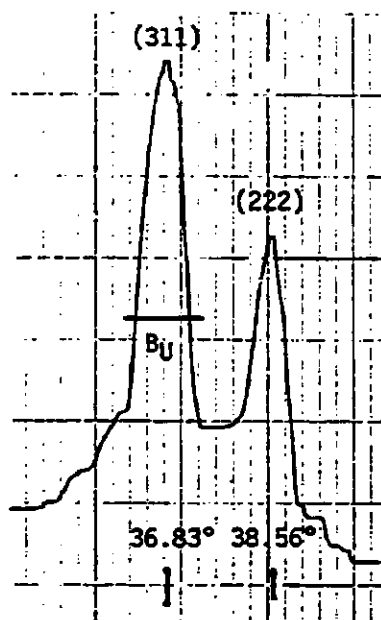
9-418

d	2.44	1.43	2.86	4.669	Co_3O_4	CORALY (II,III) Oxide				
$1/\lambda$	100	45	40	20						
Ref. CORALY A 1-7089 Filter Fe Ni Cut off $1/\lambda$ DIFFRACTION Ref. Mat. Bur. STANDARDS CIRC. 539 2 (1959)										
Syn. Cunit	SQ. $0.7 - \text{FeNi}$					$d \text{ \AA}$	$1/\lambda$	hkl	$d \text{ \AA}$	$1/\lambda$
a 0.014	b	c	α	β	γ	4.669	20	111		
	β	γ	δ	ϵ	ζ	2.860	40	220		
Ref. Ind.						2.438	100	311		
						2.333	11	222		
						2.021	25	400		
						1.650	11	422		
						1.556	35	511		
						1.423	45	440		
						1.278	5	620		
						1.233	11	533		
2θ	θ	ω	ϕ	ψ	Sign	1.219	7	622		
Ref					Color Black	1.167	3	444		
						1.132	3	711		
						1.040	7	642		
						1.052	15	731		
						1.010	7	800		
						0.952	5	022		
						.933	15	751		
						.927	5	662		
SAMPLE PREPARED AT NBS BY HEATING CORALY FLUORINE TO 1000°C FOR 24 HOURS. SYNTHOGRAPHIC ANALYSIS SHOWS 1.0- 0.1% Ca, Ni, 0.1-1.0% EACH OF AL, Fe, Mn, Si, 0.001-0.01% EACH OF Ba, Cu, Mg, 0.001-0.001% OF Zn. PATTERN MADE AT 24°C . SPINEL TYPE STRUCTURE. REPLACES 1-1152										

Figure E.2. JCPDS' Standard X-Ray Pattern (19-418).



b. ALCOA F-1 Activated Alumina.



a. $\text{Co}_3\text{O}_4/\text{Al}_2\text{O}_3$ Catalyst.

Figure E.3. Portions of XRD Patterns Used in Line Broadening Calculation.

Average Crystallite Diameter of Catalyst Calculated by XRD
Line Broadening.

Measured breadths, at half-maximum intensity, of the lines from both the standard and unknown are:

$$B_S = 3/16 \text{ and } B_U = 5/16 \text{ inches.}$$

For chart speed of 2 °/inch, $B_S = \frac{3}{8}^\circ$ and $B_U = \frac{5}{8}^\circ$.

According to Cullity [66], the crystallite broadening B is calculated as follows,

$$B = \sqrt{\left(\frac{5}{8}\right)^2 - \left(\frac{3}{8}\right)^2} = \frac{1}{2}^\circ,$$

which can be converted into $B = 0.008727$ radian.

The wavelength of copper $K\alpha_1$ is $\lambda = 1.54050 \text{ \AA}$, and

Bragg angle for the catalyst sample is $\theta_B = 36.83^\circ$.

Now plug every thing into Scherrer formula, Eqn. (4.1),

$$t = \frac{(0.9)(1.54050 \text{ \AA})}{(0.008727 \text{ radian})(\cos 36.83^\circ)} = 198.491 \text{ \AA}$$

this t is the averaged crystallite diameter of the cobalt/alumina catalyst.

Specific Surface Area of Catalyst Estimated by XRD Line Broadening.

Assume that the particle of catalyst crystallite is either cubic or spherical.

Apparent density of catalyst, ρ , is 1.6667 gm/ml or $1.6667 \times 10^6 \frac{\text{gm}}{\text{m}^3}$.

The specific surface area of the catalyst crystallite, A , can be calculated as follows [128],

$$A = \frac{6}{(1.6667 \times 10^6 \frac{\text{gm}}{\text{m}^3})(198.491 \text{ \AA})(10^{-10} \frac{\text{m}}{\text{\AA}})}$$

$$= 181.36 \frac{\text{m}^2}{\text{gm}}.$$