

LIQUID PHASE METHANOL UPDATE

by

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### ABSTRACT

The U.S. DOE and Air Products and Chemicals, Inc. are now somewhat over halfway through a 42-month R&D program to prove the feasibility of the Liquid Phase Methanol (LPMeOH) technology. The program began in September 1981. Chem Systems Inc., inventor of the LPMeOH technology, is performing as a key subcontractor in the program. Cost sharing participants are DOE, Air Products, Fluor Engineers, Inc. and the Electric Power Research Institute. LPMeOH technology has the potential to be a lower-cost conversion route to methanol-from-coal than current gas-phase processes. Laboratory work to date shows LPMeOH technology particularly suited to coal-derived synthesis gas rich in carbon monoxide because: it is capable of processing feed gas of varying CO and H<sub>2</sub> contents; it can achieve high CO conversion per pass; and it permits effective recovery of heat liberated during reaction. In this program, a DOE-owned skid mounted process development unit was transferred from Chicago, refurbished, expanded for service as the LPMeOH Process Development Unit (PDU), and has been relocated to Air Products' LaPorte, TX facility. Synthesis feed gas from the facility will be used to test the unit beginning late this year. A liquid-fluidized (ebullated bed) mode and a liquid-entrained (slurry) mode will be tested. The PDU operation is supported by an extensive 42-month laboratory program, conducted principally at Chem Systems' labs, with a complementary research effort at Air Products. Chem Systems is providing technical management for the project. Air Products is providing overall program management and is responsible for engineering design, construction, and operation.

In the bench scale effort of the laboratory support program, new baseline data was established for gas phase and liquid phase methanol synthesis. CO conversion and methanol specific productivity in the liquid phase are comparable to the gas phase at matching conditions. Optimum in-situ reduction conditions for slurry powders have been identified which give performance very similar to gas phase reduction. Several commercial catalyst powders achieve acceptable liquid phase performance. Of the 28 new slurry catalysts prepared in this program, 3 have recently advanced from gas phase to liquid phase screening. Autoclave life tests were initially affected by the trace contaminants iron carbonyl and organic chloride; these are strong poisons of methanol synthesis catalysts. Tests in support of the LaPorte PDU design demonstrated that hot (350°C) alpha-alumina provided an effective guard bed against iron carbonyl. In the laboratory systems, ambient activated carbon has been effective. With strong guard measures in place, liquid phase life data from the autoclaves has been encouraging. CO-rich gas was run in the autoclave in excess of 1500 hours, and balanced gas in excess of 1000 hours. Activities were

stable with a very slow rate of decline. The slow liquid phase deactivation is less than expected for gas phase synthesis with dilute balanced gas.

In the Lab PDU effort, modifications to the Fairfield unit to allow either ebullated or slurry operation were completed. A second ebullated test was performed with an improved catalyst and with CO-rich gas. Catalyst activity maintenance over the 500 hour run was excellent but there is still an apparent catalyst attrition problem. Another catalyst candidate is presently under test. The first Lab PDU slurry run was recently accomplished with an 18 wt% loading. The early interpretation of the data suggests PDU performance similar to the autoclave (i.e., good mass transfer, mixing). With balanced gas at 1000 psig, 250°C, very high methanol specific productivities were achieved, up to  $55 \text{ g mole/hr-kg} = 1.8 \text{ kg/hr-kg}$  at a space velocity of 22,000 1/kg-hr. Gas holdup at superficial gas velocities up to 0.23 ft/sec. appears to be lower than expected from cold flow tests.

In the fundamental modeling effort, the slurry reactor has been described as a back flow cell model. Intrinsic rate constants were back calculated from autoclave data. Mass transfer is calculated from the  $k_L a$  correlation of Akita and Yoshida, using the gas holdup determined in cold flow tests. When applied to the recent Lab PDU slurry run, the model provides a remarkably good fit of the balanced gas data at both 1000 psig and 500 psig.

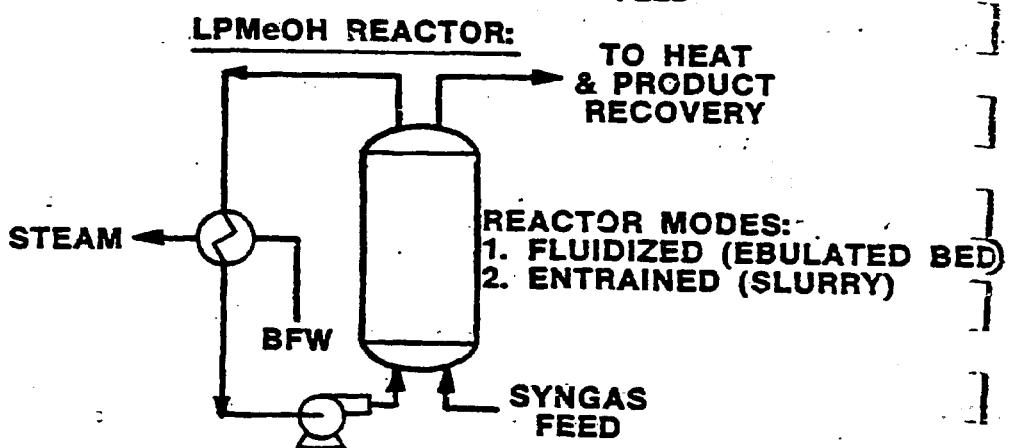
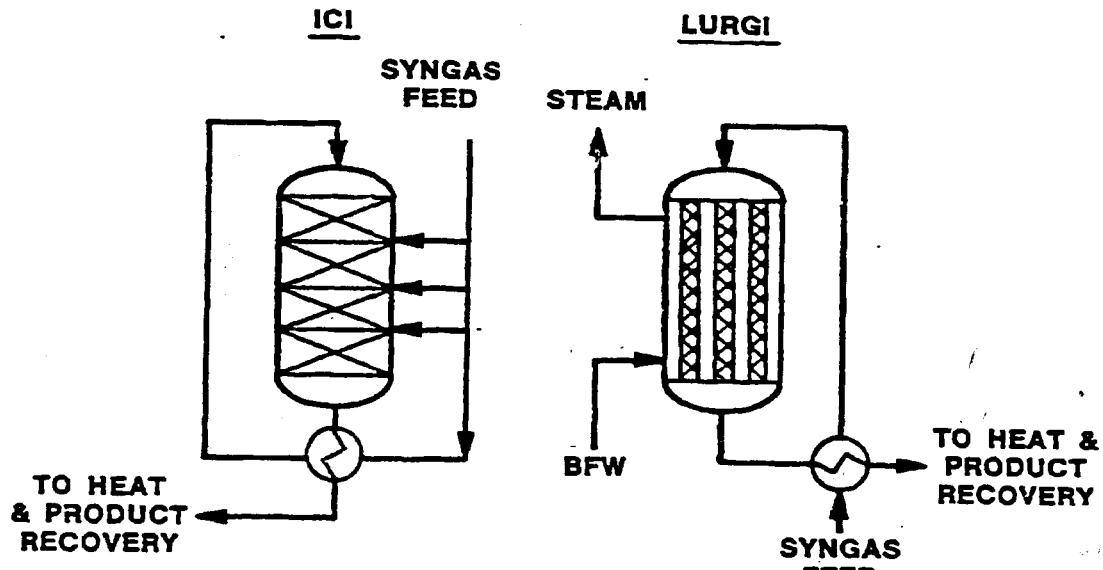
The LaPorte PDU engineering is now complete and intense activity is underway at the site. Mechanical work is 95% complete and instrument/electrical work is 60% along. Integrated shakedown will begin in November. The ebullated operating campaign will start in December, with the slurry mode to follow in the spring of 1984.

**LIQUID PHASE METHANOL  
UPDATE**

- INTRODUCTION
- TECHNICAL CHALLENGES
  - BENCH SCALE - RESULTS AND INTERPRETATION
  - LAB PDU - RESULTS AND INTERPRETATION
  - FUNDAMENTAL MODELING PROGRESS
- SUMMARY AND PERSPECTIVE
- LAPORTE PDU ENGINEERING STATUS

## **REACTOR SCHEMATICS**

### **CONVENTIONAL MeOH REACTORS:**



## LPMeOH TECHNICAL MOTIVATION

LIQUID, AGITATED MEDIUM

1. LARGE HEAT SINK TO CONTROL REACTION TEMPERATURE
2. POTENTIAL FOR BROADER CATALYST COMPOSITIONS  
(SLURRY)



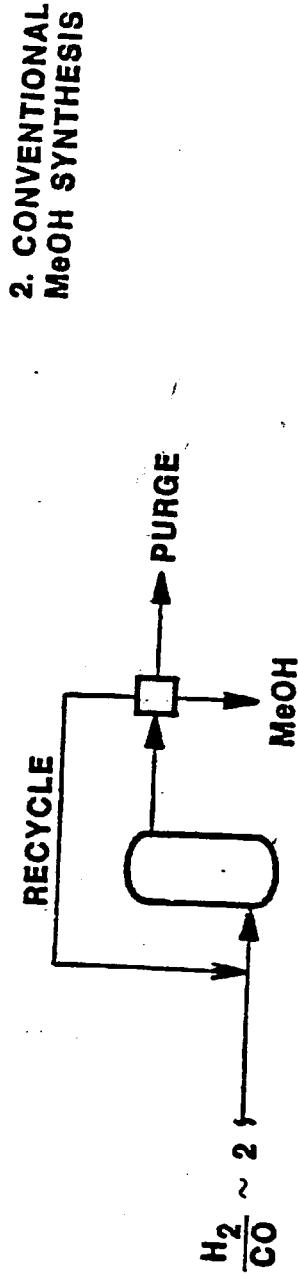
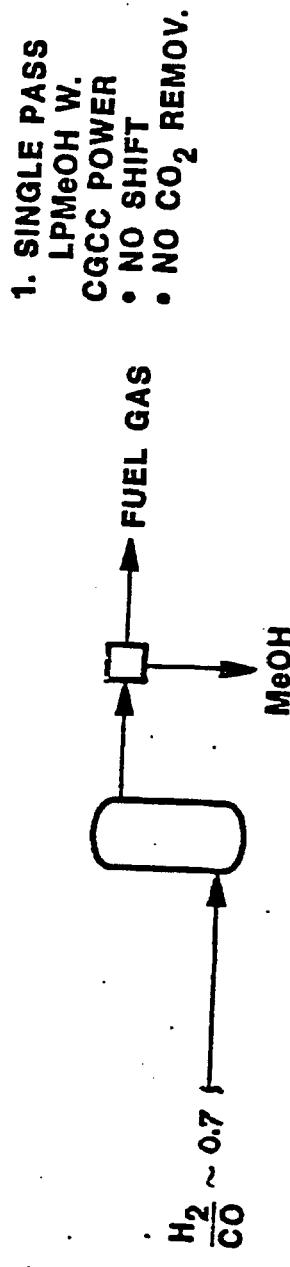
- LOW TEMPERATURE RISE
- HIGH CONVERSION PER PASS
- LOW RECYCLE RATE
- POTENTIAL TO ACCEPT LOW H<sub>2</sub>/CO FEED

## APPLICATIONS

1. METHANOL + FUEL GAS (COAL GASIF, COMBINED CYCLE)
2. ALL-METHANOL

## APPLICATIONS

1. METHANOL + FUEL GAS (COAL GAS/F. COMBINED CYCLE)
2. ALL-METHANOL



# LPMeOH PROJECT DESCRIPTION

## OBJECTIVE

- DEMONSTRATE TECHNICAL FEASIBILITY OF LPM<sub>e</sub>OH PROCESS AT PDU SCALE

~ 1 MM SCFD  
~ 5 TPD MeOH

## STRATEGY

- UTILIZE EXISTING CHICAGO LPM PDU (LIQUID-FLUIDIZED)
- RELOCATE, REFURBISH, INSTALL AT LAPORTE SYNGAS FACILITY
- OPERATE LIQUID-FLUIDIZED (EBULLATED BED) MODE
- CONCURRENT WITH ABOVE, DEVELOP NEW CATALYSTS/ DATA BASE FOR LIQUID-ENTRAINED (SLURRY) MODE
- MODIFY PDU FOR LIQUID-ENTRAINED OPERATION
- OPERATE LIQUID-ENTRAINED (SLURRY) MODE

# **LPMEOH PROJECT DESCRIPTION (CONT.)**

## **DURATION**

- 42 MONTHS (1 OCTOBER 1981 - 1 APRIL 1985)

## **PARTICIPANTS**

- DOE
- AIR PRODUCTS
- CHEM SYSTEMS
- EPRI
- FLUOR

## LAPORTE LPMeOH PDU PRINCIPAL FEED GAS COMPOSITIONS

(TEXACO GASIFIER)

METHANOL + FUEL GAS COPRODUCTS  
UNBALANCED TYPE  
REACTOR FEED

H <sub>2</sub>	34.8 MOLE %
CO	51.2
CO <sub>2</sub>	13.1
CH <sub>4</sub> , C <sub>2</sub> H <sub>6</sub>	0.1
N <sub>2</sub> , Ar, INERTS	<u>0.8</u>
TOTAL	100.0
H <sub>2</sub> /CO	0.68
<u>H<sub>2</sub></u>	0.49
<u>(CO+1.5CO<sub>2</sub>)</u>	
<u>(H<sub>2</sub>-CO<sub>2</sub>)</u>	0.34
<u>(CO+CO<sub>2</sub>)</u>	

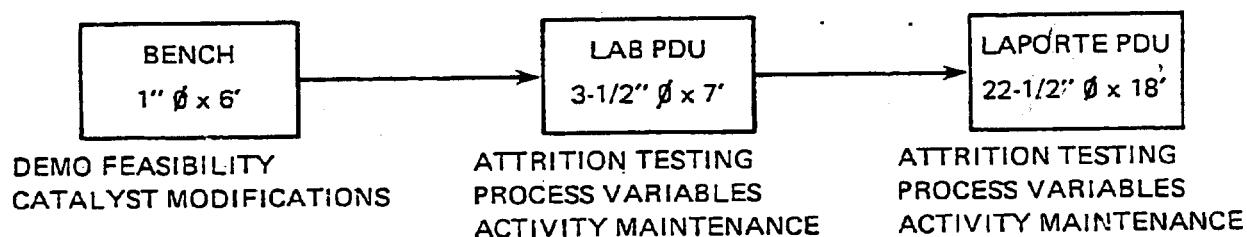
## LAPORTE LPMeOH PDU PRINCIPAL FEED GAS COMPOSITIONS (CONT.)

### ALL-METHANOL PRODUCTS

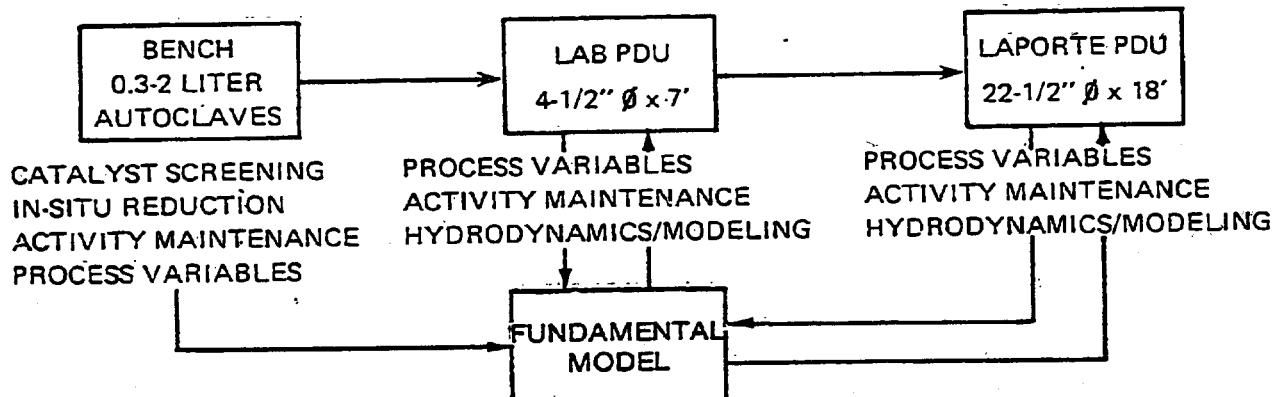
	SHIFTED FRESH FEED	BALANCED TYPE REACTOR FEED
H <sub>2</sub>	66.4 MOLE %	54.9 MOLE %
CO	30.7	18.8
CO <sub>2</sub>	2.0	4.9
CH <sub>4</sub> ,C <sub>2</sub> H <sub>6</sub>	0.1	2.1
N <sub>2</sub> , Ar, INERTS	<u>0.8</u>	<u>19.3</u>
TOTAL	100.0	100.0
H <sub>2</sub> /CO	2.16	2.92
<u>H<sub>2</sub></u> <u>(CO+1.5CO<sub>2</sub>)</u>	1.97	2.10
<u>(H<sub>2</sub>-CO<sub>2</sub>)</u> <u>(CO+ CO<sub>2</sub>)</u>	1.97	2.11

# LPMeCH TECHNOLOGY DEVELOPMENT

## LIQUID-FLUIDIZED (EBULLATED BED) REACTOR 1975-1983



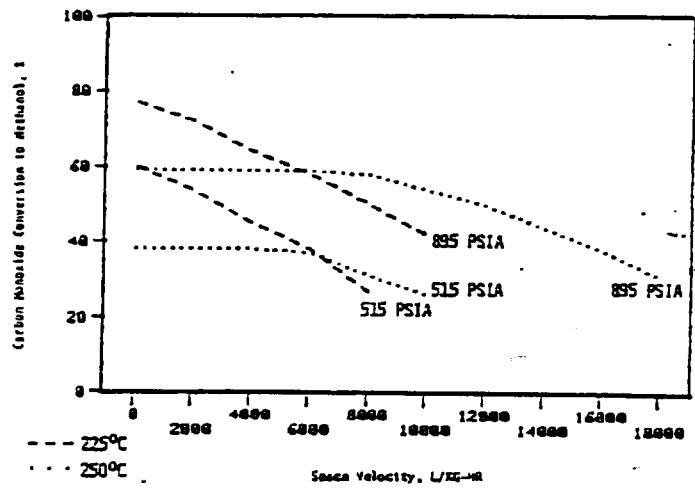
## LIQUID-ENTRAINED (SLURRY) REACTOR 1979-1983



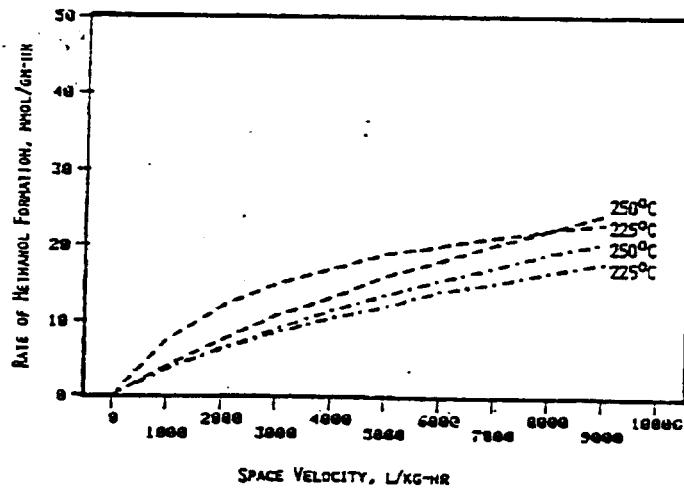
## **LABORATORY AUTOCLAVE PROGRAM**

- OBJECTIVE:** EXPLOIT SLURRY SYSTEM ON CATALYST ACTIVITY,  
SELECTIVITY, LIFE ISSUES
- COMPARE GAS PHASE TO LIQUID PHASE  
OPERATION
- SCREEN COMMERCIAL CATALYSTS
- SCREEN NEW EXPLORATORY CATALYSTS
- OPTIMIZE REDUCTION TECHNIQUES
- CONDUCT PROCESS VARIABLE SCANS AND LIFE  
TESTS
- USE DATA TO MODEL PROCESS

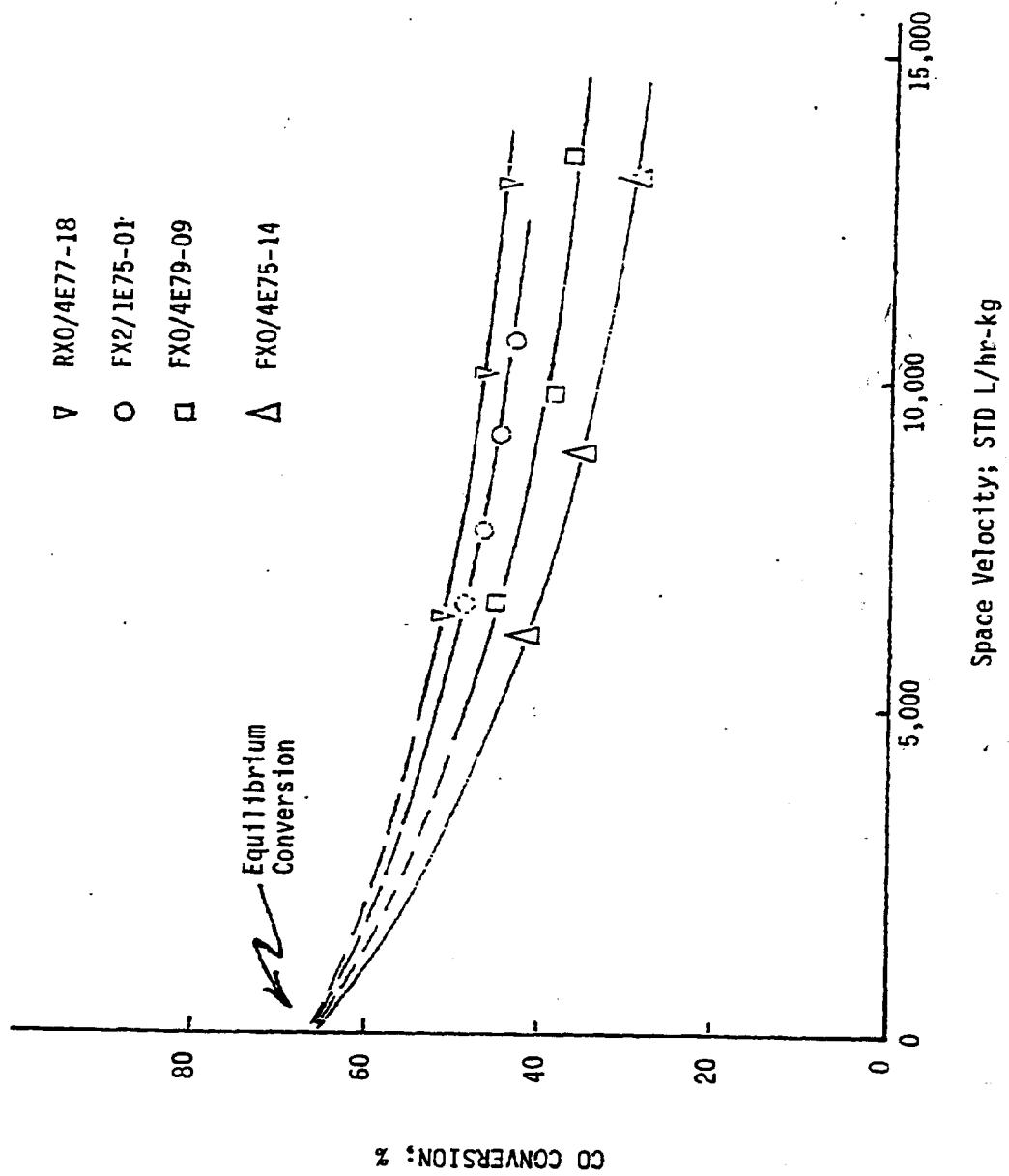
**CATALYST ACTIVITY IN  
APCI GAS PHASE SCREENING REACTOR**  
 FEED GAS = 55% H<sub>2</sub>/19% CO/5% CO<sub>2</sub>/N<sub>2</sub>  
 CATALYST = F50/4E75-01  
 REACTOR LOAD = 3 gm (AS OXIDE)



**EFFECT OF GAS COMPOSITION**  
**APCI GAS PHASE SCREENING REACTOR**  
 FEED GAS = 55% H<sub>2</sub>/19% CO/5% CO<sub>2</sub>/N<sub>2</sub>  
 FEED GAS = 35% H<sub>2</sub>/50% CO/13% CO<sub>2</sub>/N<sub>2</sub>  
 CATALYST = F50/4E75-01  
 PRESSURE = 515 PSIA (3500 KPA)



**CO CONVERSION VS. S. ACE VELOCITY -  
BALANCED GAS AT 250°C;  
7000 kPa IN FREEZENE-100**



## SOME NEW APCI LPMeOH CATALYST POWDERS

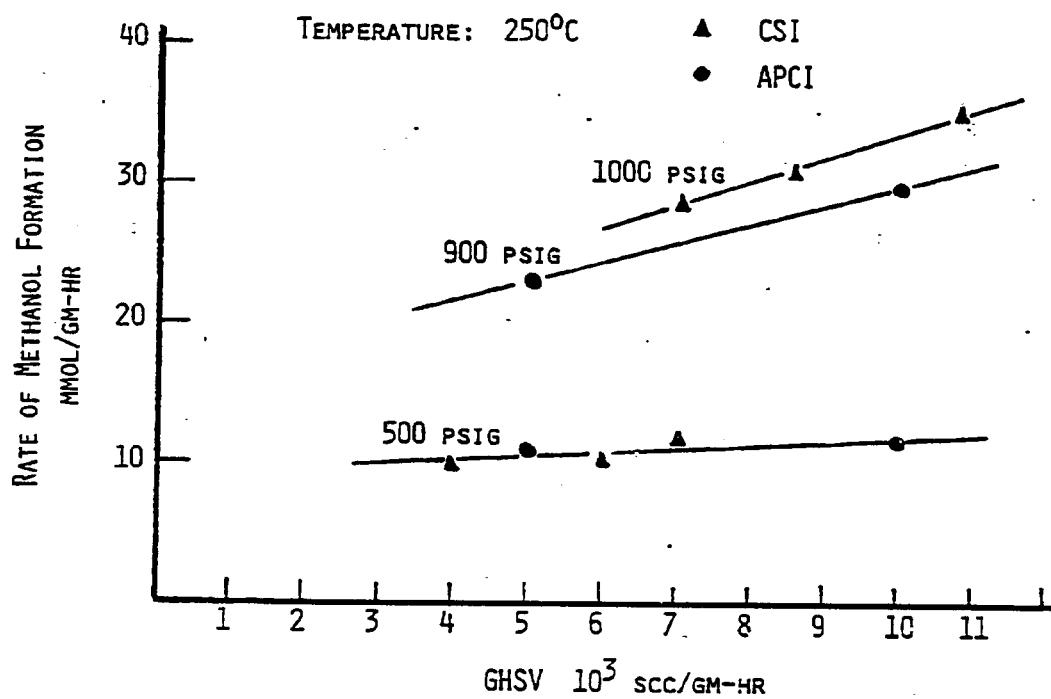
CATALYST PREP NO.	COMPOSITION	MeOH YIELD*, m.mol/gm-hr
ICI Commercial	Cu/Zn/Al <sub>2</sub> O <sub>3</sub>	18.6
66	Cu/Zn/Al <sub>2</sub> O <sub>3</sub> (Stilles)	14.6
56	Cu/Zn/Cr <sub>2</sub> O <sub>3</sub>	NII
50	Cu/Zn (Binary)	0.7
60	Cu/Zn/Boron	7.8
63	Cu-Zn-Al Raney (stored)	1.0
75	Cu-Zn/Al Raney (fresh)	6.0
70	Ni-Boride	0.05 (25.8 CH <sub>4</sub> )

\*Gas-Phase Screening @ 500 psig, 225°C, 5000 l/hr-kg.  
Balanced Gas: 55% H<sub>2</sub>, 19% CO, 5% CO<sub>2</sub>, 21% N<sub>2</sub>

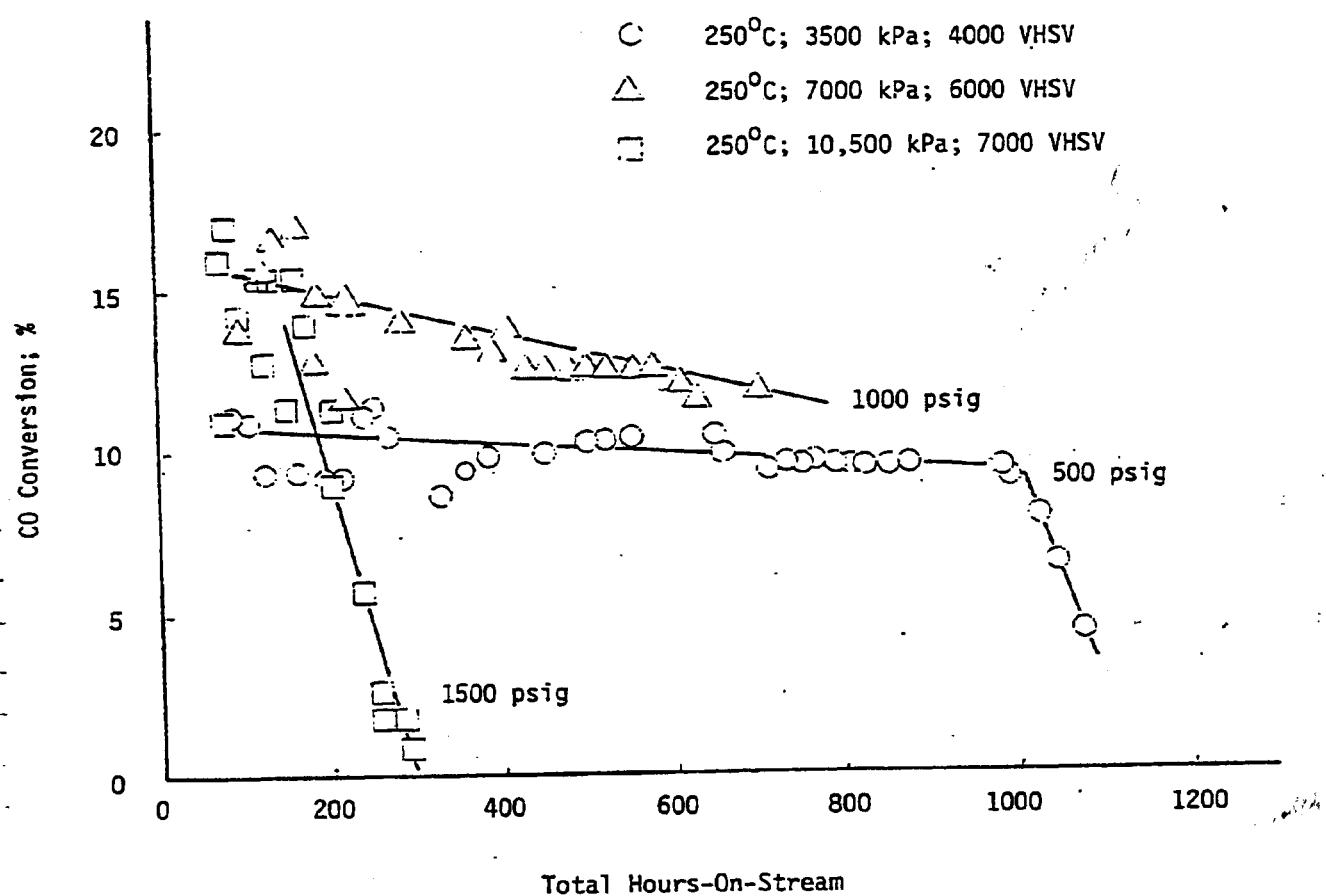
## APCI STIRRED AUTOCLAVE

### CROSS-CHECK WITH CSI AUTOCLAVE:

- SAME BASELINE COMMERCIAL CATALYST (F50/4E75)
- GAS PHASE REDUCTION
- BALANCED GAS, 55% H<sub>2</sub>/ 19% CO/5% CO<sub>2</sub>/21% N<sub>2</sub>
- 15 WT% SOLIDS
- 30 DAYS OPERATION



**CO Conversion Vs. Total Hours  
on Stream for Catalyst  
F50/4E75-01; In-Situ Reduction  
in Freezene-100**



## CARBONYLS AT LAPORTE

- CONCERN WITH CO RICH GAS (UNBALANCED CASE)



- POTENTIAL IMPACT
  - LINE CORROSION
  - CATALYST POISONING

### INDUSTRIAL EXPERIENCE

#### LIMITED TO GAS PHASE SYNTHESIS

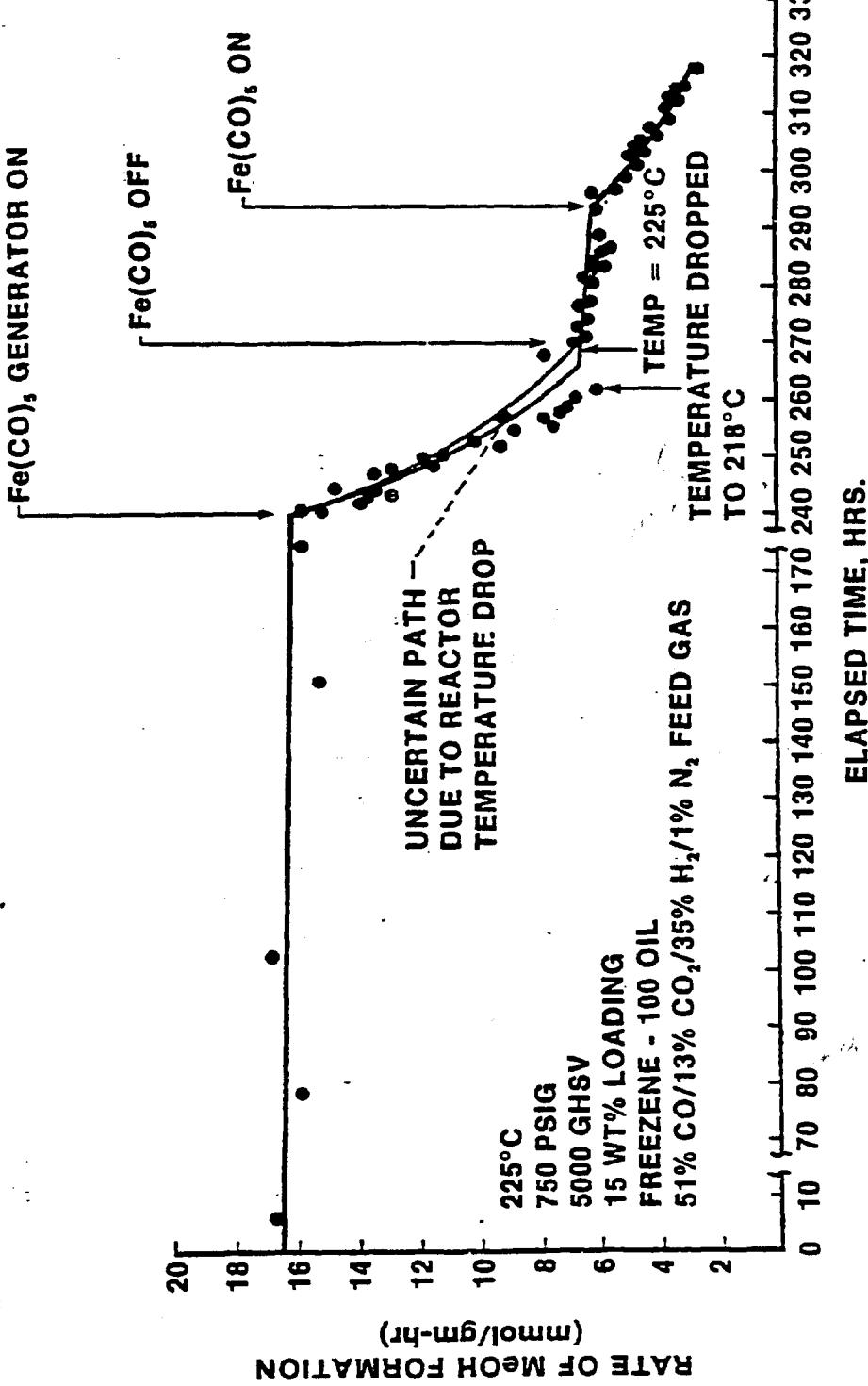
#### SIGNIFICANT IN HIGH PRESSURE MeOH

- CATALYST DEACTIVATION
- METHANATION

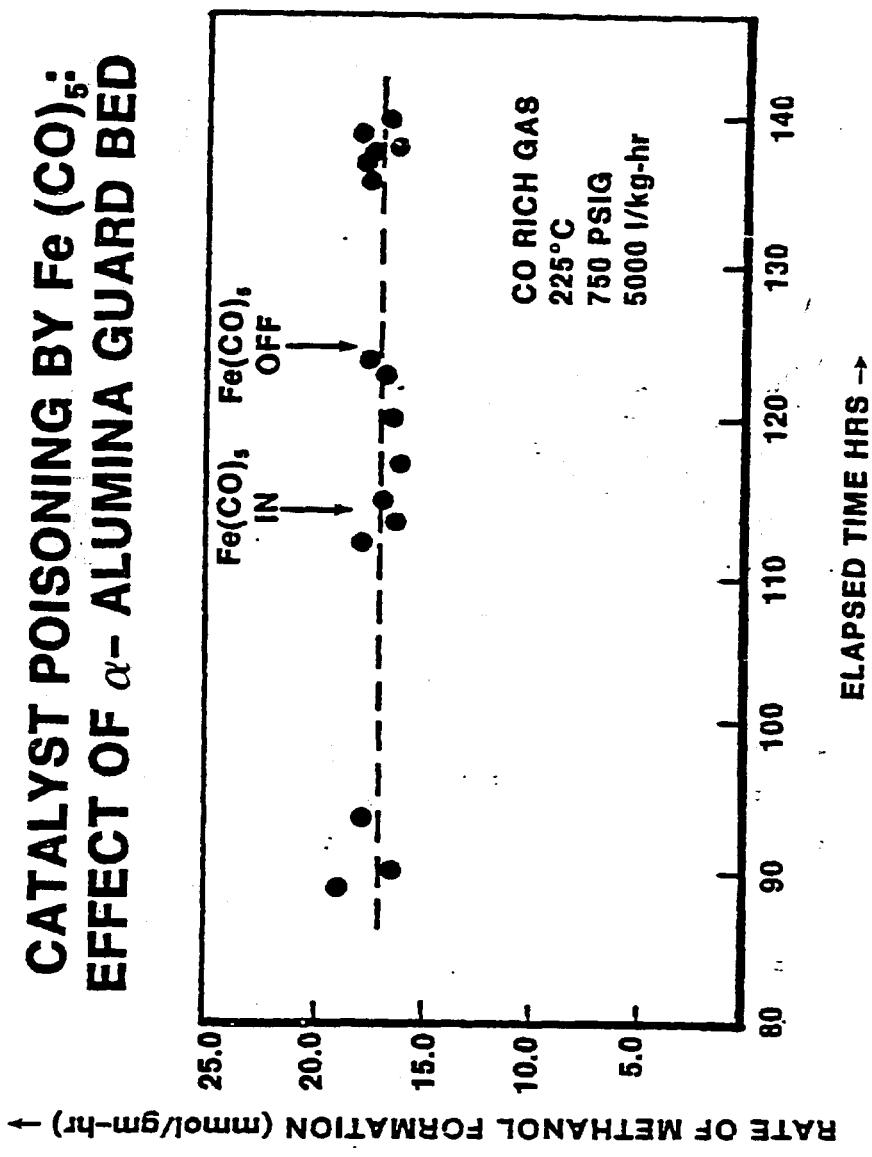
#### CORRECTIVE ACTION

- IMPROVED STEELS/COPPER LINING
- ACTIVATED CARBON
- MAINTAIN HIGH CO STREAMS AT T°F  
WHERE  $200^\circ > T > 600^\circ$

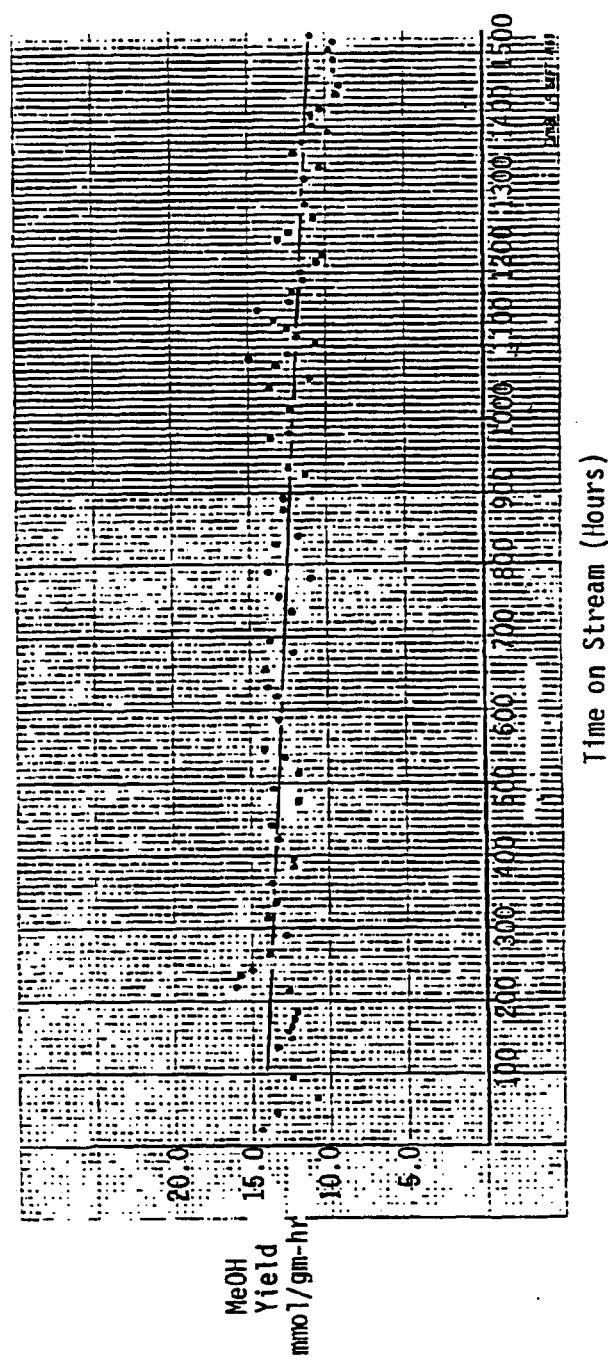
CATALYST POISONING STUDIES:  
 EFFECT OF  $\text{Fe}(\text{CO})_5$   
 COMMERCIAL CATALYST



CATALYST POISONING BY  $\text{Fe}(\text{CO})_5$ :  
EFFECT OF  $\alpha$ -ALUMINA GUARD BED



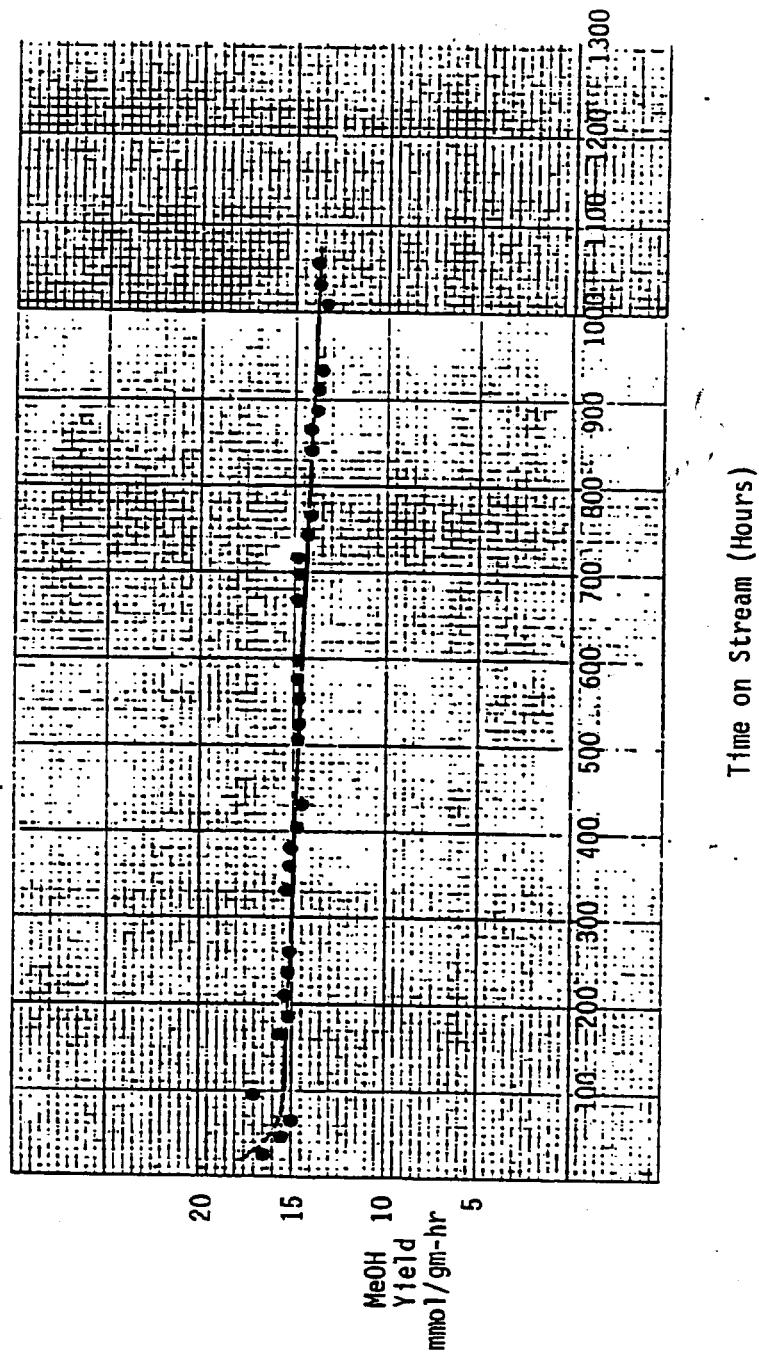
**MeOH PRODUCTIVITY WITH TIME**  
**CSI AUTOCLAVE: CO-RICH, 250°C, 750 PSIG,**  
**15 WT% CATALYST**

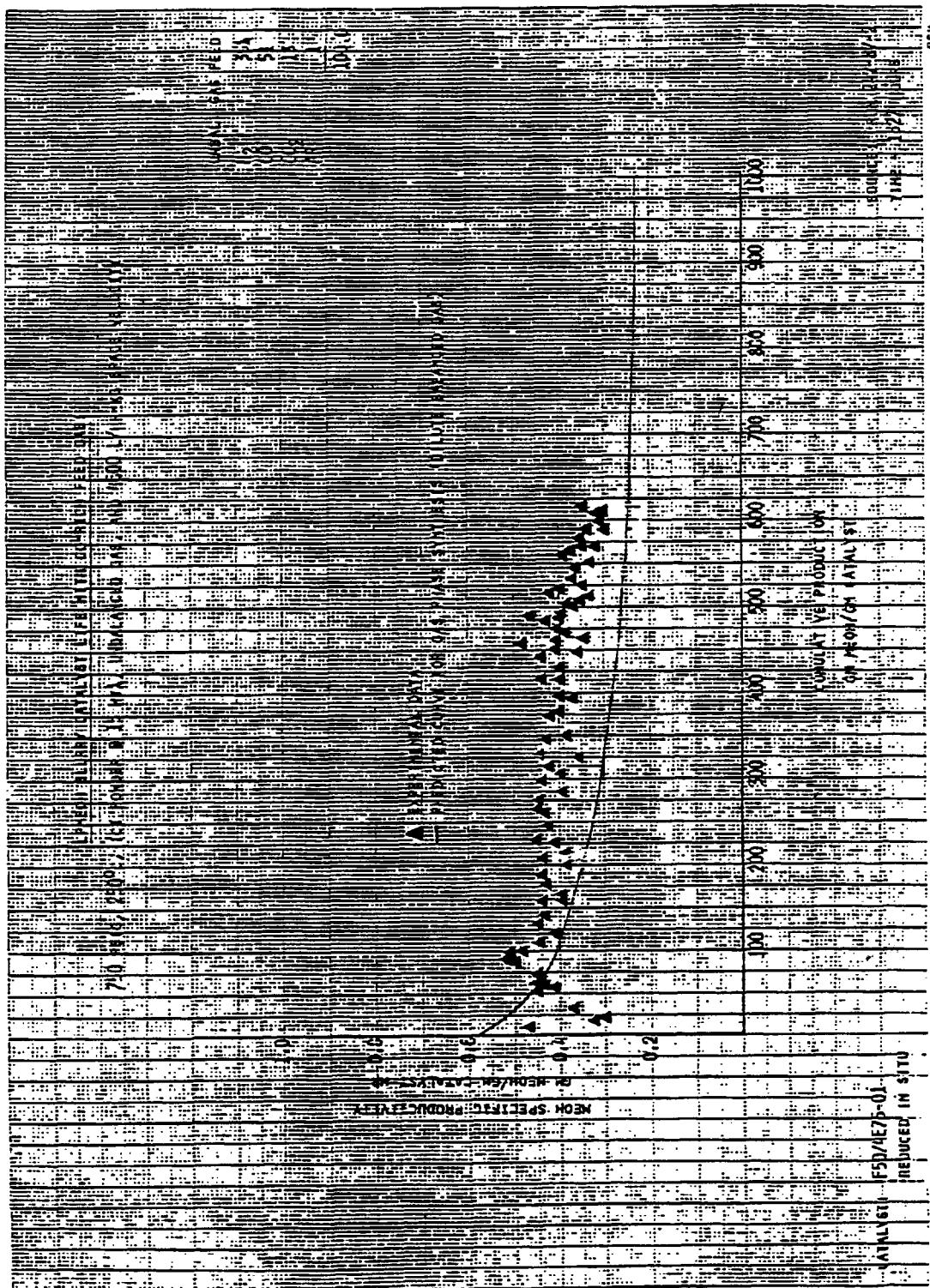


LIQUID PHASE

Balanced Gas: 15 wt% F50/4E75-01  
250°C, 750 psig, 5000 GHSV

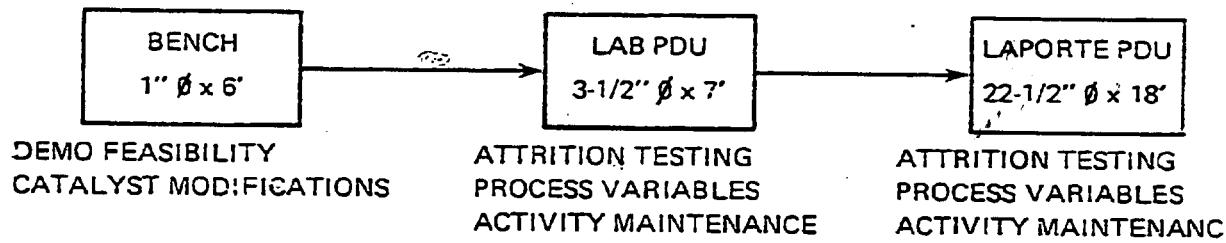
Methanol Productivity with Time



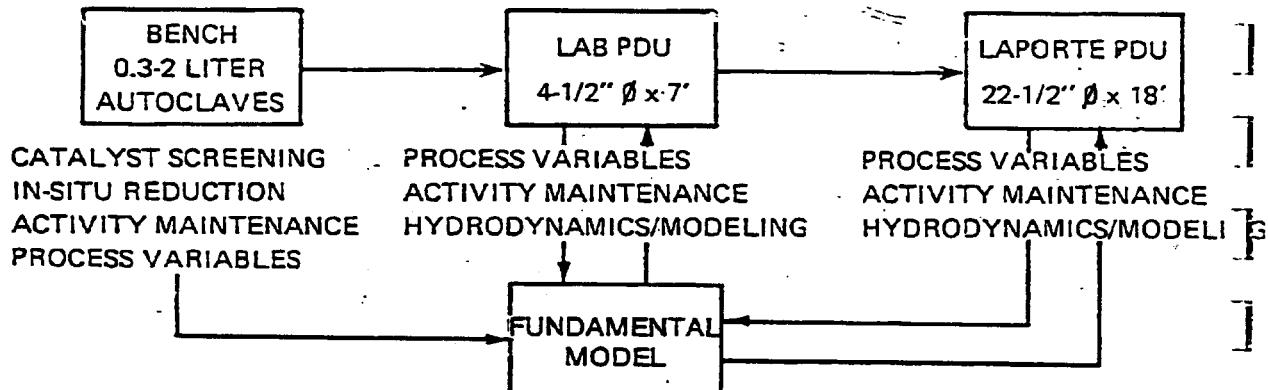


# LPMeOH TECHNOLOGY DEVELOPMENT

## LIQUID-FLUIDIZED (EBULLATED BED) REACTOR 1975-1983



## LIQUID-ENTRAINED (SLURRY) REACTOR 1979-1983

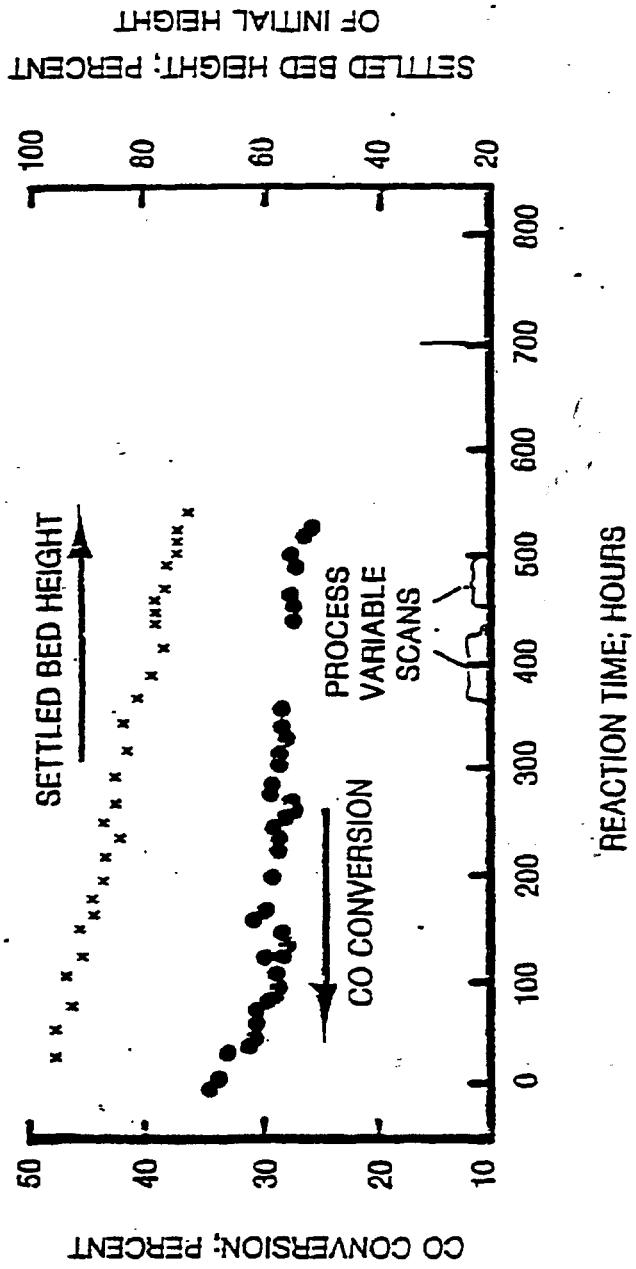


## **OBJECTIVES OF LAB PDU PROGRAM**

- DEMONSTRATE EBULLATED AND SLURRY MODES  
AT A SCALE INTERMEDIATE TO LAPORTE
  - EBULLATED FACES CATALYST ATTRITION AND  
ACTIVITY ISSUES
  - SLURRY FACES CATALYST LOADING AND  
ACTIVITY ISSUES

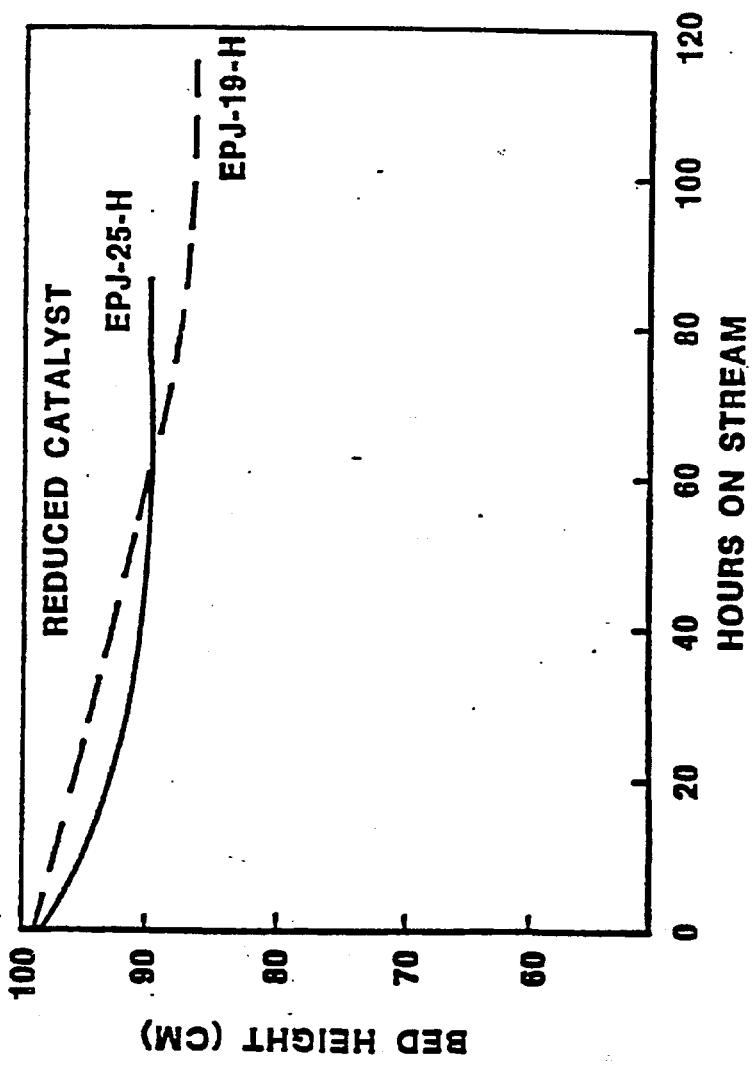
## CATALYST ACTIVITY AND ATTRITION IN CSI LIQUID-FLUIDIZED LAB PDU

2/1 H<sub>2</sub>/CO FEED GAS WITH 10% CO<sub>2</sub>  
250°C, 7000 kPa  
3000 L/Hr. - Kg CAT APPARENT SPACE VELOCITY  
3.8 CM/SEC. SUPERFICIAL LIQUID VELOCITY



HRI COLD-FLOW TESTS

WET SETTLED BED HEIGHT COMPARISON

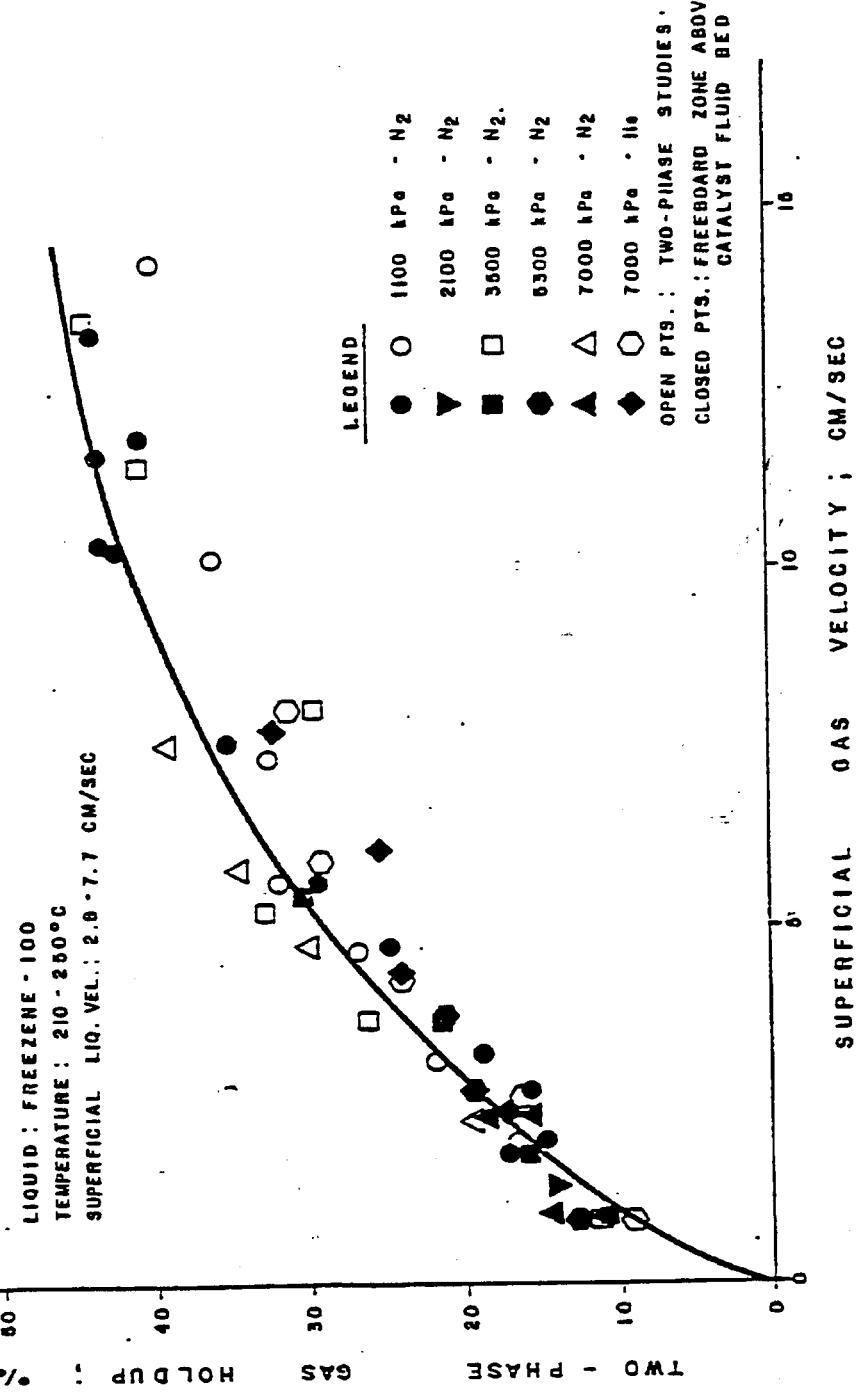


# HRI COLD-FLOW TESTS ON EBULLATED CATALYST CANDIDATES

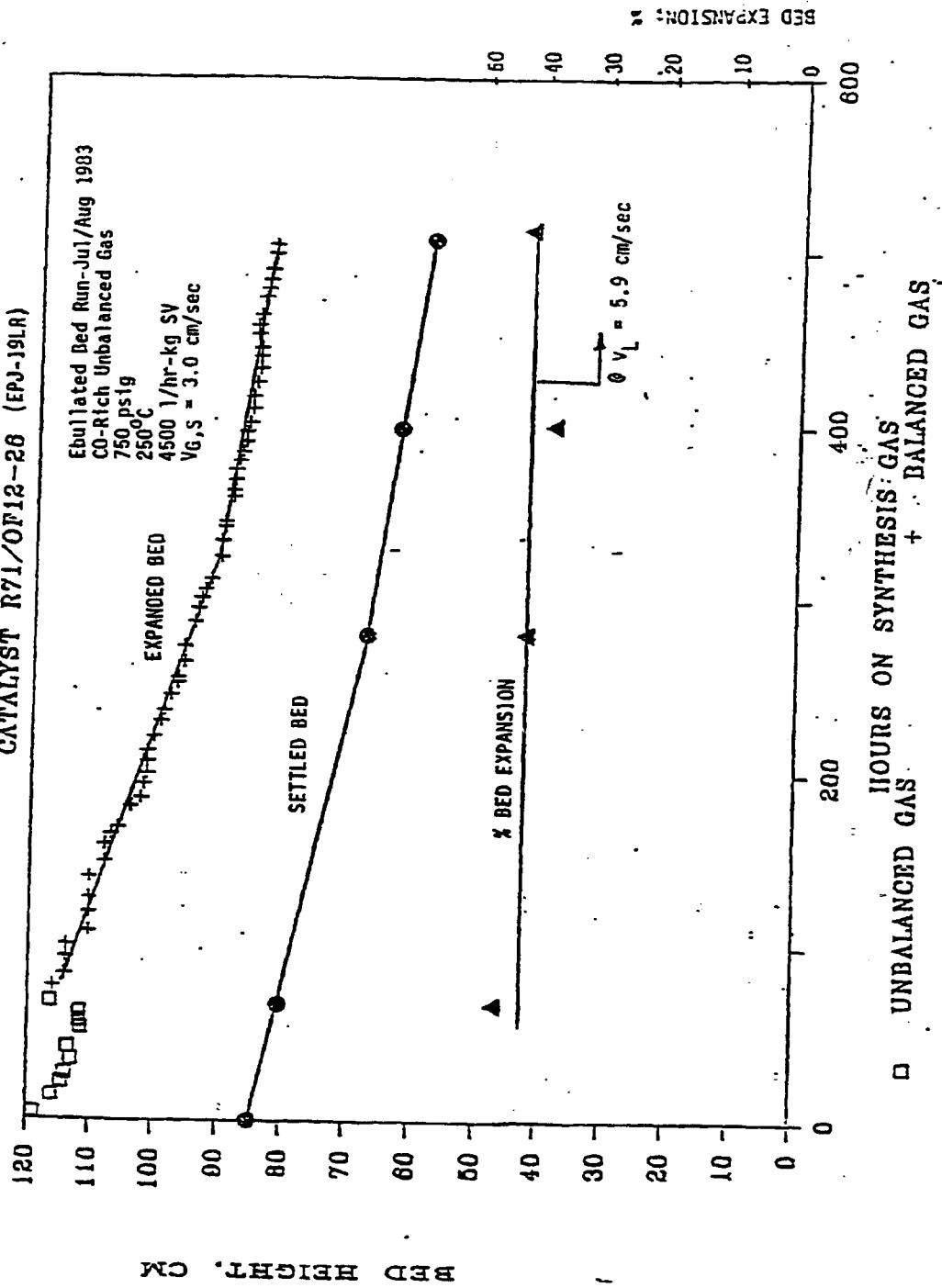
## CONCLUSIONS: EPJ 19-H AND EPJ 25-H

- COMPARABLE AND ACCEPTABLE ATTRITION RATES  
0.5 WT% PER DAY
- FINES GENERATION PREDOMINATE  
LOW STEADY STATE CARRYOVER RATE
- UNIFORM BED DENSITY FOR 19-H  
SLIGHT DENSITY GRADIENT FOR 25-H
- DECREASE IN CATALYST LENGTH SEEN  
NO CHANGE IN DIAMETER

GAS HOLDUP DATA DURING EXPANSION  
STUDIES IN THE CSI PDU

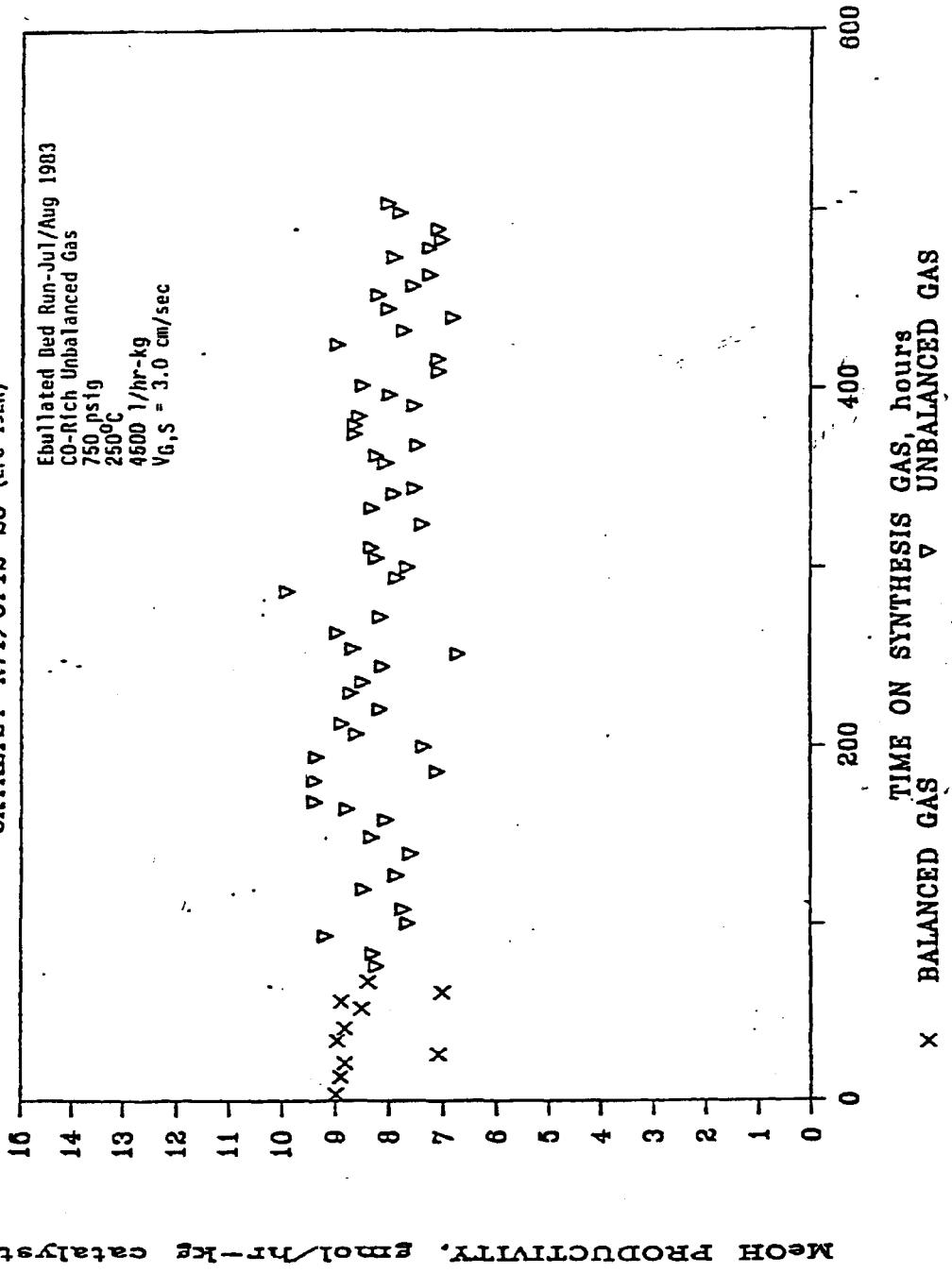


*CSI LAB PDU RUN*  
*CATALYST R71/OP12-26 (EPJ-19LR)*

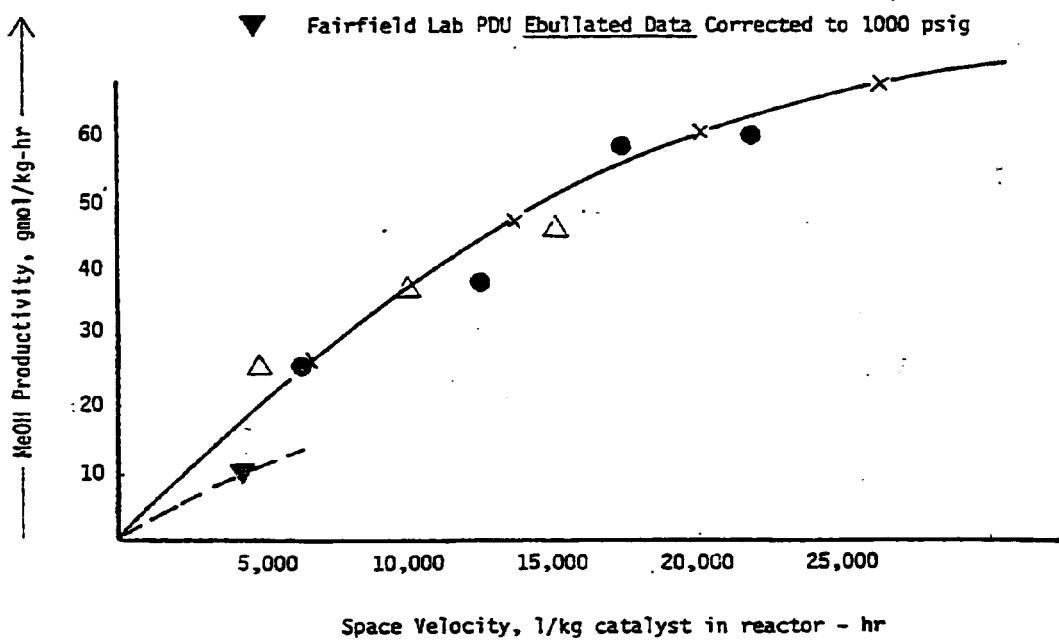
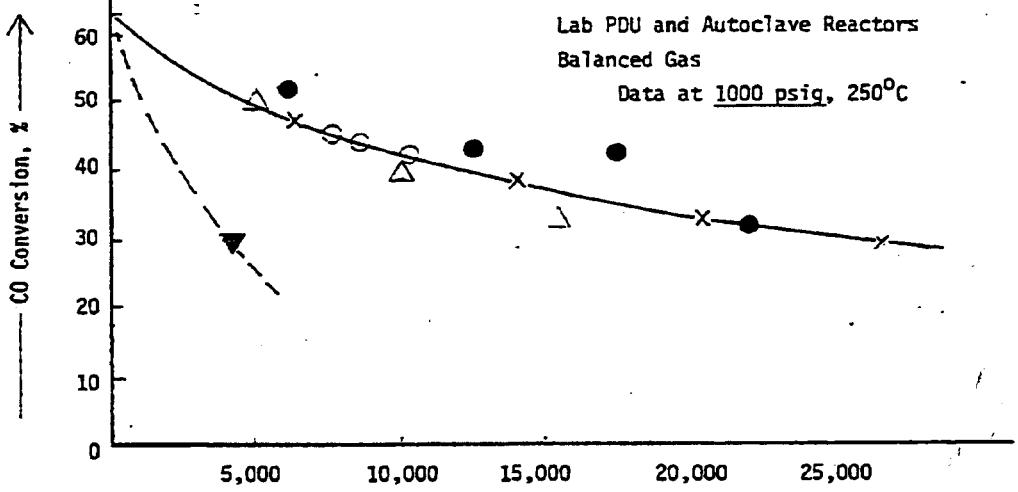


# LA PORTE BATCH CHECK IN CSI LAB PDU

CATALYST R71/0F12-26 (EPJ-19LR)



## LPMeOH SLURRY PERFORMANCE



## **LAB PDU - PRELIMINARY CONCLUSIONS**

- EBULLATED STILL HAS CATALYST ATTRITION QUESTION
- ACTIVITY MAINTENANCE IN EBULLATED LOOKS GOOD
- NO EVIDENCE OF MASS TRANSFER EFFECTS IN PDU

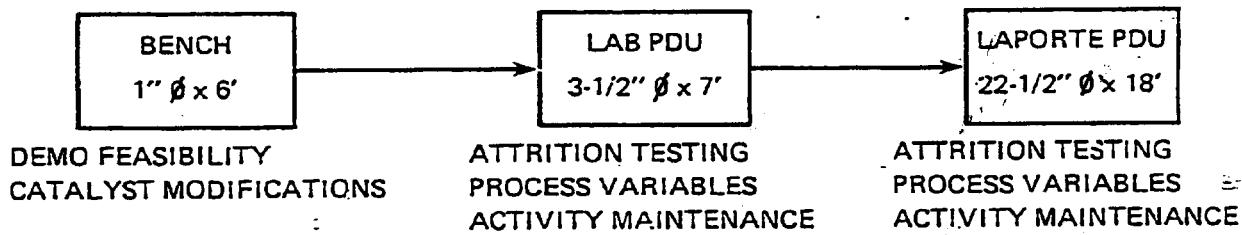
## **FURTHER LAB PDU**

**TEST OTHER EBULLATED CATALYST CANDIDATE  
COMPLETE FURTHER SLURRY WORK**

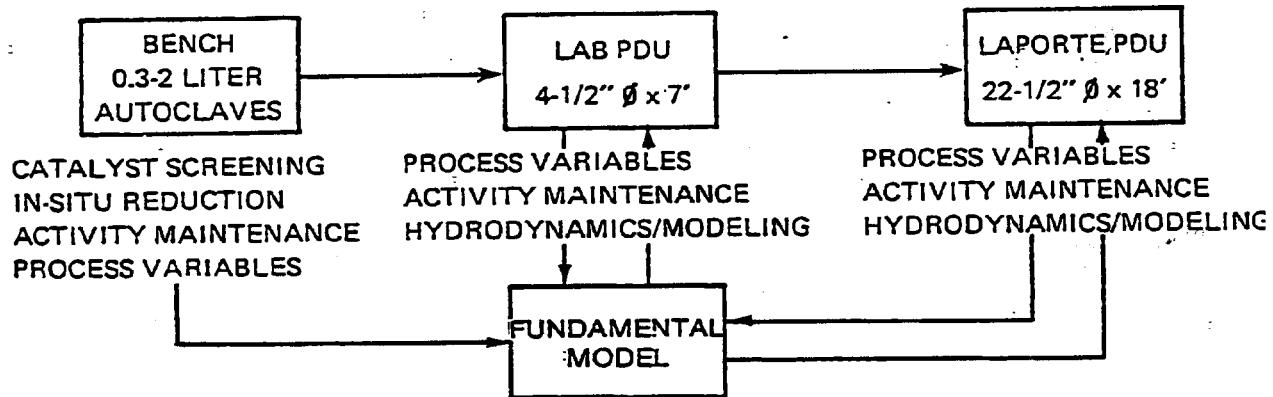
- > MAXIMIZE THROUGHPUT**
- > MAXIMIZE SLURRY LOADING**

# LPMeOH TECHNOLOGY DEVELOPMENT

## LIQUID-FLUIDIZED (EBULLATED BED) REACTOR 1975-1983



## LIQUID-ENTRAINED (SLURRY) REACTOR 1979-1983



## **MODELING PROGRAM**

### **GOAL:**

**CONSTRUCT A REALISTIC MATHEMATICAL MODEL WHICH  
ALLOWS PREDICTION OF REACTANT CONVERSION AND  
PRODUCT YIELD FOR A RANGE OF OPERATING CONDITIONS**

### **APPLICATIONS:**

- RECALCULATION OF PROCESS HEAT AND MATERIAL BALANCES**
- ASSIST IN DEFINING OPERATING CONDITIONS FOR LAPORTE**
- USE IN COMMERCIAL LP MeOH DESIGN STUDIES AND ONGOING ECONOMIC EVALUATIONS**

## **SLURRY REACTOR MODEL**

**THREE PHASES, MIXING BETWEEN PLUG FLOW AND PERFECTLY  
MIXED**

- **DISPERSION MODEL - CLOSER TO PLUG FLOW**
- **TANKS IN SERIES MODEL - CLOSER TO PERFECT MIXING**

**LAPORTE REACTOR PROBABLY CLOSER TO PERFECT MIXING**

**REACTOR VOLUME V DIVIDED INTO N EQUAL TANKS WHERE**

$$\frac{1}{N} = 2\left(\frac{D}{VL}\right) - 2\left(\frac{D}{VL}\right)^2(1 - e^{-VL/D})$$

**WHICH FOR LARGE DISPERSIONS       $N \rightarrow VL/2D$**

**V = SUPERFICIAL VELOCITY**

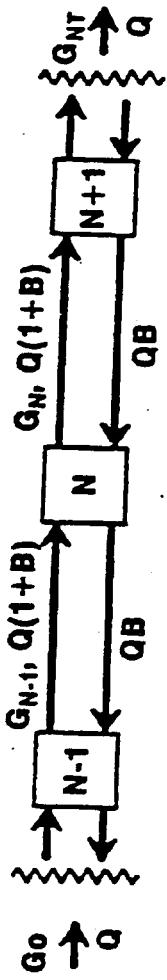
**L = REACTOR LENGTH**

**D = DISPERSION COEFFICIENT**

**FOR MULTI-PHASE, NEED N FOR EACH PHASE**

## INTRODUCE BACKFLOW MODIFICATION

- USES A SINGLE NUMBER OF TANKS IN SERIES WITH N BASED ON THE LEAST DISPERSED PHASE
- MIXING OF OTHER PHASES DESCRIBED BY DIFFERENT DEGREES OF BACKFLOW, B

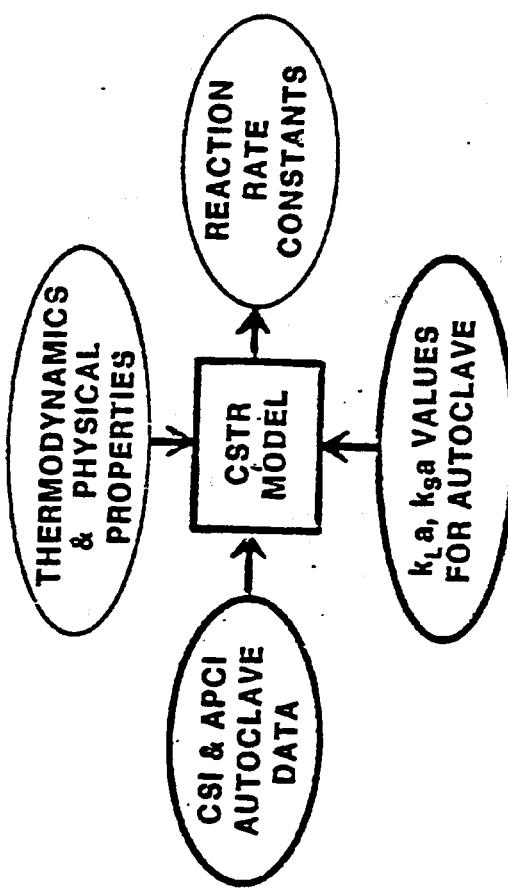


- CAN HAVE B FOR MORE THAN ONE PHASE, HENCE DESCRIBING DEGREES OF MIXING FOR ALL PHASES
- CURRENT MODEL DESCRIBES 7 TANKS IN SERIES WITH BACKFLOW RATIOS UP TO 10 AVAILABLE

## REACTION RATE DATA

CSI & APCI AUTOCLAVES: ADVANCE FUNDED WORK, CURRENT  
DOE CONTRACT

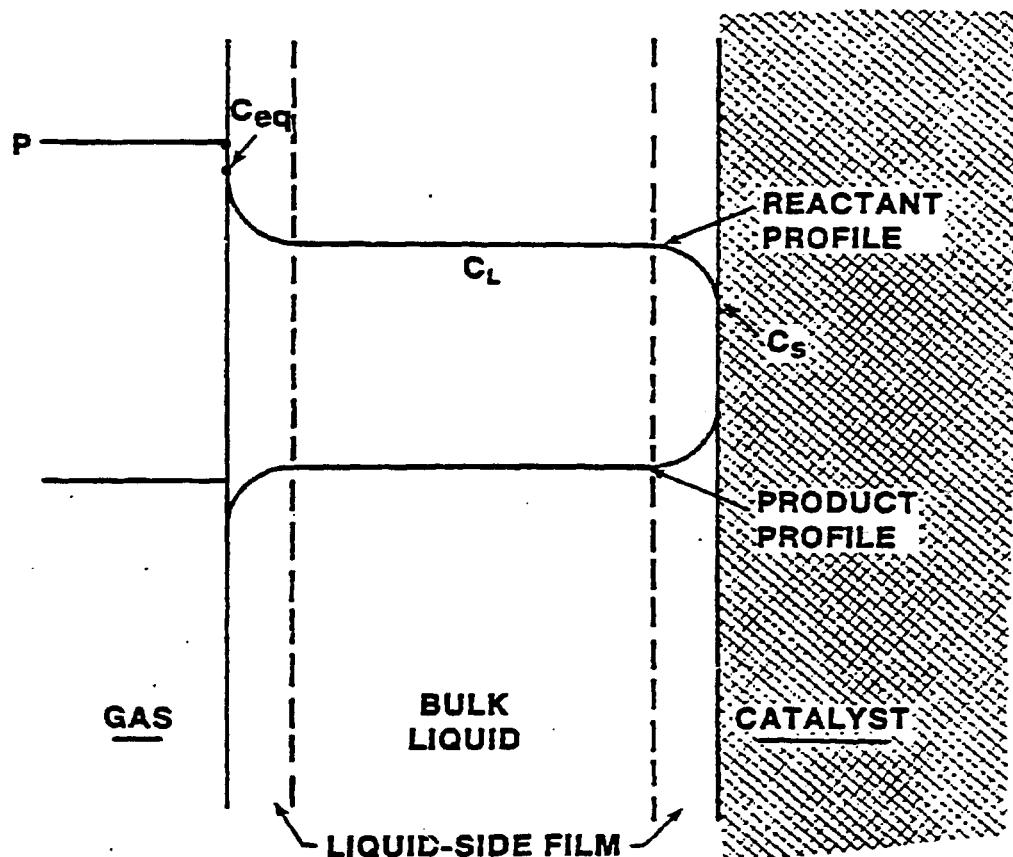
USE A CSTR MODEL TO DESCRIBE AUTOCLAVE OPERATION



### ASSUMPTIONS

- WELL MIXED REACTOR
- NO INTRA PARTICLE DIFFUSION

## CONCENTRATION PROFILES IN STIRRED AUTOCLAVE REACTOR



$C_{eq}$  = LIQUID CONCENTRATION IN EQUILIBRIUM  
WITH GAS

$C_L$  = BULK LIQUID CONCENTRATION

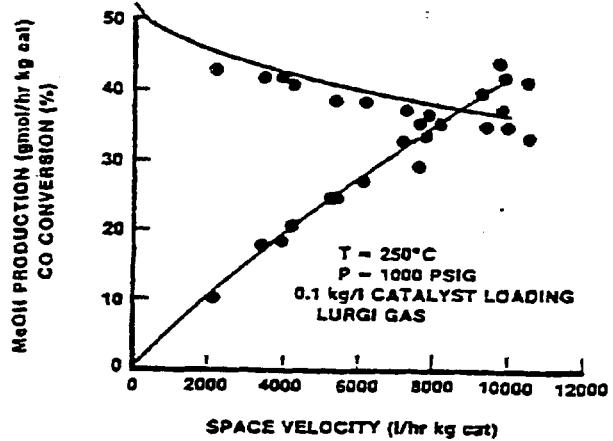
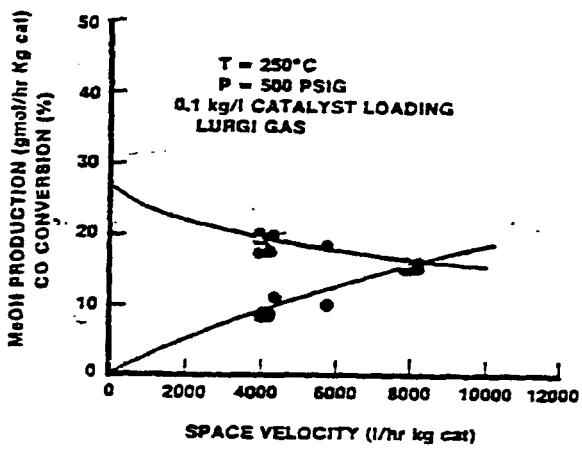
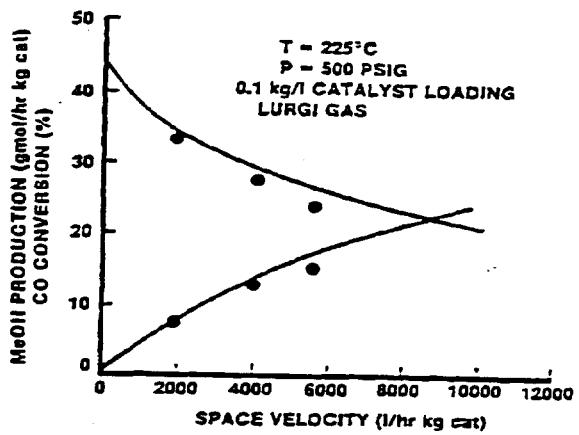
$C_s$  = LIQUID CONCENTRATION AT CATALYST  
SURFACE

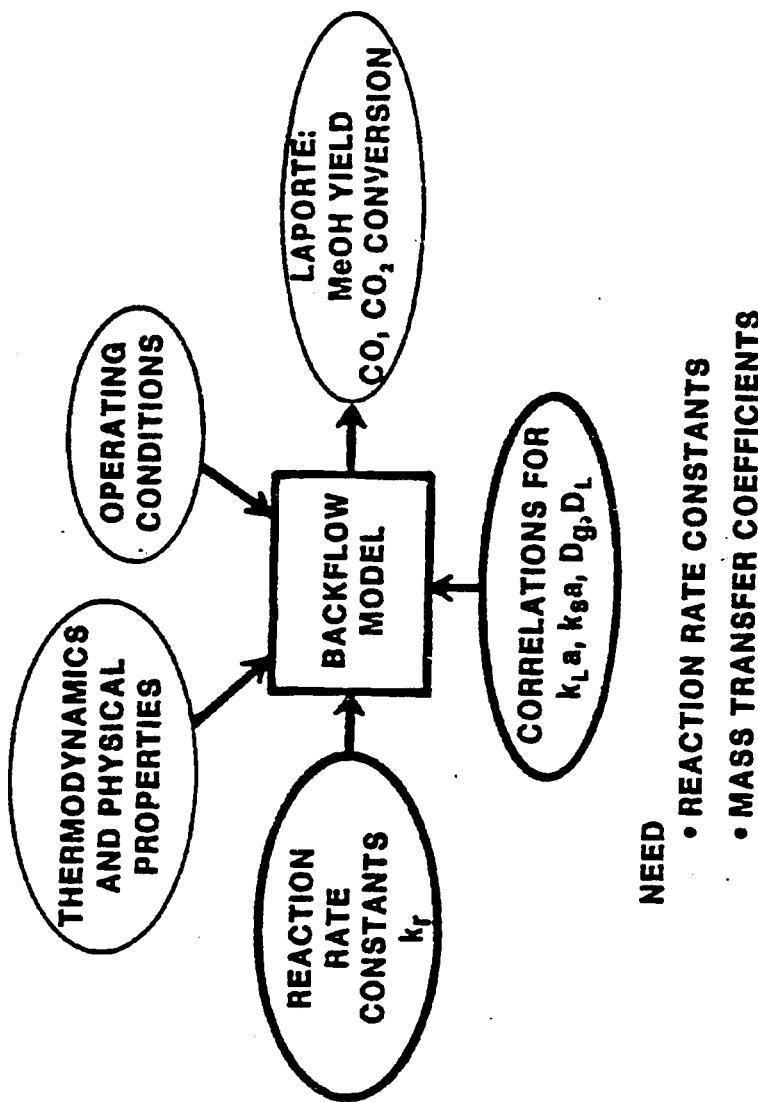
## **CALCULATION OF RATE CONSTANT**

**UNDER REACTION RATE CONTROL,**

$$\begin{aligned}\text{RATE} \\ (\text{MEASURED}) &= k_L a (C_{eq} - C_L) \\ &= k_{sa} (C_L - C_s) \\ &= k_r w_c (C_s - C_{seq})\end{aligned}$$

**KNOWING  $k_L a$ ,  $k_{sa}$ , A STEPWISE CALCULATION GIVES  $k_r$**   
 **$k_{sa}$  CORRELATIONS AVAILABLE: NEED  $k_L a$  DETERMINATION**





NEED

- REACTION RATE CONSTANTS
- MASS TRANSFER COEFFICIENTS

## LAPORTE PDU : $k_L a$ DETERMINATION

### CORRELATIONS USED:

- AKITA AND YOSHIDA (1973) FOR  $k_L a$ :

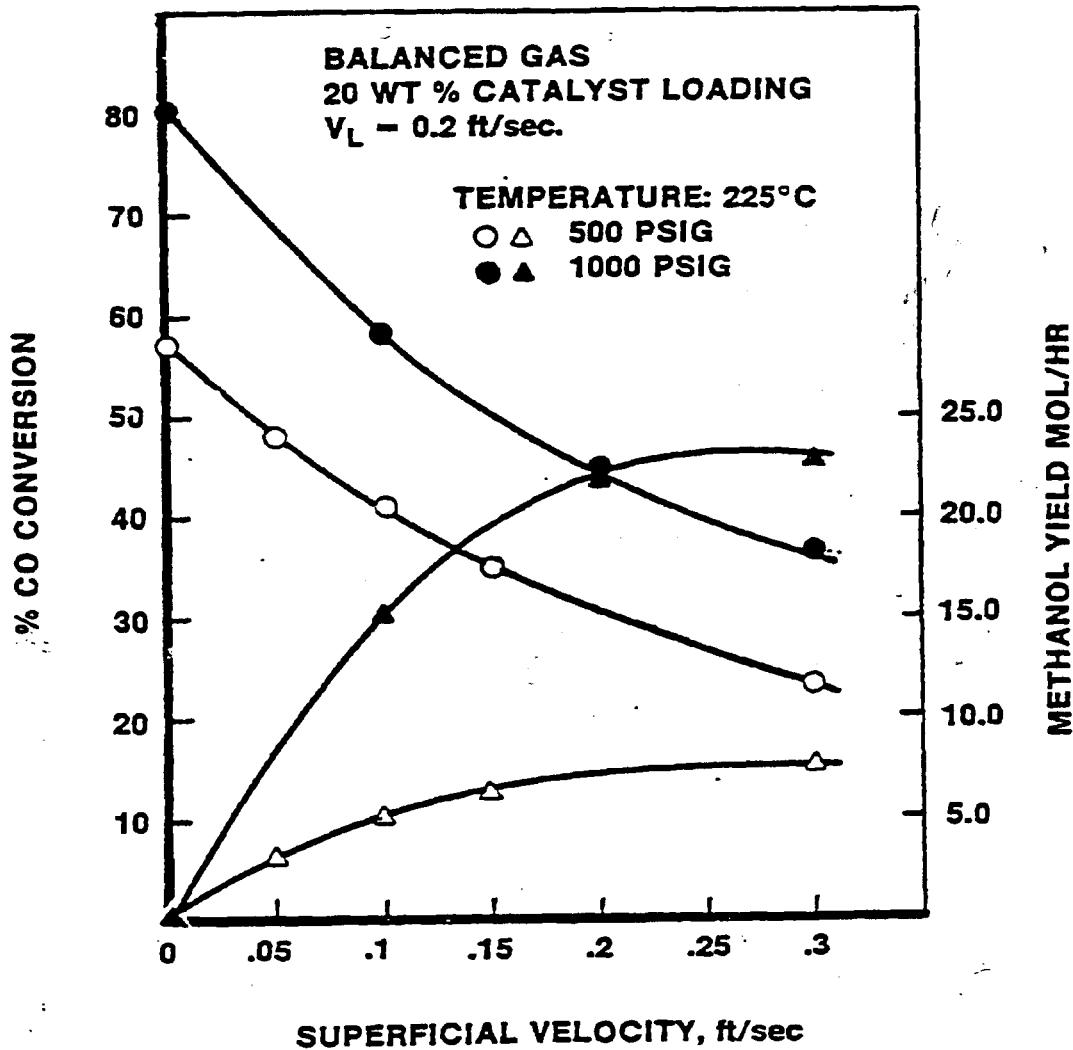
$$\frac{k_L a \cdot D_C^2}{D_i} = 0.6 \left( \frac{\nu_L}{D_i} \right)^{0.5} \cdot \left( \frac{g D_C^2 \rho_L}{\sigma} \right)^{0.62} \cdot \left( \frac{g D_C^3}{\nu_L^2} \right)^{0.31} \cdot (E_g)^{1.1}$$

- SANGER AND DECKWER (1981) FOR  $k_{sa}$ :

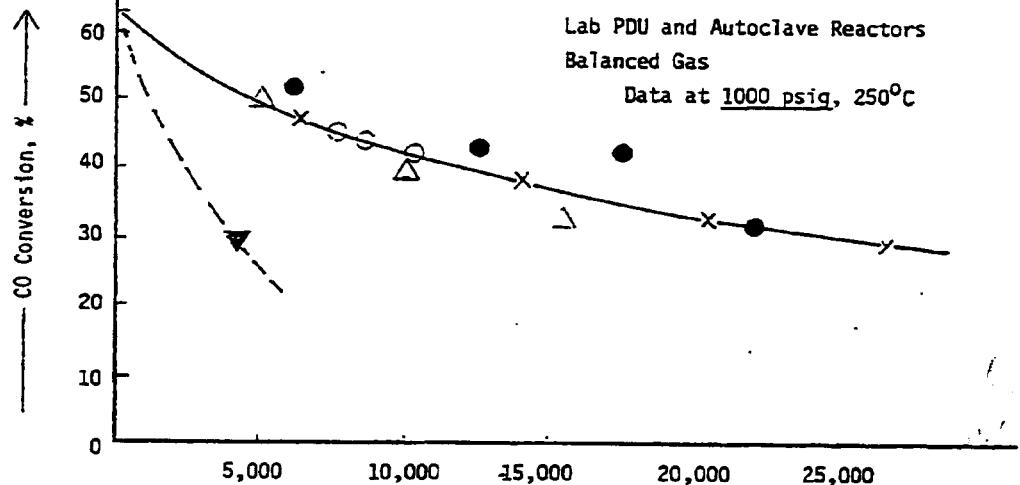
$$\frac{k_s d_p}{D_i} = 2 + 0.545 \left( \frac{\mu_L}{\rho_L D_i} \right)^{0.33} \cdot \left( \frac{g V_g d_p^4}{\nu_L^3} \right)^{0.264}$$

- THE GAS HOLD-UP,  $E_g$ , IS CALCULATED FROM CSI'S PDU RESULTS,

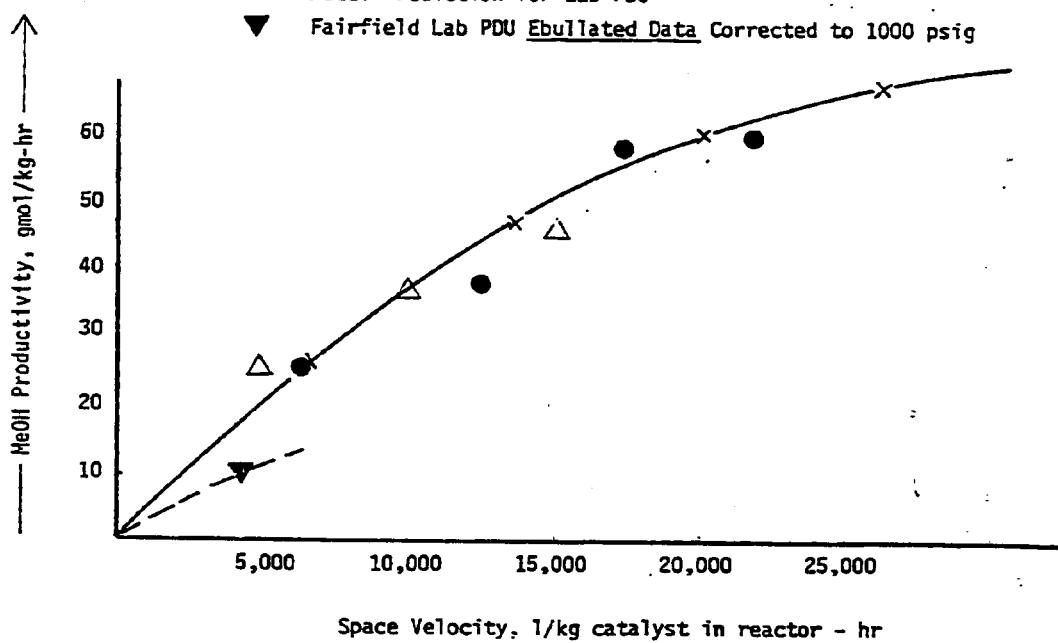
$$E_g = 0.5 (1 - \exp (-0.1735 V_g))$$



## LPMeOH SLURRY PERFORMANCE



- Fairfield Lab DPU Data, Using Measured  $E_g$  to Calculate Catalyst in the Reactor
- CSI 2-liter Autoclave data
- △ APCI 1-liter Autoclave data
- × Model Prediction for Lab PDU
- ▼ Fairfield Lab PDU Ebullated Data Corrected to 1000 psig

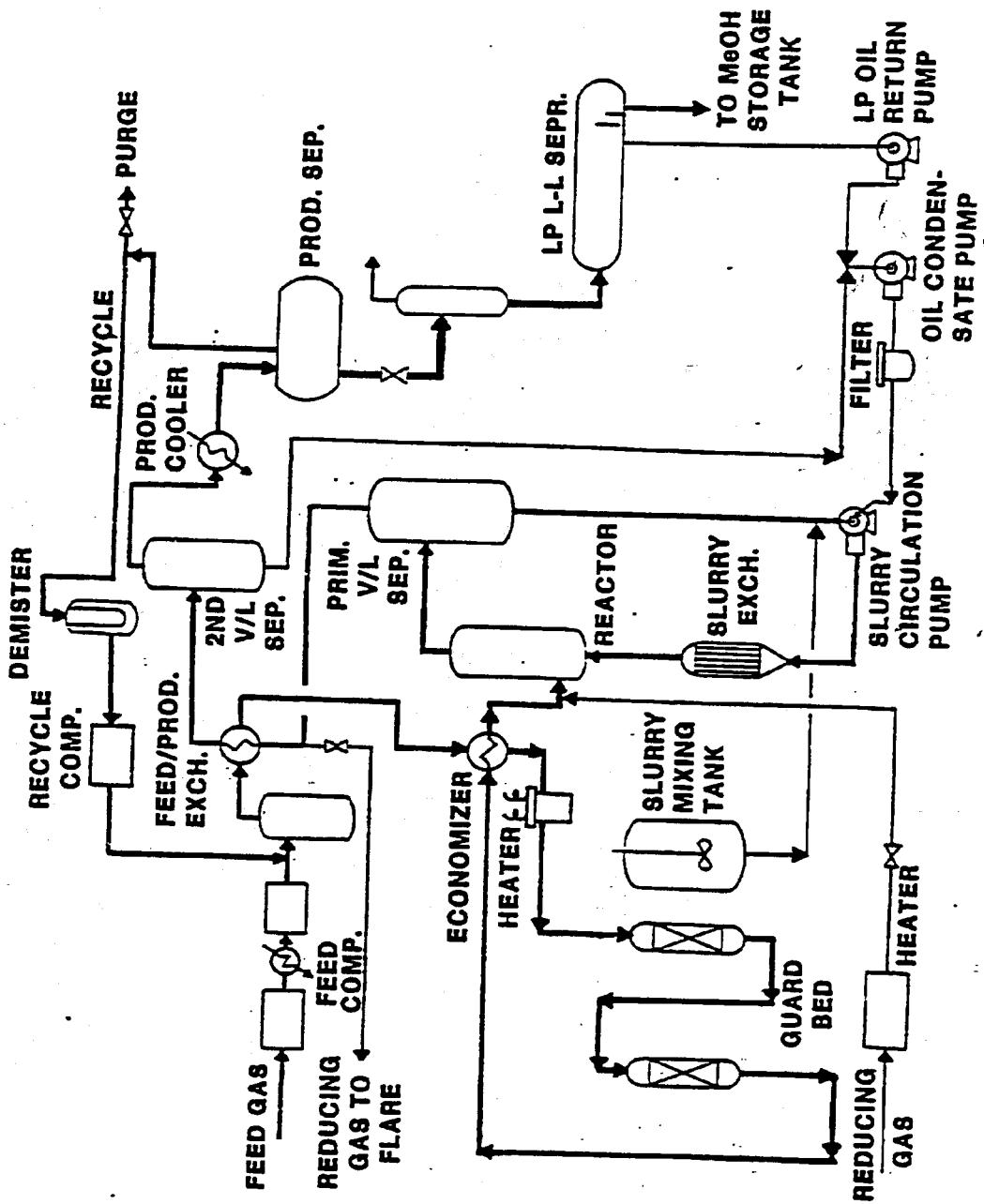


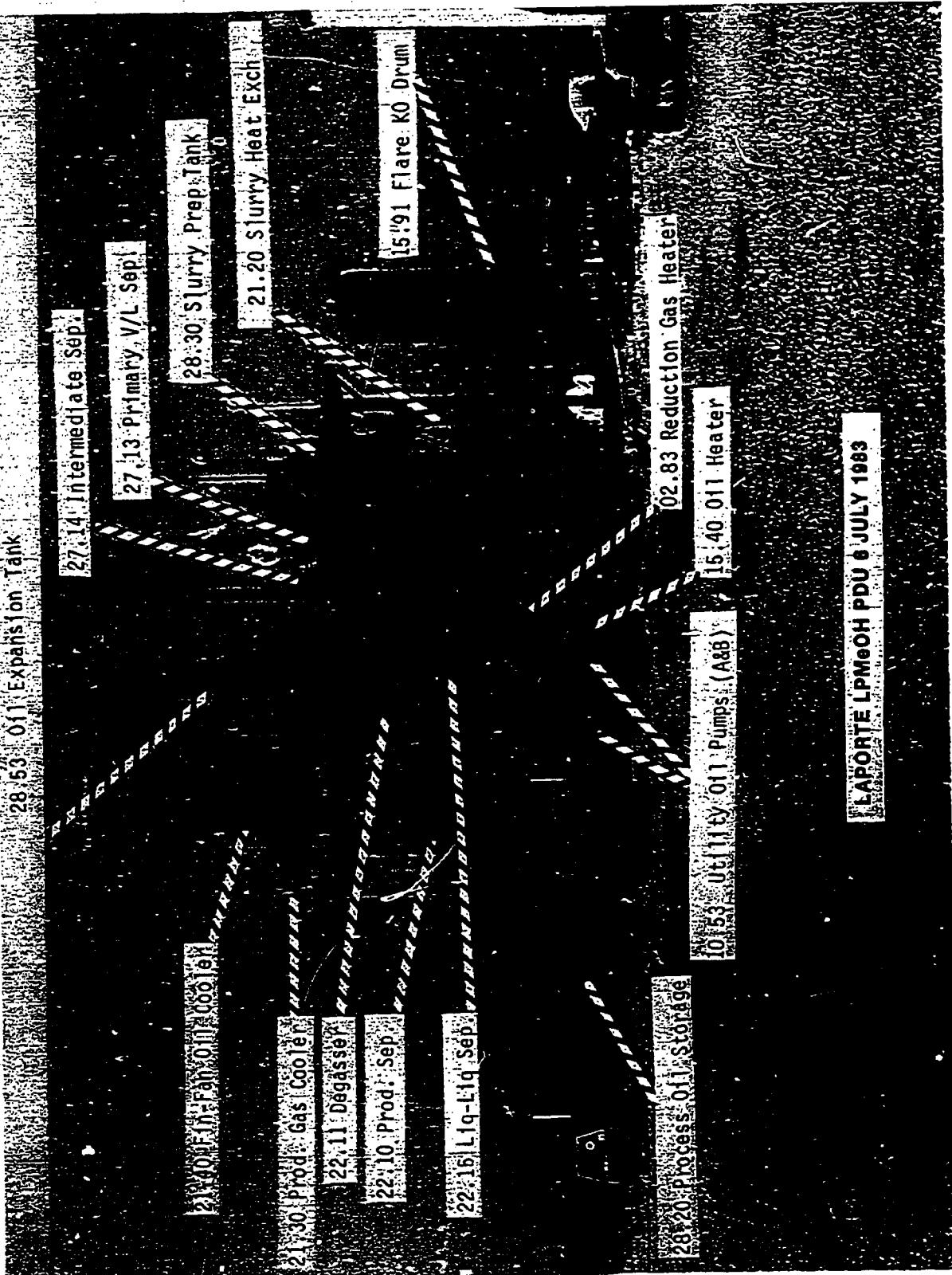
## I-APORTE LPMeOH PDU ENGINEERING STATUS

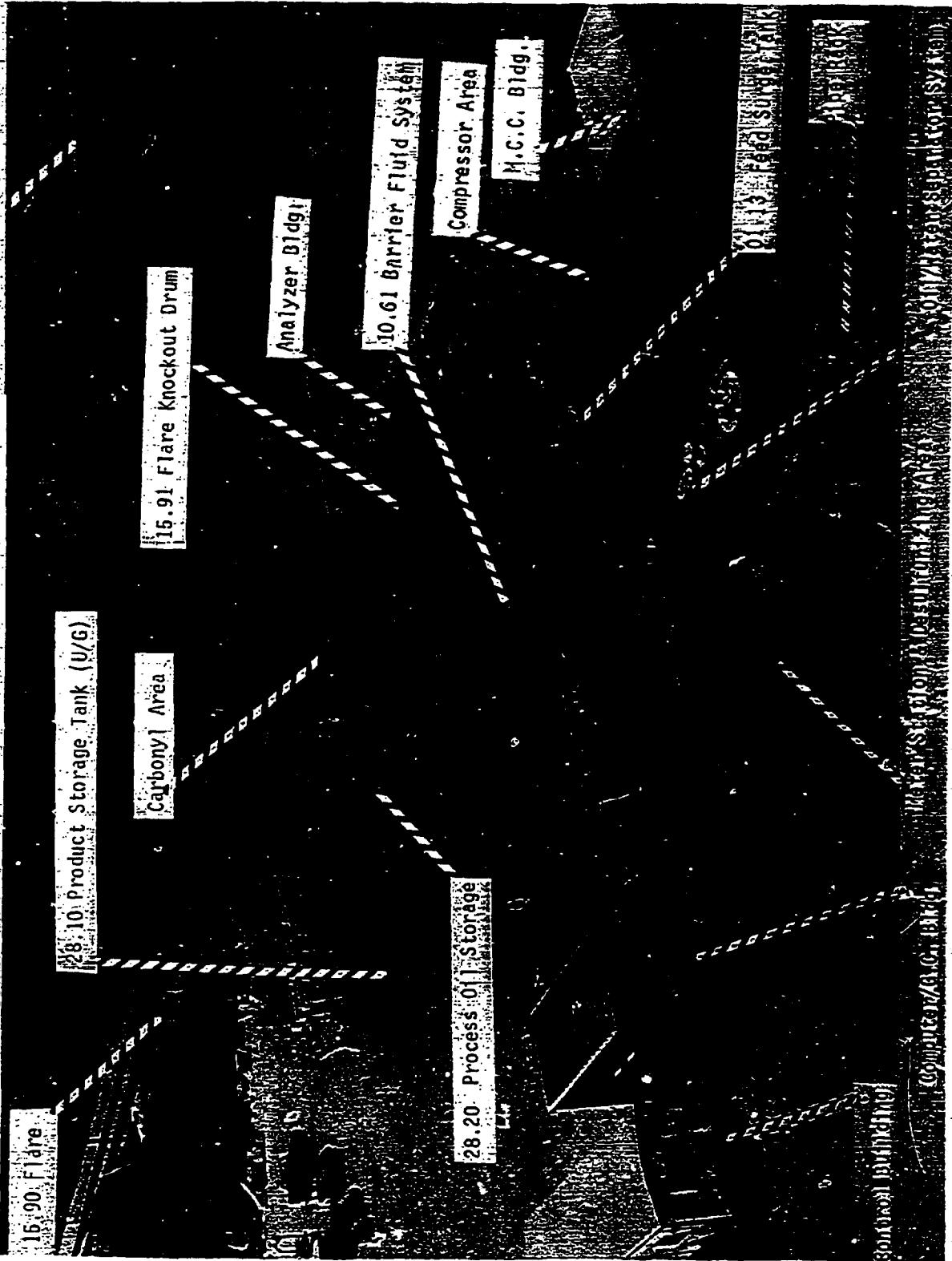
- LPM UNIT RELOCATION / INSPECTION
  - PROCESS
  - EQUIPMENT
  - MECHANICAL
  - INSTRUMENTATION/ELECTRICAL

COMPLETE  
COMPLETE  
COMPLETE  
95% COMPLETE  
60% COMPLETE

SIMPLIFIED PROCESS FLOWSHEET FOR  
LAPORTE LP MeOH PDU







27.13 PRIMARY V/L SEPARATOR

28.53 OIL EXPANSION TANK

27.10 REACTOR

28.30 SLURRY PREP TANK

VENT HEADER LINE

16.91 FLARE KNOCKOUT DRUM

21.40 UTILITY OIL COOLER

21.10 FEED/PRODUCT EXCHANGER

22.15 LP LIQ-LIQ SEP

22.10 PRODUCT SEPARATOR

ANALYZER BUILDING

28.20 PROCESS OIL STORAGE  
LAPORTE LPMeOH PDU 30 AUG 1983

28.60 SLURRY PUMP

22.20 SLURRY BYPASS