

Table B17 (continued)

FILE: 1256102D DATA

A1

C5+-END PT	73.85	73.77	74.20	74.59	73.80
ISO/NORMAL MOLE RATIO					
C4	0.0219	0.0220	0.0214	0.0218	0.0223
C5	0.0364	0.0361	0.0376	0.0364	0.0334
C6	0.1134	0.1186	0.1185	0.1207	0.1201
C4=	0.0743	0.0733	0.0741	0.0744	0.0752
PARAFFIN/OLEFIN RATIO					
C3	3.3004	3.4610	3.4624	3.4411	3.5250
C4	2.5612	2.6429	2.6021	2.5870	2.6087
C5	5.1179	5.1539	5.1720	5.0221	5.1048
SCHULZ-FLORY DISTRIB					
ALPHA (EXP(SLOPE))	0.7943	0.8024	0.8512	0.8086	0.8535
RATIO CH4/(1-A)**2	3.4843	3.7642	6.6350	3.8823	7.0010
ALPHA FRM CORRELATION			0.7901		0.7916
ALPHA (EXPTL/CORR)			1.0774		1.0782
WYCH4 FRM CORRELATION			28.7817		28.5474
WYCH4 (EXPTL/CORR)			0.5102		0.5265
LIQ HC COLLECTION					
PHYS. APPEARANCE	OIL WAX	OIL WAX	OIL WAX	OIL WAX	OIL WAX
DENSITY					
N. REFRACTIVE INDEX					
SIMULT'D DISTILATN					
10 WT % @ DEG F			290		291
16			301		305
50			492		510
84			744		791
90			826		877
RANGE(16-84 %)			443		486
WT % @ 420 F			38.10		36.10
WT % @ 700 F			79.50		75.80

V. Run 47 (11617-06) with Catalyst 47 (Co/X₉/X₁₀/TC-103)

This run continued the attempt, begun with Run 44, to replicate the carbon number cut-off at the heavy end of the diesel range, which was first observed with Catalyst 20 and later with Catalyst 31. Formulation of the catalyst was similar to that of Catalyst 31 but with one added step. Both catalysts were activated under hydrogen at 350C, but the hydrogen treatment of this catalyst was followed first by exposure to 350C air, then by a second exposure to 350C hydrogen. The theoretical content of cobalt, X₉ and X₁₀ was, respectively, 7.8, 0.35 and 0.47 percent.

Simulated distillations of the C₅⁺ product are plotted in Figs. B130-131. Carbon number product distributions are plotted in Figs. B132-133. Chromatograms from simulated distillations are reproduced in Figs. B134-135. Detailed material balances appear in Table B18.

The initial activity was extremely low at only 8 percent conversion of the syngas—far lower even than the poor (27 percent) conversion of Catalyst 31 under similar conditions. The run was accordingly terminated after only two samples and 72.5 hours on stream.

Apparently the exposure to air during preparation, which was attempted on the possibility that it might improve the catalyst's activity, had exactly the opposite effect. As to the carbon num-

ber cut-off which was the subject of the test, the run was too short to provide any indication whether or not the result with Catalyst 31 was real.

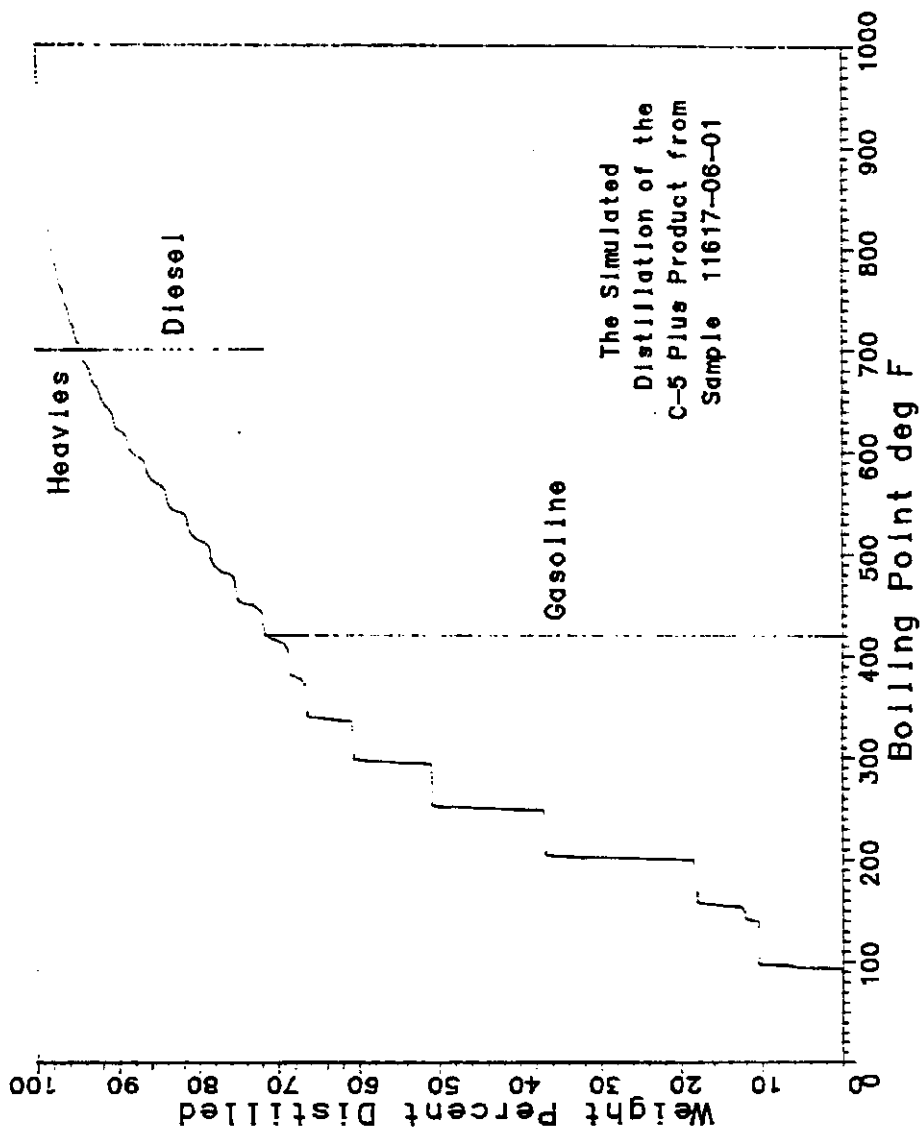


Fig. B130

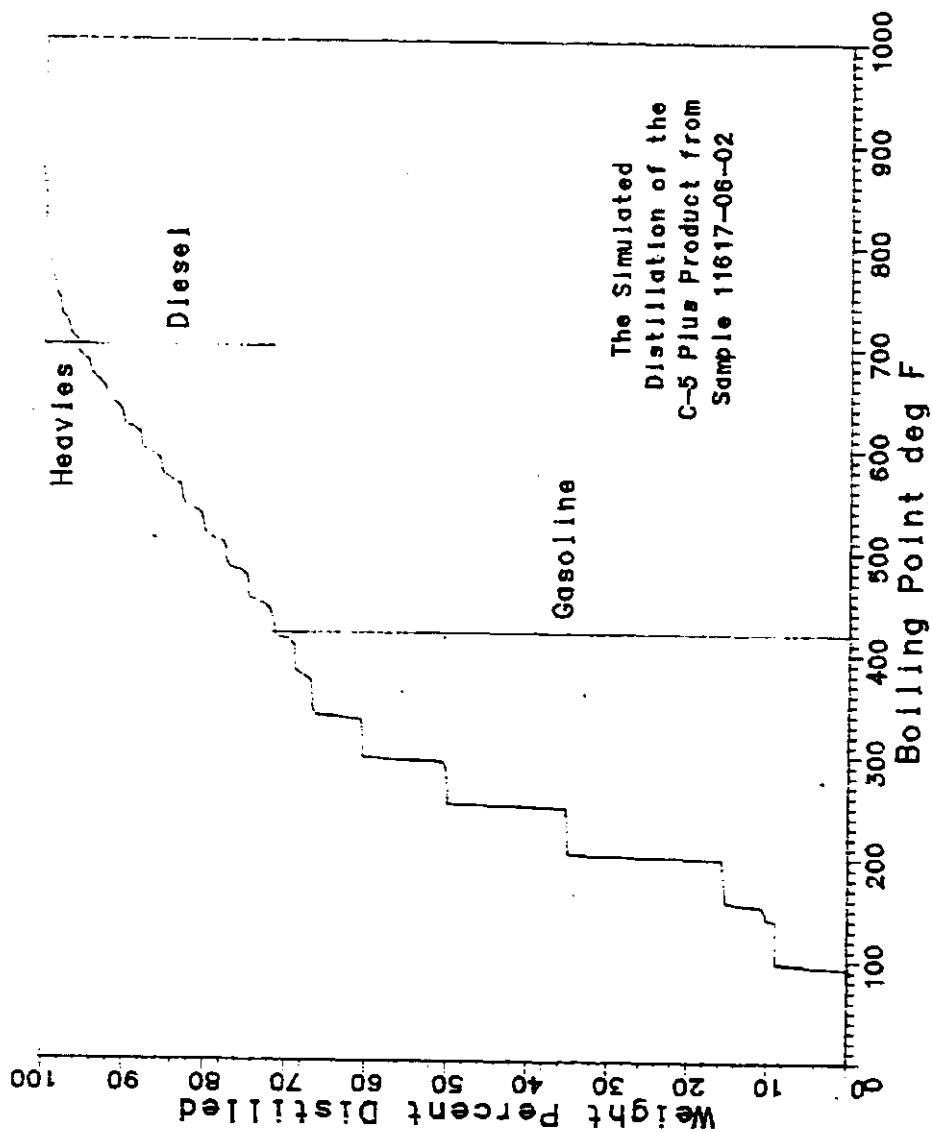


Fig. B131

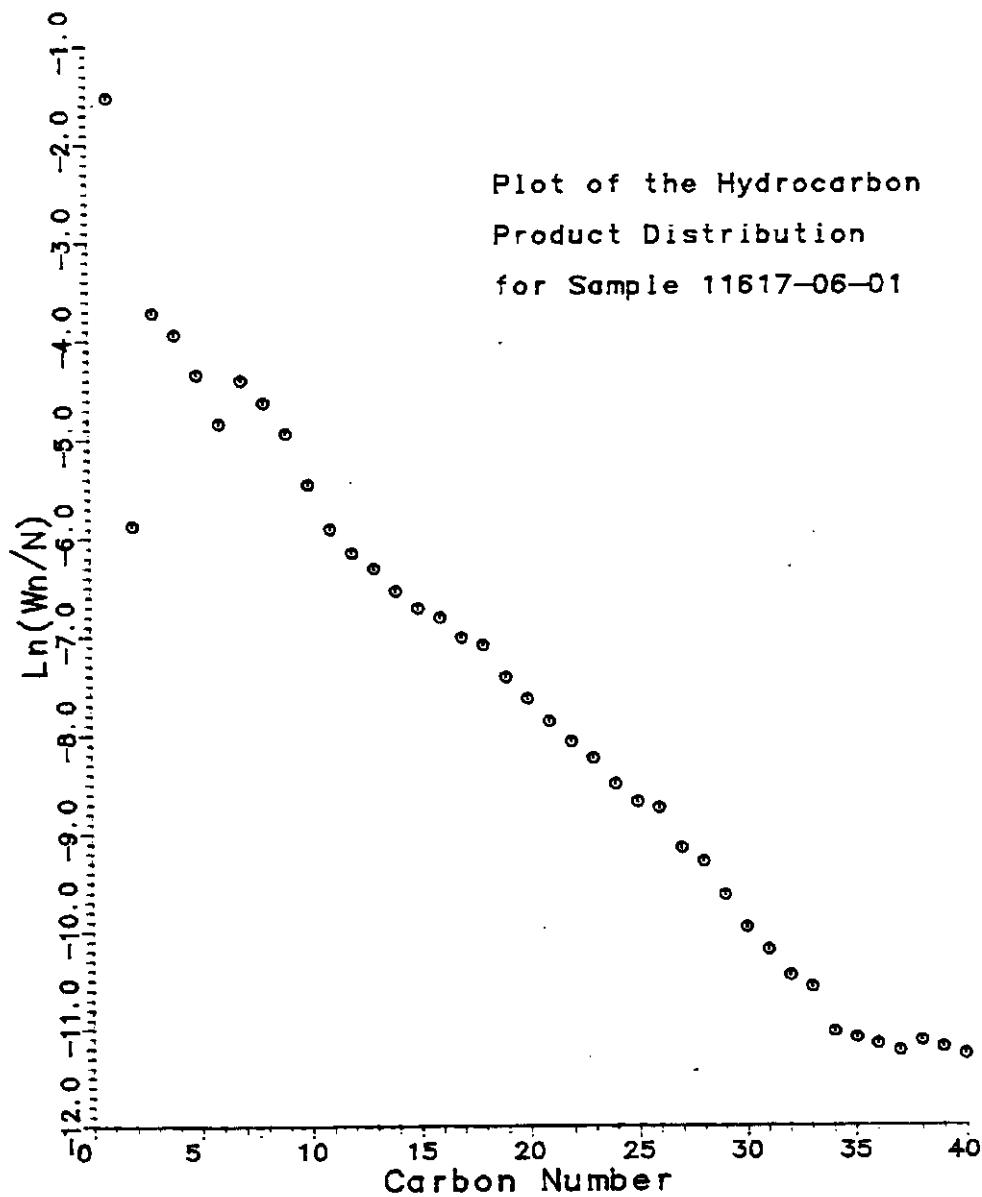


Fig. B132

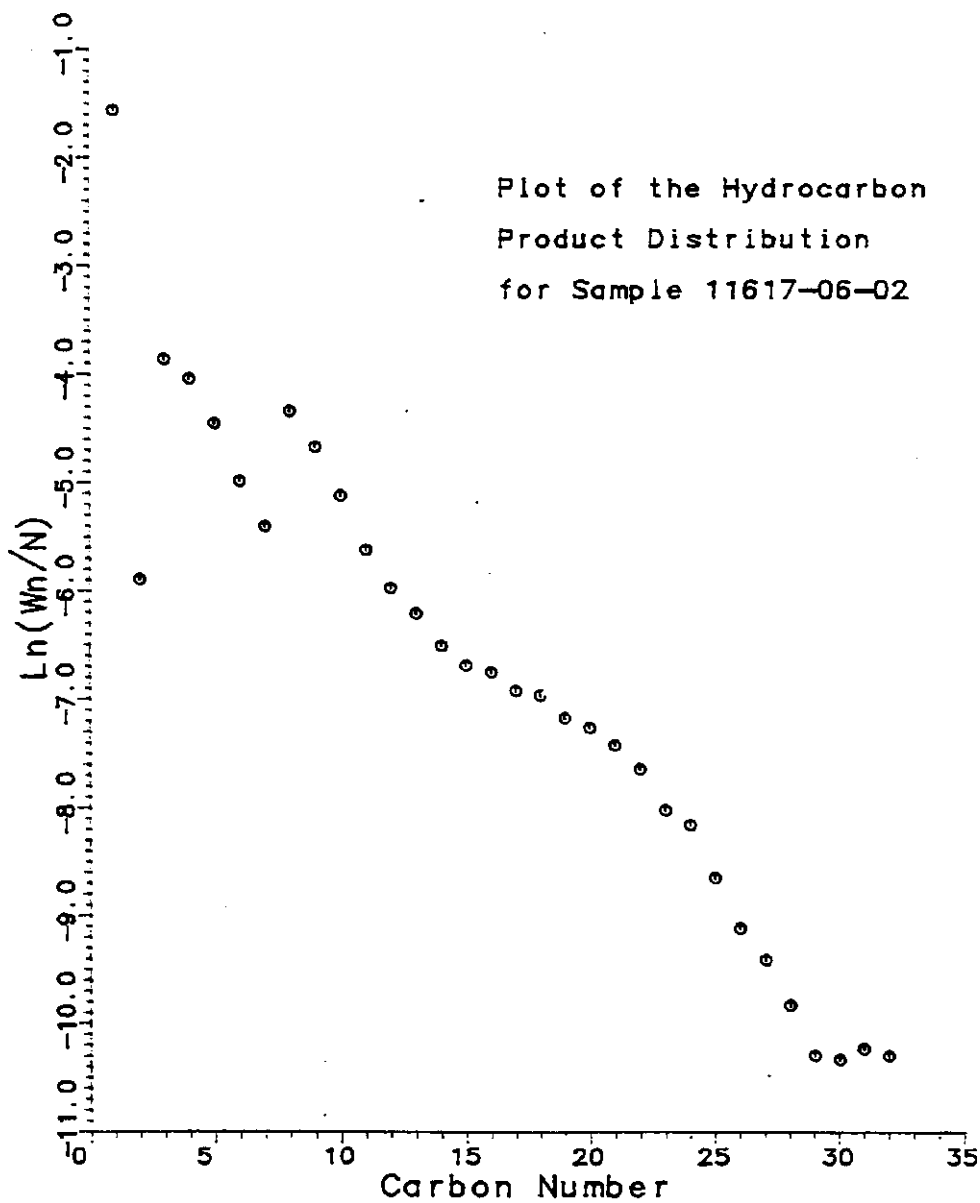


Fig. B133

DATA TEMP NOT READ

ST: 11111 1.11

ST: 1111 1111=3200 1111=3200 1111=40500

ST: 1111 1111=3200 1111=3200 1111=40500

ST: 1111 1111=3200 1111=3200 1111=40500

ST: 1111 1111=3200 1111=3200 1111=40500

ST: 1111 1111

DATA 1111-06-01

Fig. B134

- B182 -

7/25/72 10:00 AM

7/25/72 10:00 AM

7/25/72 10:00 AM 7/25/72 10:00 AM 7/25/72 10:00 AM

7/25/72 10:00 AM 7/25/72 10:00 AM 7/25/72 10:00 AM

7/25/72 10:00 AM 7/25/72 10:00 AM 7/25/72 10:00 AM

7/25/72 10:00 AM 7/25/72 10:00 AM 7/25/72 10:00 AM

7/25/72 10:00 AM

7/25/72 10:00 AM

Fig. B135

- B183 -

Table B18

FILE: 1161706A T5Q4

A1

RESULT OF SYNGAS OPERATION

RUN NO. 11617-06
 CATALYST CO/X9/X10-TC103 250 CC 124.6 G AFTER USE:150.2 G (+25.6 G)
 FEED H2:CO OF 50:50 @ 1260 CC/MN OR 300 GHSV (CAT#12524-17)

RUN & SAMPLE NO. 11617-06-01 617-06-02

FEED H2:CO:AR	50:50: 0	50:50: 0
HRS ON STREAM	26.5	72.5
PRESSURE, PSIG	300.00	300.00
TEMP. C	239.00	239.00
FEED CC/MIN	1260.00	1260.00
HOURS FEEDING	26.50	46.00
EFFLNT GAS LITER	1767.29	3068.20
GM AQUEOUS LAYER	27.52	45.22
GM OIL	4.77	9.26
MATERIAL BALANCE		
GM ATOM CARBON %	91.39	91.20
GM ATOM HYDROGEN %	95.35	95.70
GM ATOM OXYGEN %	94.87	94.31
RATIO CHX/(H2O+CO2)	0.5116	0.5519
RATIO X IN CHX	2.4846	2.4629
USAGE H2/CO PRDPT	3.1194	2.9759
FEED H2/CO FRM EFFLNT	1.0433	1.0493
RESIDUAL H2/CO RATIO	0.9554	0.9635
RATIO CO2/(H2O+CO2)	0.0096	0.0093
K SHIFT IN EFFLNT	0.0093	0.0091
SPECIFIC ACTIVITY SA	0.1608	0.1721
CONVERSION		
ON CO %	4.06	4.26
ON H2 %	12.15	12.09
ON CO+H2 %	8.19	8.27
PRDPT SELECTIVITY, WT %		
CH4	21.93	21.02
C2 HC'S	0.56	0.55
C3H8	2.25	1.96
C3H6=	5.06	4.36
C4H10	2.76	2.53
C4H8=	5.07	4.53
C5H12	2.68	2.39
C5H10=	3.85	3.46
C6H14	4.48	3.83
C6H12= & CYCLO'S	0.27	0.28
C7+ IN GAS	29.22	31.71
LIQ HC'S	21.86	23.36
TOTAL	100.00	100.00
SUB-GROUPING		
C1 -C4	37.63	34.96
C5 -420 F	44.55	45.93
420-700 F	14.36	16.26
700-END PT	3.45	2.85

Table B18 (continued)

FILE: 1161706A TSQ4

A1

CS+-END PT	62.37	65.04
ISO/NORMAL MOLE RATIO		
C4	0.0000	0.0000
C5	0.0000	0.0000
C6	0.2984	0.2832
C4=	0.0000	0.0000
PARAFFIN/OLEFIN RATIO		
C3	0.4247	0.4290
C4	0.5250	0.5399
C5	0.6765	0.6708
SCHULZ-FLORY DISTRBTN		
ALPHA (EXP(SLOPE))	0.8056	0.8044
RATIO CH4/(1-A)**2	5.8027	5.4936
ALPHA FRM CORRELATION	0.8142	0.8138
ALPHA (EXPTL/CORR)	0.9894	0.9884
WYCH4 FRM CORRELATION	20.8474	20.9753
WYCH4 (EXPTL/CORR)	1.0520	1.0024
LIQ HC COLLECTION		
PEYS. APPEARANCE	CLD OIL	CLD OIL
DENSITY		
N, REFRACTIVE INDEX		
SIMULT'D DISTILATN		
10 WT % @ DEG F	381.00	379.00
16	416.00	414.00
50	541.00	543.00
84	699.00	684.00
90	745.00	710.00
RANGE(16-84 %)	283.00	270.00
WT % @ 420 F	18.50	19.20
WT % @ 700 F	84.20	87.80

VI. Run 48 (11617-07) with Catalyst 48 (Co/X₁₁/X₁₂/TC-123)

The purpose of this run was to assess the effect of adding a second promoter, X₁₂, to the Co/X₁₁/TC-123 type of catalyst. When combined with a Co/X₁₁/TC-103 formulation in Catalyst 39, X₁₂ had demonstrated certain desirable properties such as improved activity and lower methane production. The results are to be compared with those of both Catalyst 39 (Co/X₁₁/X₁₂/TC-103) and Catalyst 45 (Co/X₁₁/TC-123).

Formulation of the catalyst was by the same method as for Catalyst 39 except for the replacement of TC-103 by TC-123. The theoretical content of cobalt, X₁₁ and X₁₂ was 7.6, 1.4 and 5.0 percent respectively.

Conversion, product selectivity, isomerization of the pentane, and percent olefins of the C₄'s are plotted against time on stream in Figs. B136-139. Simulated distillations of the C₅⁺ product are plotted in Figs. B140-148. Carbon number product distributions are plotted in Figs. B149-157. Chromatograms from simulated distillations are reproduced in Figs. B158-165. Detailed material balances appear in Tables B19-22.

The catalyst's initial conversion, selectivity, and product quality compared as follows with those of Catalysts 39 and 45. Test conditions were the same for all three: H₂:CO feed ratio 1:1, 300 psig, 240C.

	Catalyst 48 <u>Co/X₁₁/X₁₂/TC-123</u>	Catalyst 39 <u>Co/X₁₁/X₁₂/TC-103</u>	Catalyst 45 <u>Co/X₁₁/TC-123</u>
Conversion, pct	51	51	47
CH ₄ , pct	4.5	4.5	3.5
C ₅ ⁺ , pct	87	87	91
C ₄ olefin/paraffin	1.4	1.8	2.7

Adding X₁₂ to the formulation of Catalyst 45 improved conversion by 4 percentage points, but at the cost of substantially poorer selectivity: higher methane production, lower production of C₅⁺, and much lower olefin content of the C₄ fraction. As compared with Catalyst 39, conversion and selectivity were essentially the same and C₄ olefin content again much lower. All three catalysts showed excellent stability under the test conditions noted.

When the H₂:CO feed ratio was raised to 1.5:1 at 217 hours on stream the conversion rose sharply to about 67 percent, as against about 58 percent with Catalyst 45, and remained essentially unchanged during the 150 hours on stream at this feed ratio. Again, however, the improved conversion and stability were offset by losses in selectivity: methane production rose to about 17 percent and C₅⁺ production dropped to about 70 percent, as against 7 and 86 percent respectively with Catalyst 45. Evidently the X₁₂ additive increases the catalyst's sensitivity in an undesirable fashion to the H₂:CO ratio of the syngas feed.

During the last 150 hours on stream, at an increased pressure of 500 psig, the conversion improved somewhat to about 69 percent, the methane production dropped to about 9 percent, and

the selectivity remained essentially constant. There was, however, a significant loss of stability. Catalyst 45, at the same high pressure, performed in much the same way.

These results demonstrate once again that X_{12} , when added to a formulation promoted with X_{11} , improves the catalyst's activity but also makes it more sensitive, in terms of product selectivity, to the $H_2:CO$ ratio of the feed.

RUN 11617-07

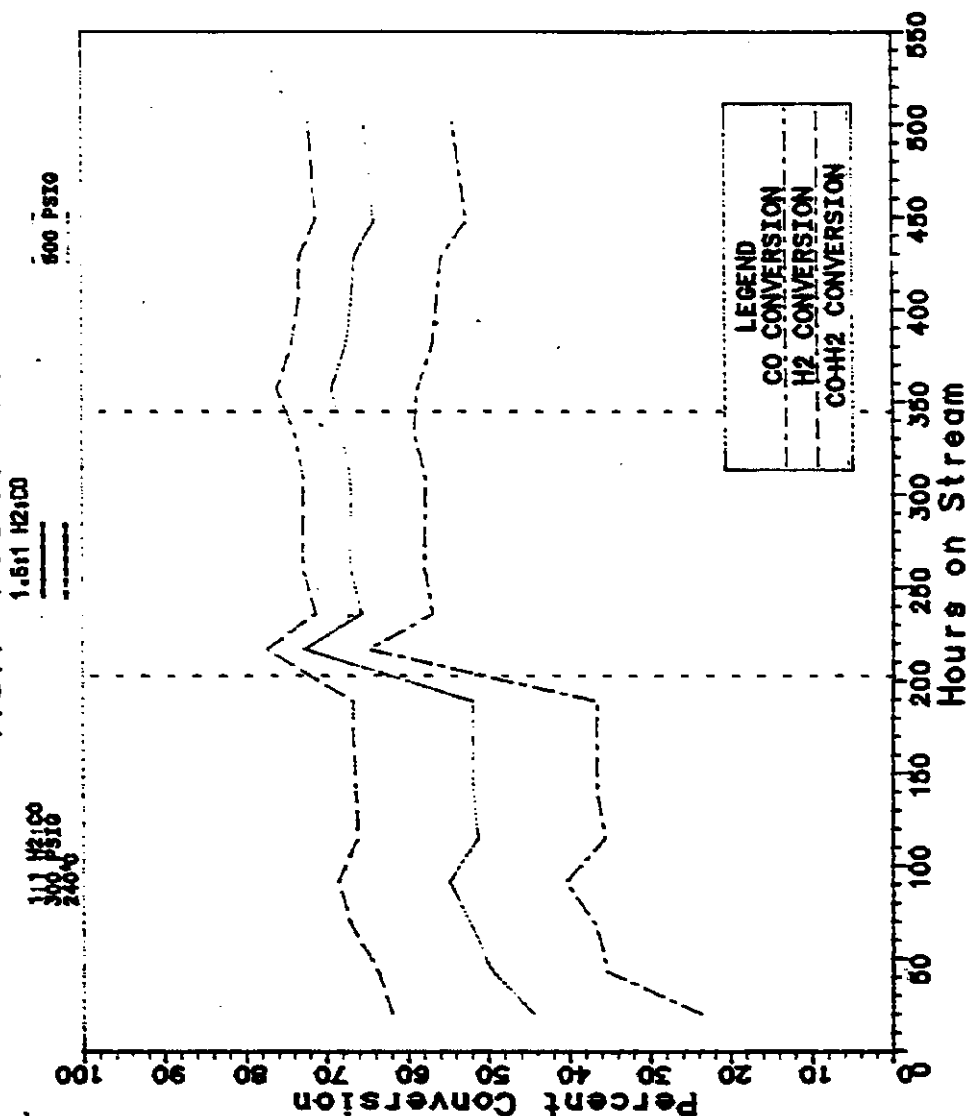


Fig. B136

RUN 11617-07

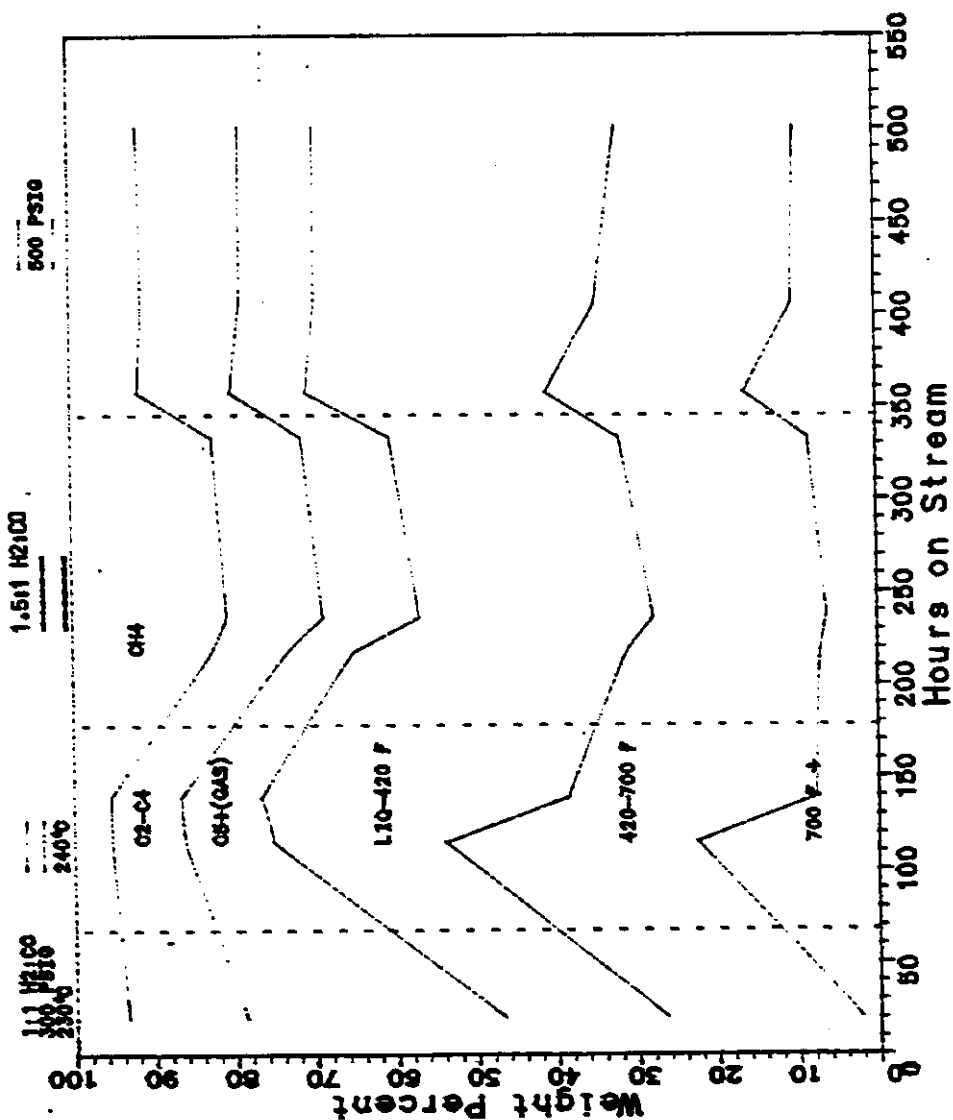


Fig. B137

RUN 11617-07

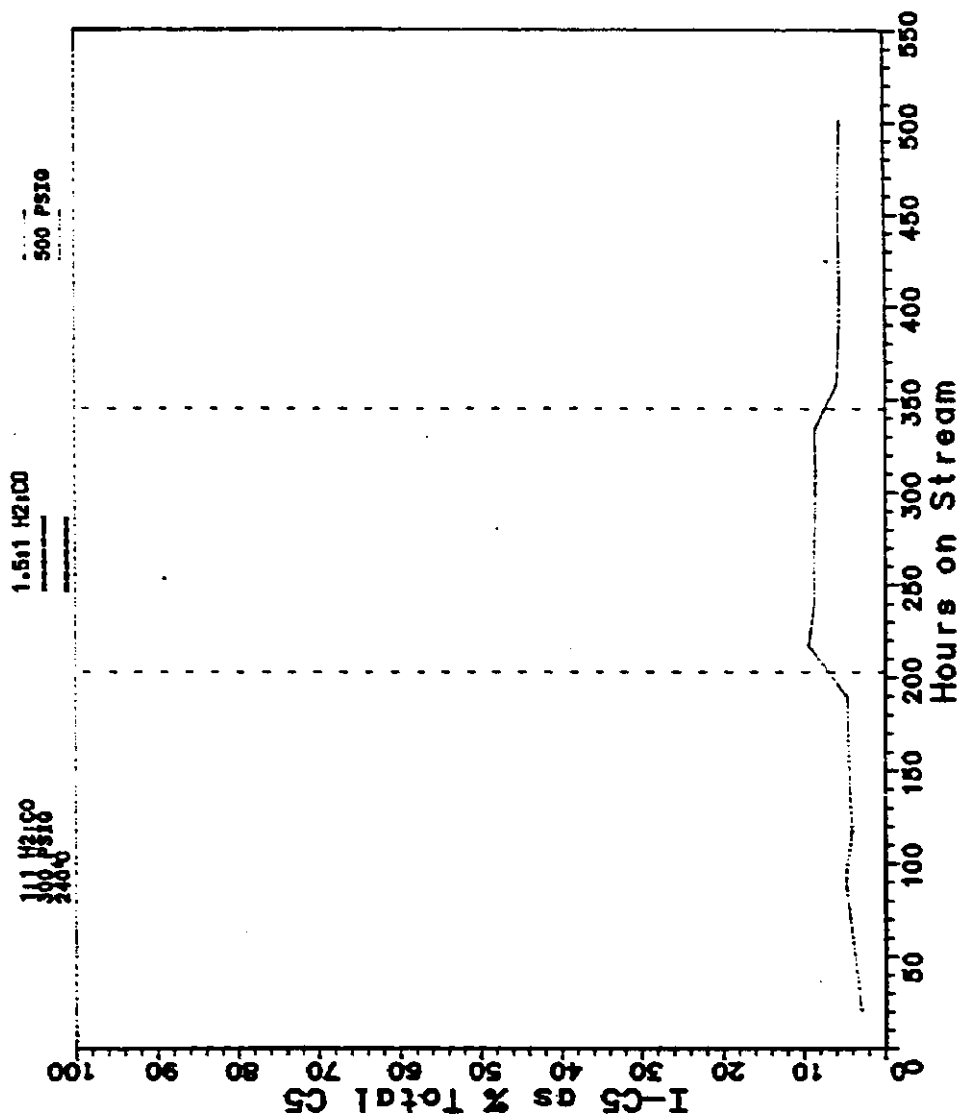


Fig. B138

RUN 11617-07

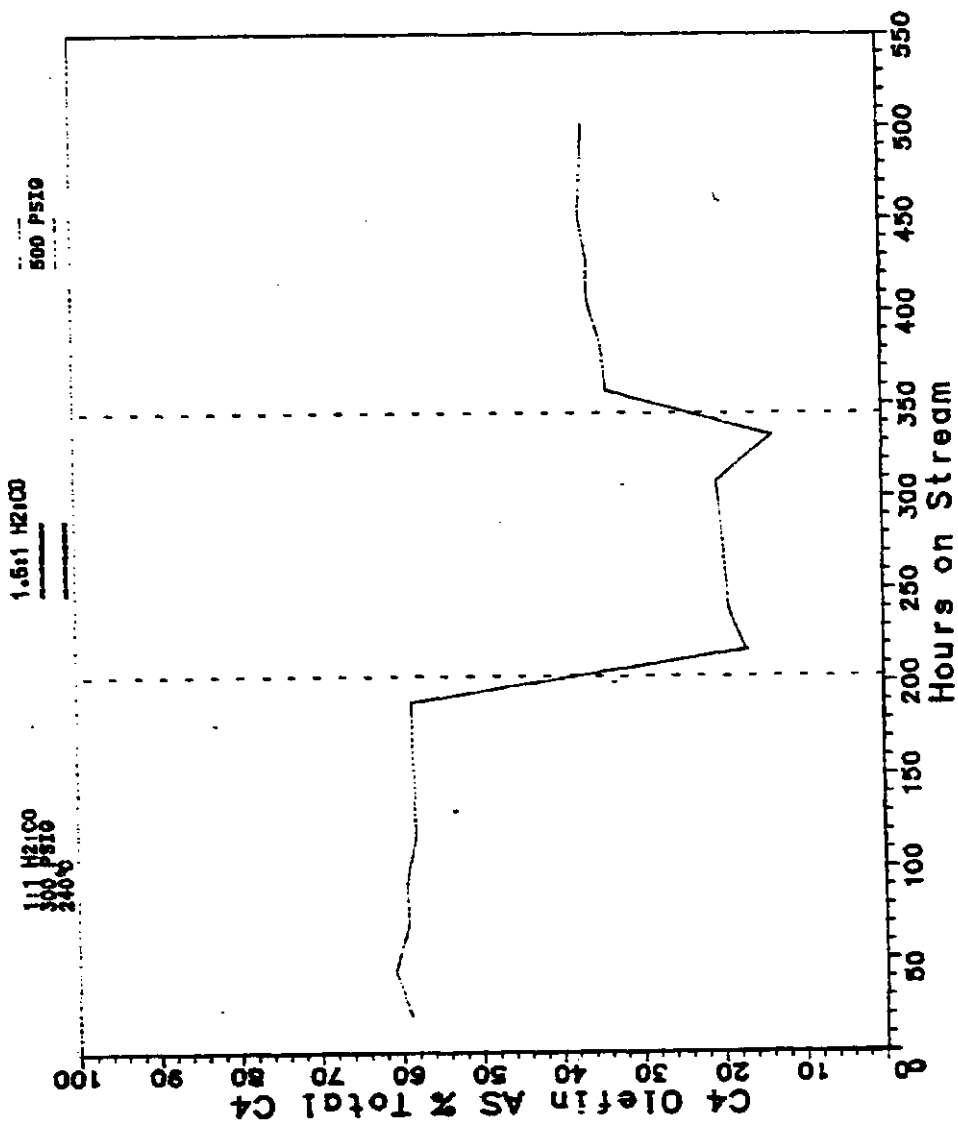


Fig. B139

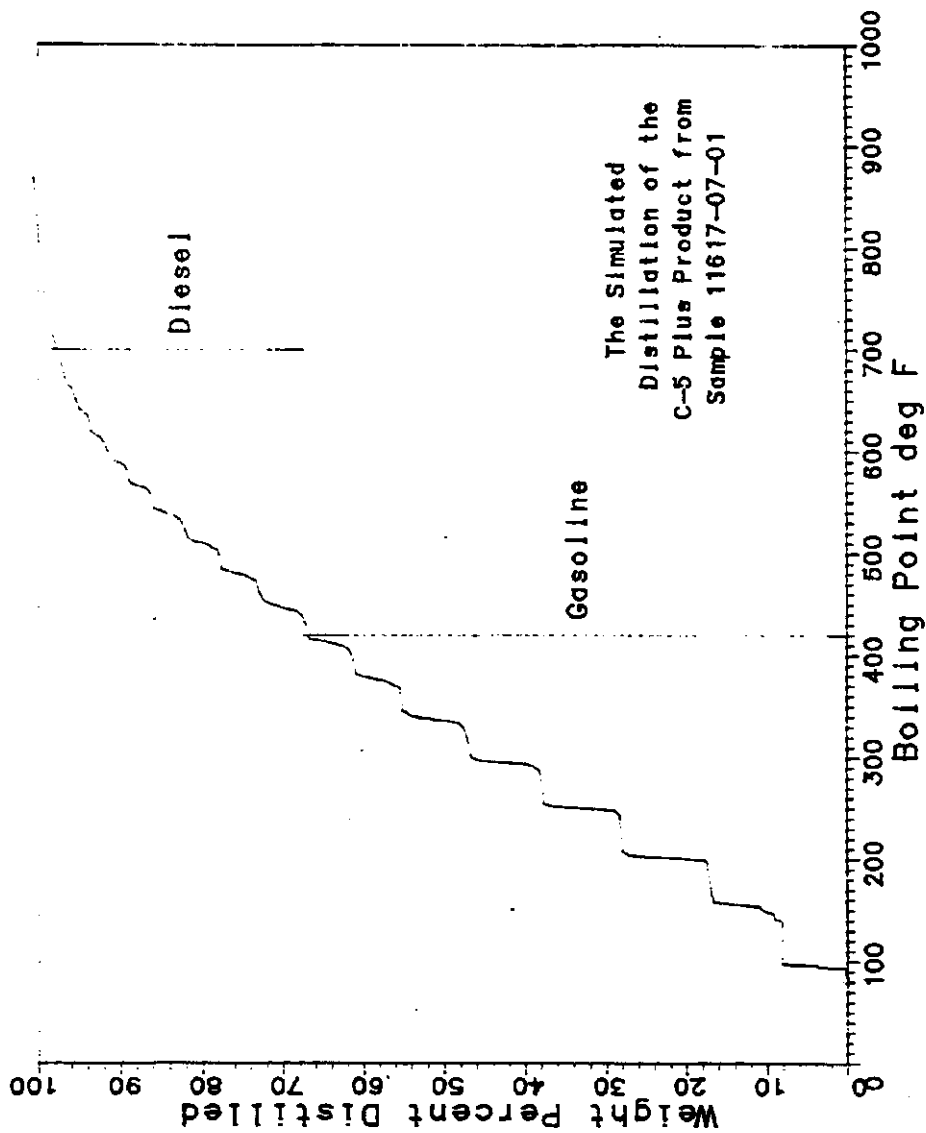


Fig. B140

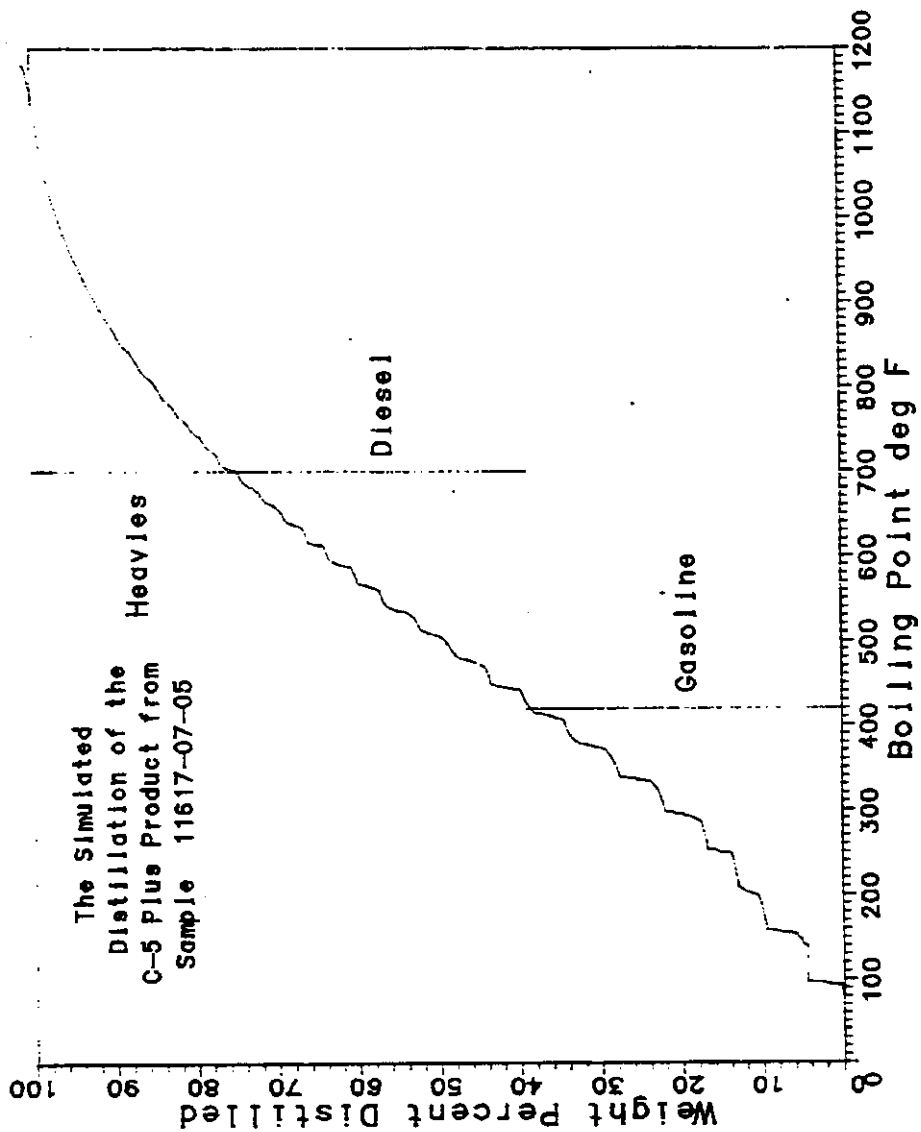


Fig. B141

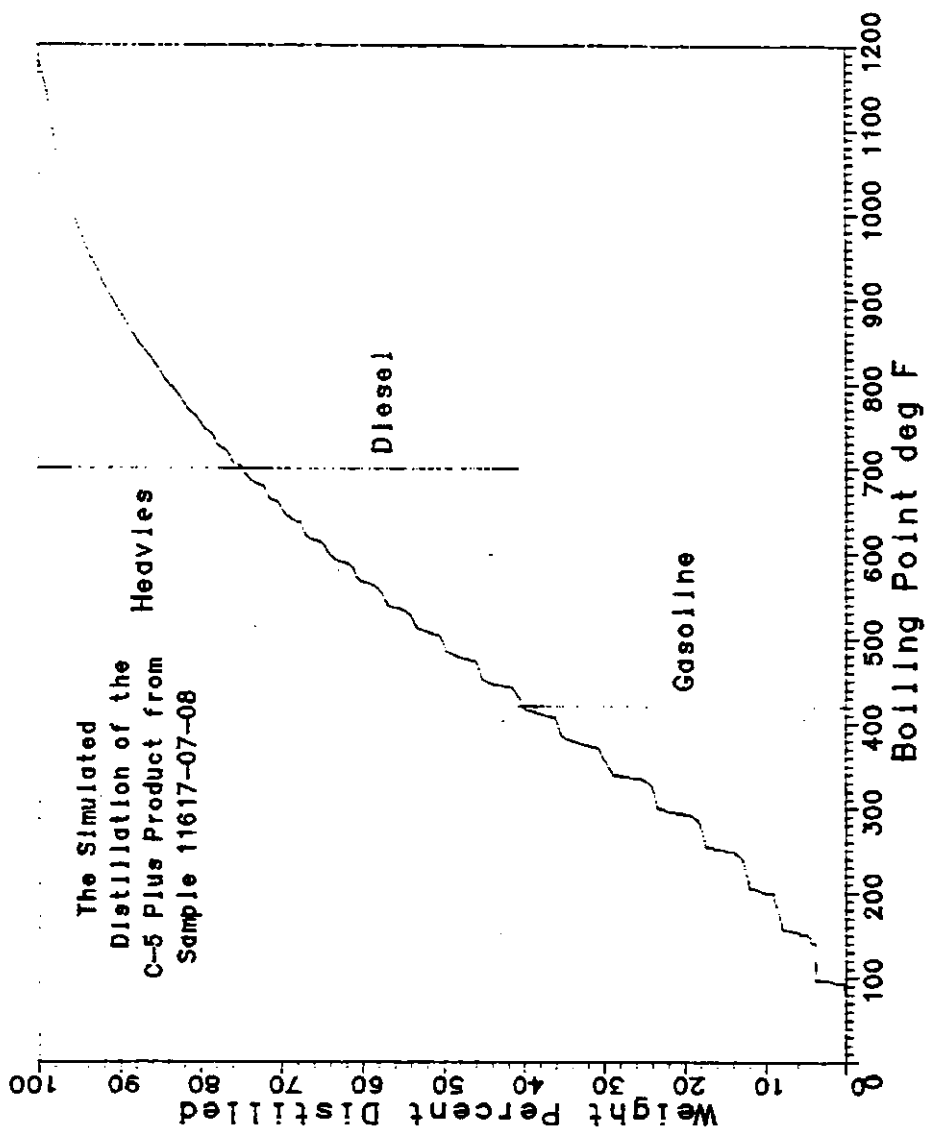


Fig. B142

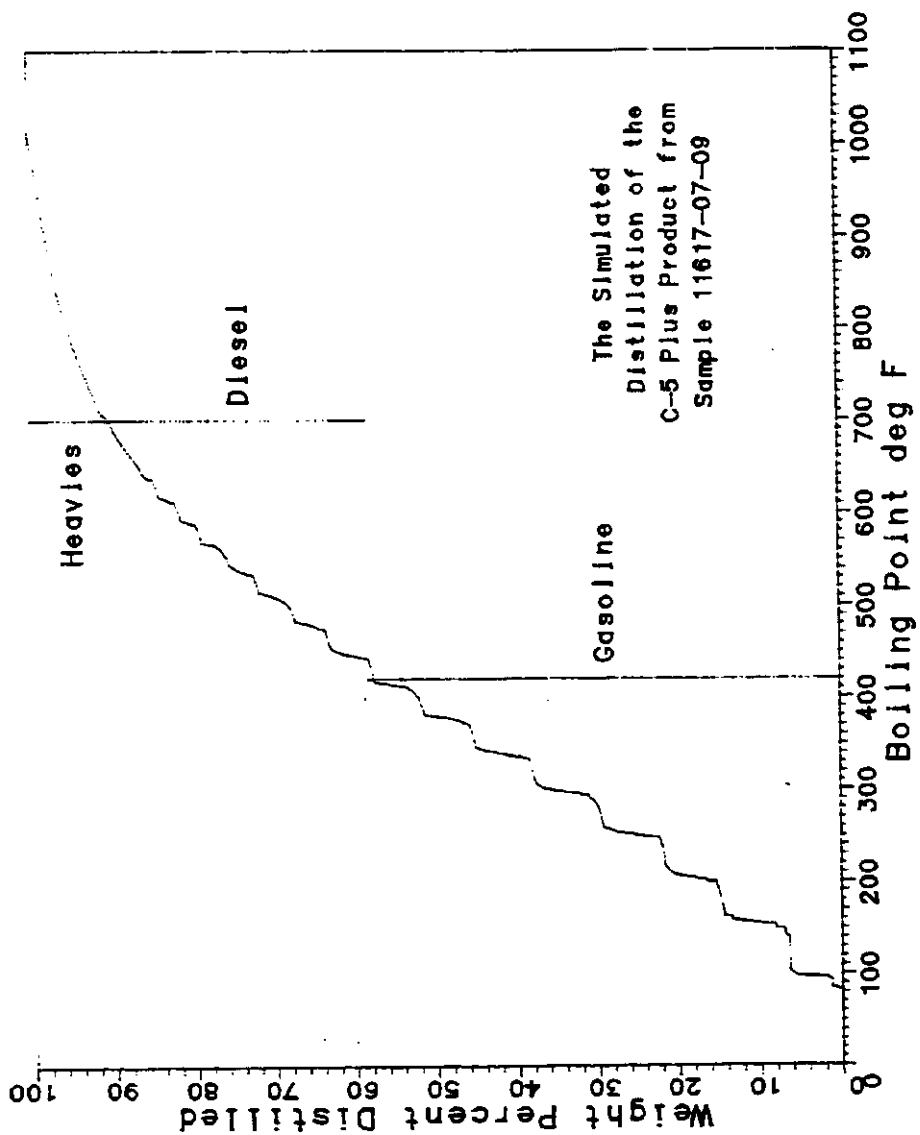


Fig. B143

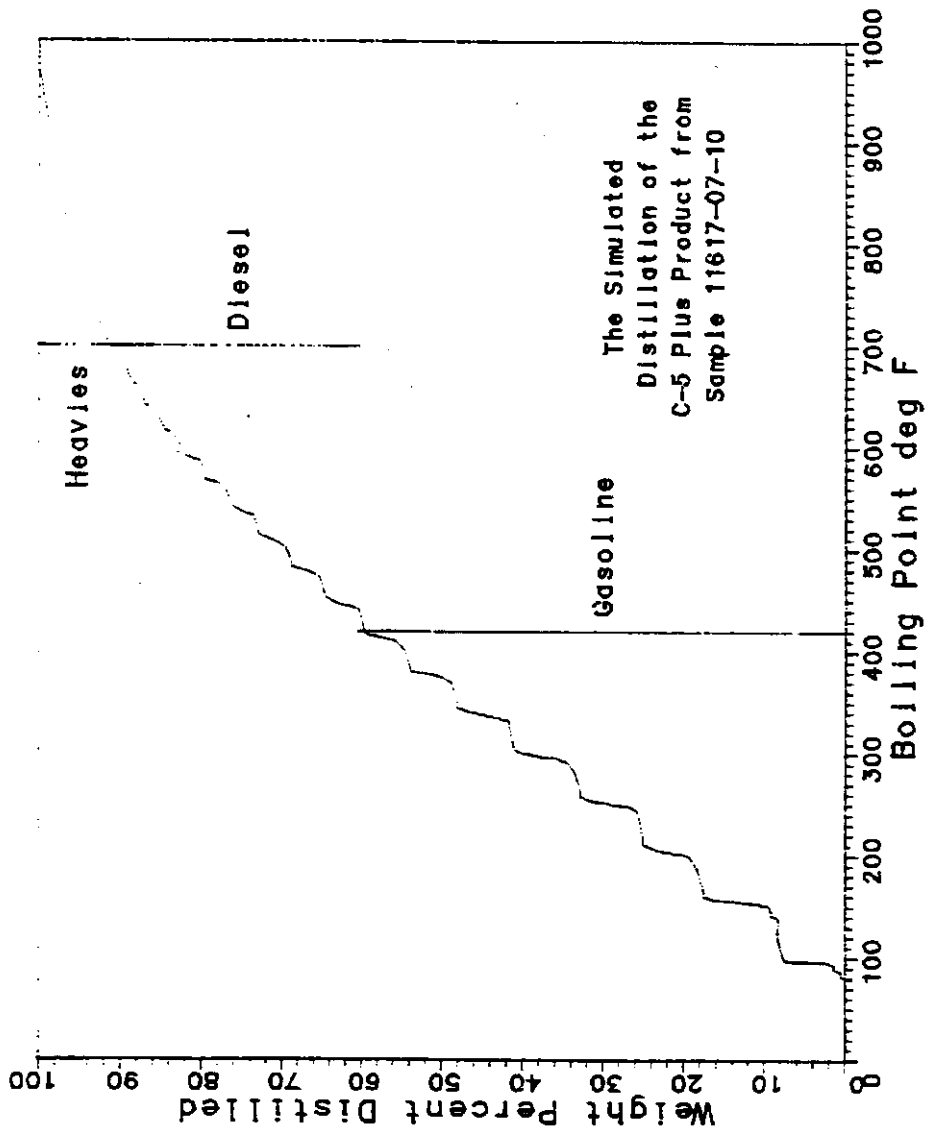


Fig. B144

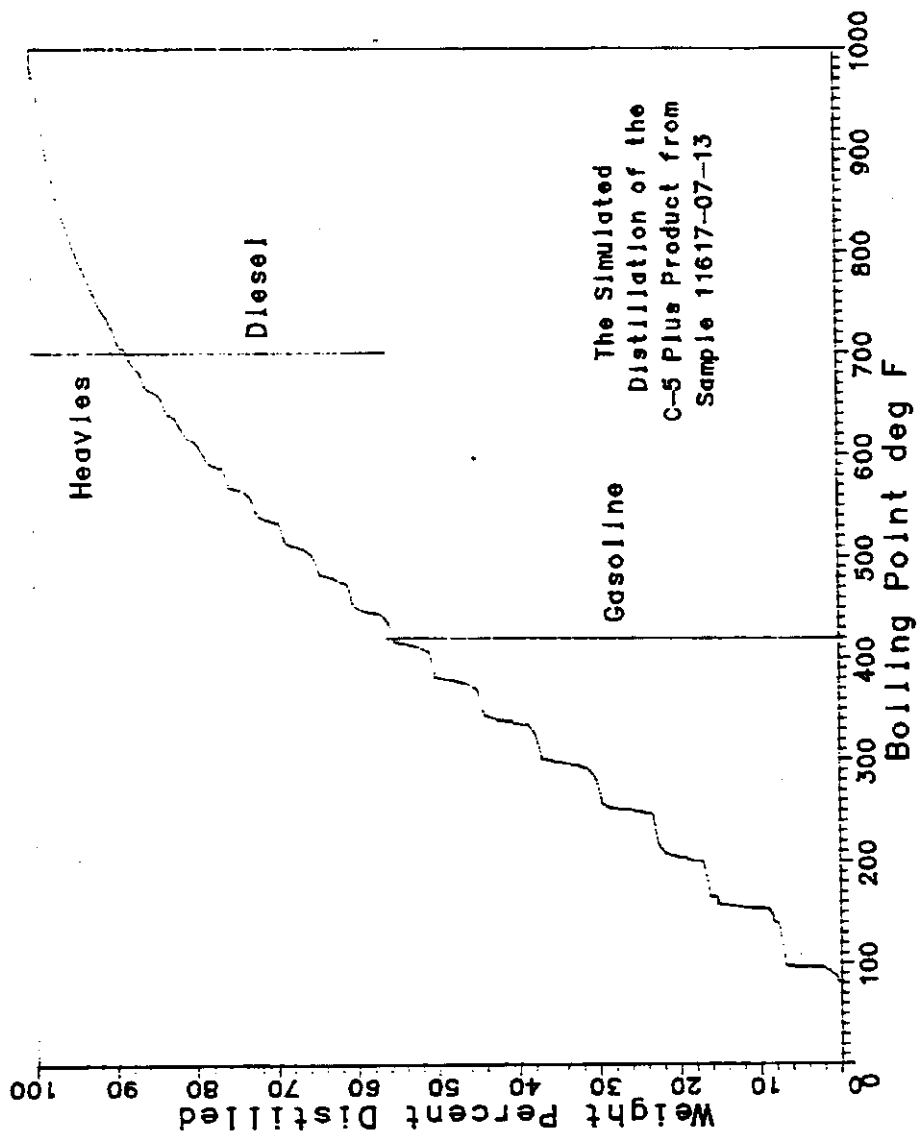


Fig. B145

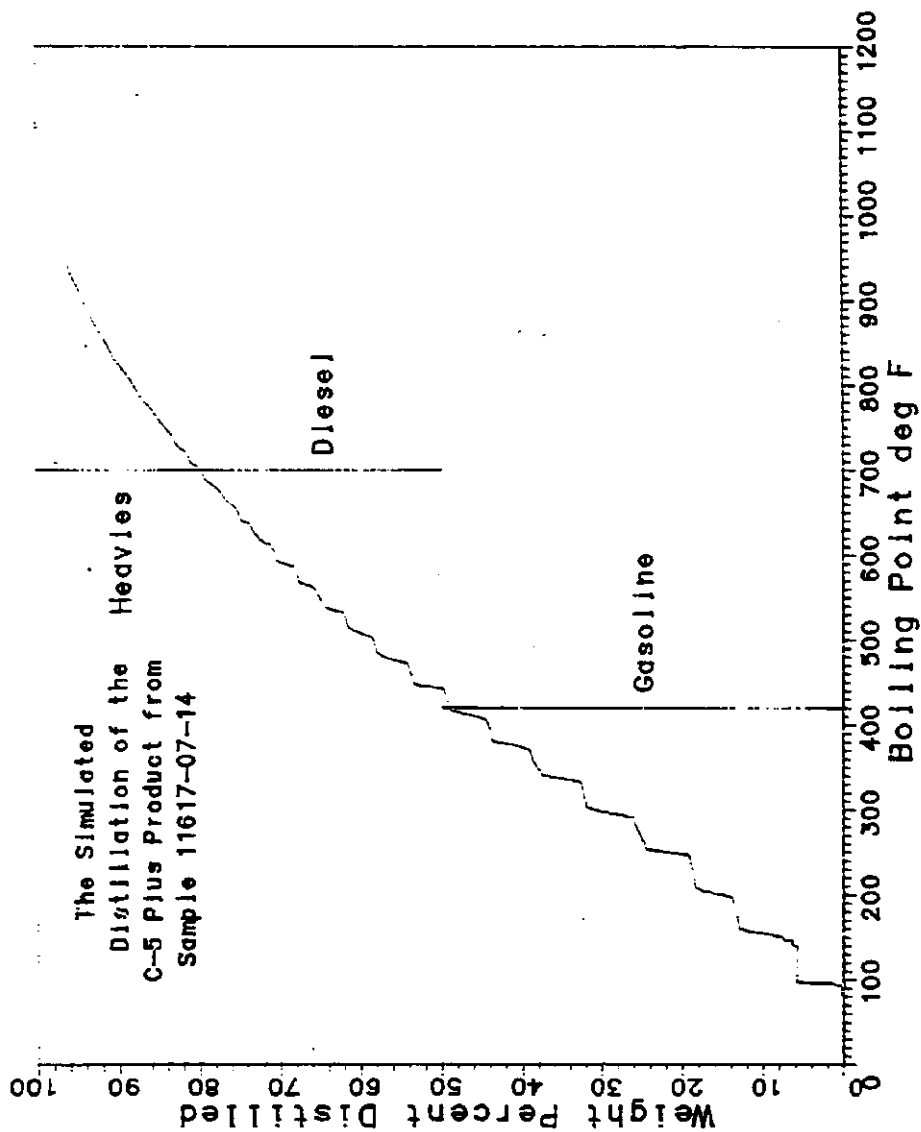


Fig. B146