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Novel Technologies for Gaseous Contaminants Control

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Description

During gasification of a carbonaceous feedstock, some of the fuel-bound contaminants naturally present in carbonaceous materials convert into gaseous impurities, such as H₂S, COS, NH₃, HCN, HCl, and alkali. Research Triangle Institute (RTI) is working on a hybrid process for sulfur removal consisting of a bulk sulfur removal process and a polishing sulfur removal process. The team members actively involved in the bulk sulfur removal research are Membrane DuPont Air Liquide, North Carolina State University, and RTI. The bulk sulfur removal process is being developed with polymer membranes engineered to specifically remove H₂S, CO₂, and H₂O from syngas. In the polishing sulfur removal process, RTI is collaborating with the Prototech Company to develop regenerable ZnO-coated monoliths for removing any remaining sulfur to sub-ppm levels. These monoliths will achieve the same sub-ppm-level sulfur removal available with disposable commercial ZnO-based materials, but unlike these materials the monolith can be regenerated for repeated use. For chloride removal, RTI is working with SRI International for the development of an inexpensive disposable sorbent. Ammonia present in syngas will be removed by adsorption on highsurface-area acidic supports and subsequently desorbed by temperature and/or pressure swings to generate a small NH₃ tail gas stream which will be converted to nitrogen and water with commercially available catalytic oxidation processes. It is projected that these combined technologies could reduce pollutants to near-zero levels while costing about as much, or potentially less than, commercial amine systems.

The following figure shows a schematic of the overall process for removal of sulfur, chlorine, and nitrogen compounds from syngas to ultra-clean part per billion levels. As seen, this approach consists of a number of different modular processes that can be integrated as needed for the different syngas-conditioning requirement for fuel cell, chemical production, or power generation applications. It is contemplated that various individual modules can be used in a "mix and match" fashion depending on the type and level of contaminants and cleanup requirements.



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Raw Syngas Off-Gas Αir Polymer Module CO₂-Rich Tail Gas tion H₂O Sulfur HCI H₂S/CO₂/H₂O Polishing NH₃ Removal Removal H₂S Removal and Sulfur Recovery NH₃ Removal HCI 60-90% H₂S Removal 60-90% CO₂ Removal HoS in Clean Syngas: 60-500 ppby Removal to to <10 ppmy -100% H₂O Removal <10 ppbv

Goals

The goal of this study is to develop technologies for cleaning/conditioning gasifiergenerated syngas to meet contaminant tolerance limits for fuel cell, chemical production, and power generation applications. The specific goals are to develop processes for (1) removal of reduced sulfur species to sub-ppm levels (< 60 ppbv) using a hybrid process consisting of a polymer membrane and a regenerable ZnO-coated monolith, (2) removal of hydrogen chloride vapors to sub-ppm levels (< 10 ppbv)using an inexpensive, high-surface-area material; and (3) removal of NH₃ (< 10 ppmv) with acidic adsorbents followed by conversion of this NH₃ into nitrogen and water.

Benefits

This technology, if successfully developed, could replace conventional systems that are currently used for syngas cleanup and achieve near-zero emissions of all pollutants. The potential cost of this desulfurization technology could be 50% lower than the cost of a conventional Rectisol-based system. As shown in the process configuration, a concentrated CO₂ stream can be obtained, thus allowing effective CO₂ sequestration without any major additional processing costs.

The combination of the bulk desulfurization by membranes with the polishing capability of monoliths provides both a significant cost advantage as well as reduced environmental impact. This process has vast potential to reduce net emissions of sulfur gases without sacrificing any thermal efficiency. Use of the monolith process, if successfully demonstrated, can potentially avoid the pollution implications of Zn recovery as well as offer substantial cost savings in terms of using a regenerable material rather than a disposable sorbent. Successful development of inexpensive chloride removal sorbents may reduce the cost of chloride removal processes that are currently in practice.

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