

LITERATURE CITED

- Bukur, D. B., D. Petrović, and J. G. Daly, "Hydrodynamics of Fischer-Tropsch Synthesis in Slurry Bubble Column Reactors," *Proc. DOE Indirect Liquefaction Contractors' Mtg.*, Houston, TX, 479 (Dec. 2-5, 1985).
- Bukur, D. B., D. Petrović, and J. G. Daly, "Flow Regime Transitions in a Bubble Column with a Paraffin Wax as the Liquid Medium," *Ind. Eng. Chem. Res.*, **26**, 1087 (1987a).
- Bukur, D. B., S. A. Patel, and R. Matheo, "Hydrodynamic Studies in Fischer-Tropsch Derived Waxes in a Bubble Column," *Chem. Eng. Commun.*, **60**, 63 (1987b).
- Calderbank, P. H., F. Evans, R. Farley, G. Jepson, and A. Poll, "Rate Processes in the Catalyst-Slurry Fischer-Tropsch Reaction," *Catalysis in Practice - Instn. Chem. Engrs.*, 66 (1963).
- Cova, D. R., "Catalyst Suspension in Gas-Agitated Tubular Reactors," *Ind. Eng. Proc. Des. Dev.*, **5**, 20 (1966).
- Deckwer, W.-D., *Oil Gas J.*, **78**, No. 45, 198 (1980).
- Deckwer, W.-D., Y. Louisi, A. Zaidi, and M. Ralek, "Hydrodynamic Properties of the Fischer-Tropsch Slurry Process," *Ind. Eng. Chem. Process Des. Dev.*, **19**, 699 (1980).
- Far, L. S., S. Satija, and K. Wisecarver, "Pressure Fluctuation Measurements and Flow Regime Transitions in Gas-Liquid-Solid Fluidized Beds," Paper 67E presented at AIChE Annual Meeting (1984).
- Glasgow, L. A., L. E. Erickson, C. H. Lee, and S. A. Patel, "Wall Pressure Fluctuations and Bubble Size Distributions at Several Positions in an Airlift Fermenter," *Chem. Eng. Comm.*, **29**, 311 (1984).
- Kato, Y., A. Nishiwaki, T. Fukuda, and S. Tanaka, "The Behavior of Suspended Solid Particles and Liquid in Bubble Columns," *J. Chem. Eng. Jap.*, **5**, 112 (1972).
- Kölbel, H., and M. Ralek, "The Fischer-Tropsch Synthesis in the Liquid Phase," *Cat. Rev. Sci. Eng.*, **21**, 225 (1980).
- Kuo, J. C. W., "Two-Stage Process for Conversion of Synthesis Gas to High Quality Transportation Fuels," Mobil Research and Development Co., Final Report to the Dept. of Energy for Contract DE-AC22-83PC60019 (1985).
- Lee, C. H., "Dynamics of Bubble Size Distribution and Wall Pressure Fluctuations in Airlift Fermentors," Masters Thesis, Kansas State University (1983).
- O'Dowd, W., D. N. Smith, J. A. Ruether, and S. C. Saxena, "Gas and Solids Behavior in a Baffled and Unbaffled Slurry Bubble Column," *AIChE J.*, **33**, 1959 (1987).
- Smith, D. N., and J. A. Ruether, "Dispersed Solid Dynamics in a Slurry Bubble Column," *Chem. Eng. Sci.*, **40**, 741 (1985).
- Zigrang, D. J., and N. D. Sylvester, "An Explicit Equation for Particle Settling Velocities in Solid-Liquid Systems," *AIChE J.*, **27**, 1043 (1980).

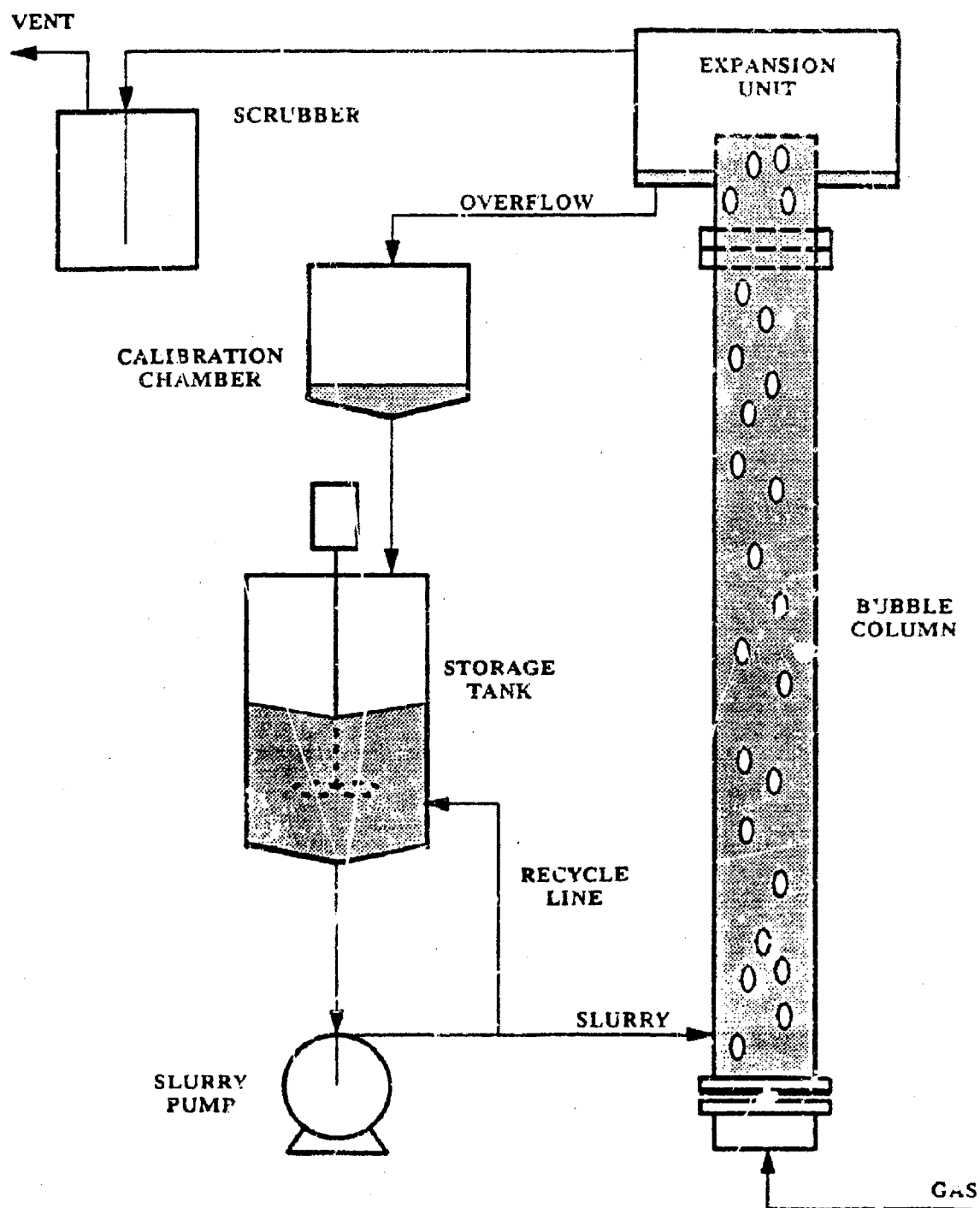


Figure 1. Schematic of the slurry bubble column apparatus.

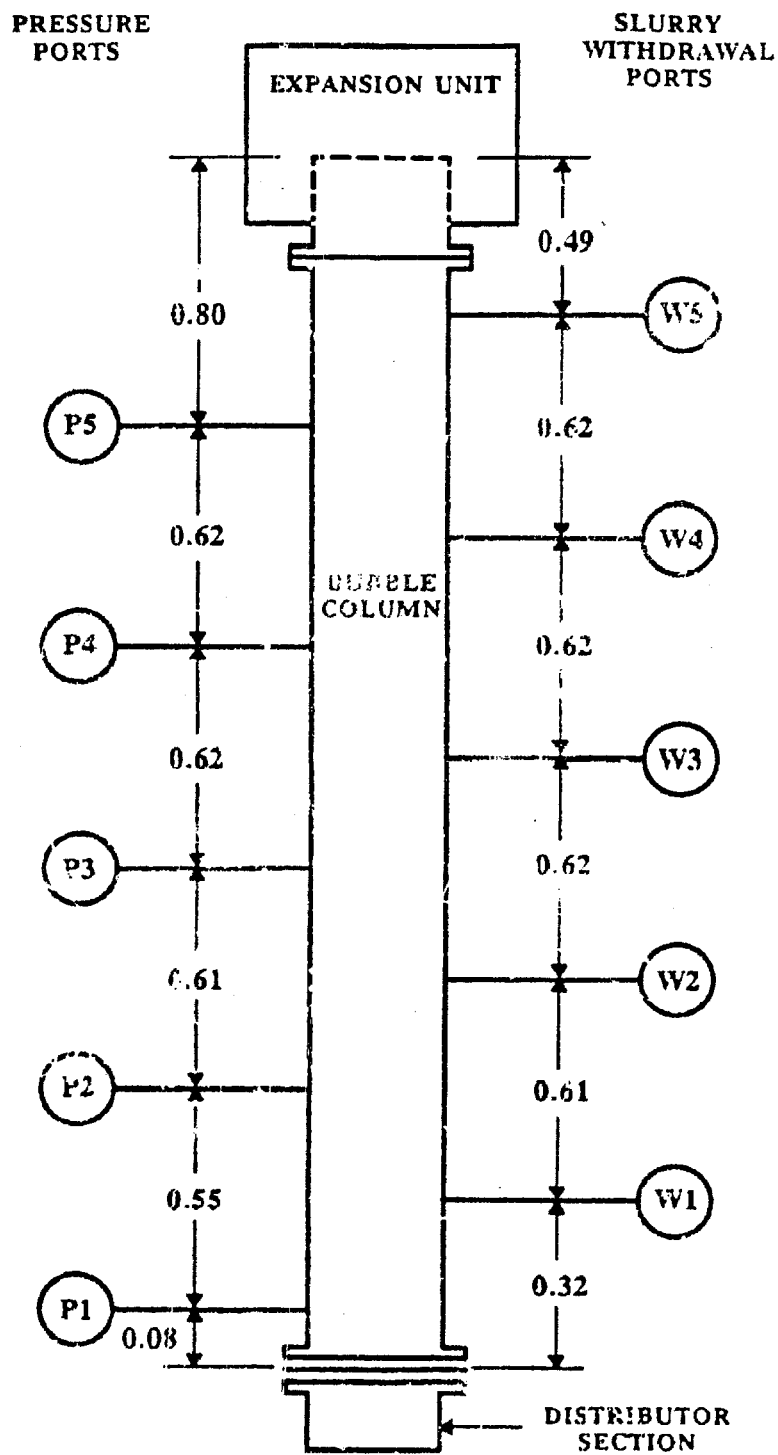


Figure 2. Schematic showing DP cell and sampling port locations.

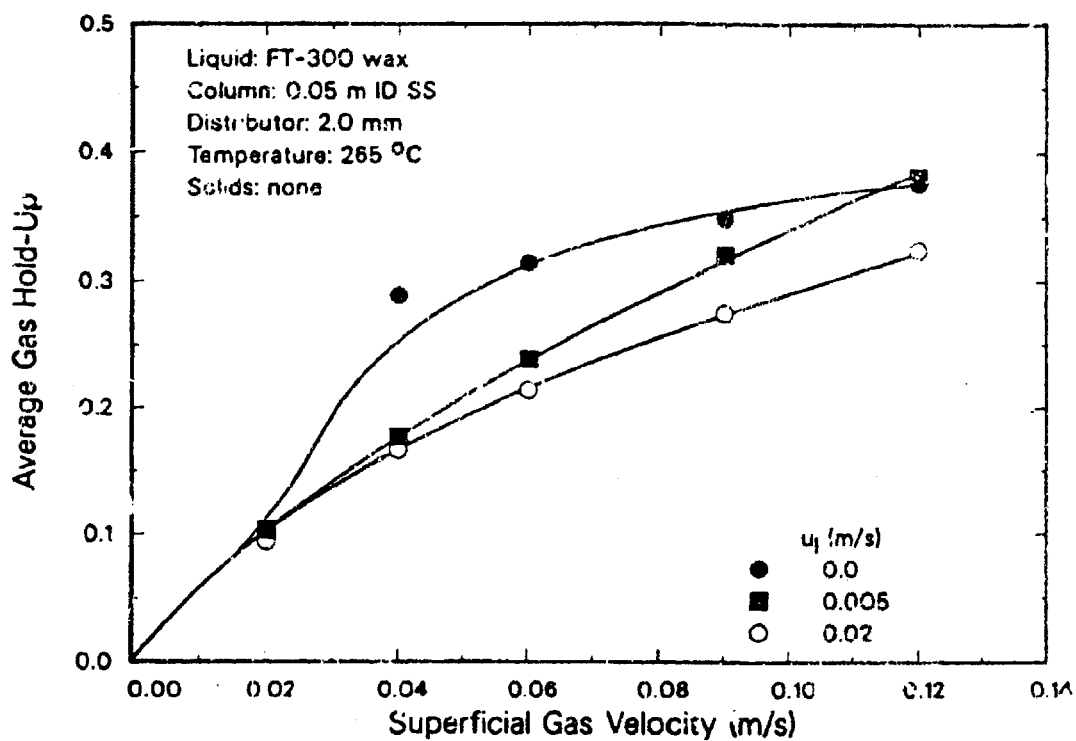


Figure 3. Effect of liquid circulation on average gas holdup in the absence of solids

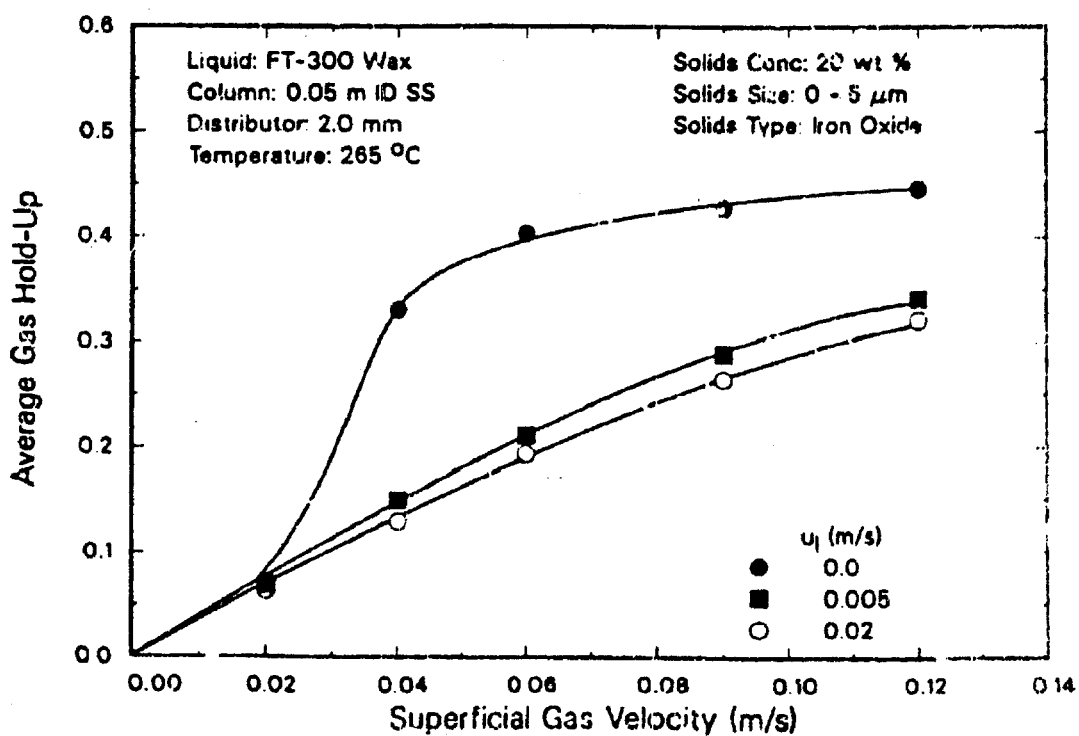


Figure 4. Effect of liquid circulation on average gas holdup in the presence of solids

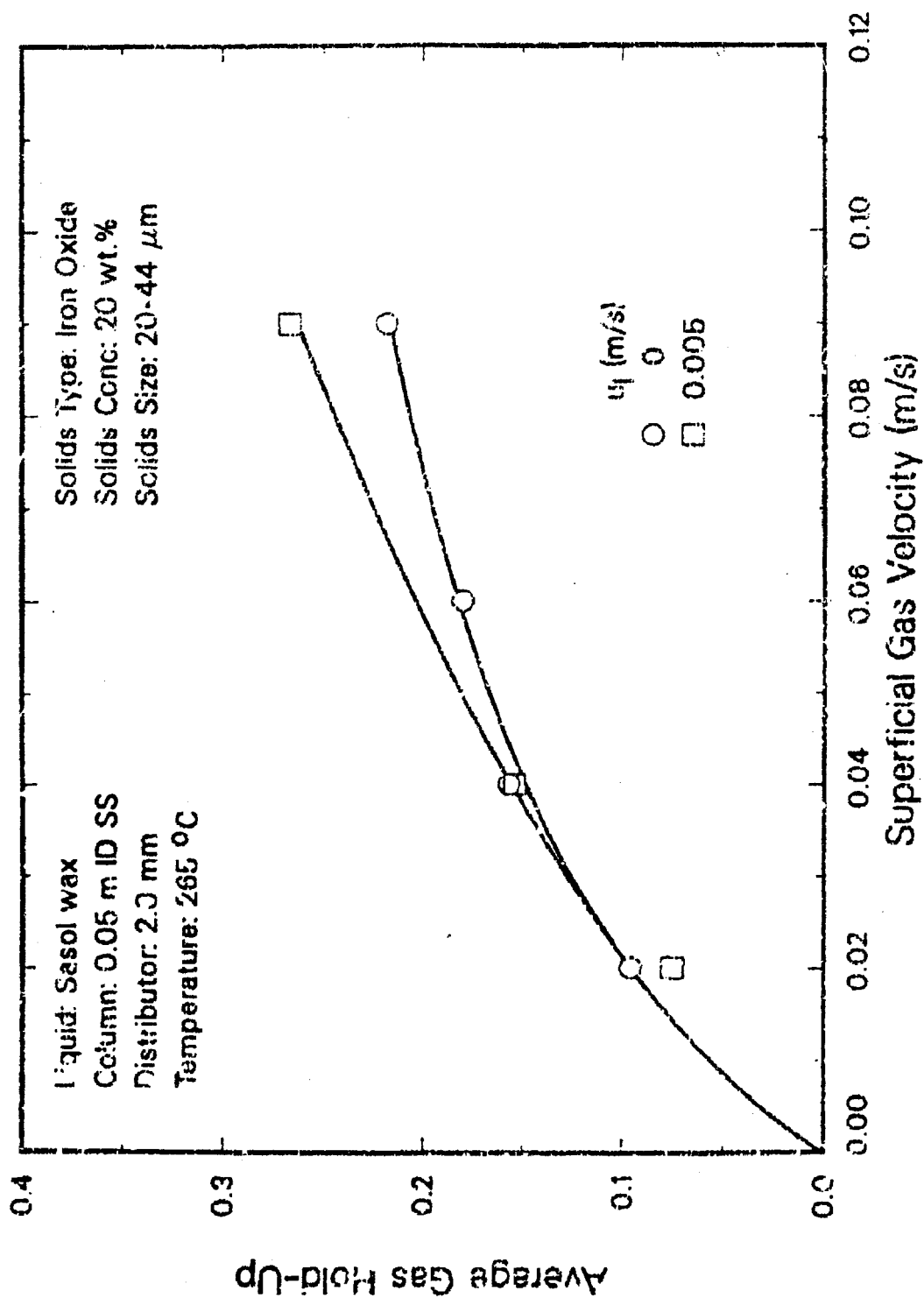


Figure 5. Effect of liquid circulation on average gas holdup for Sasol wax in the presence of solids.

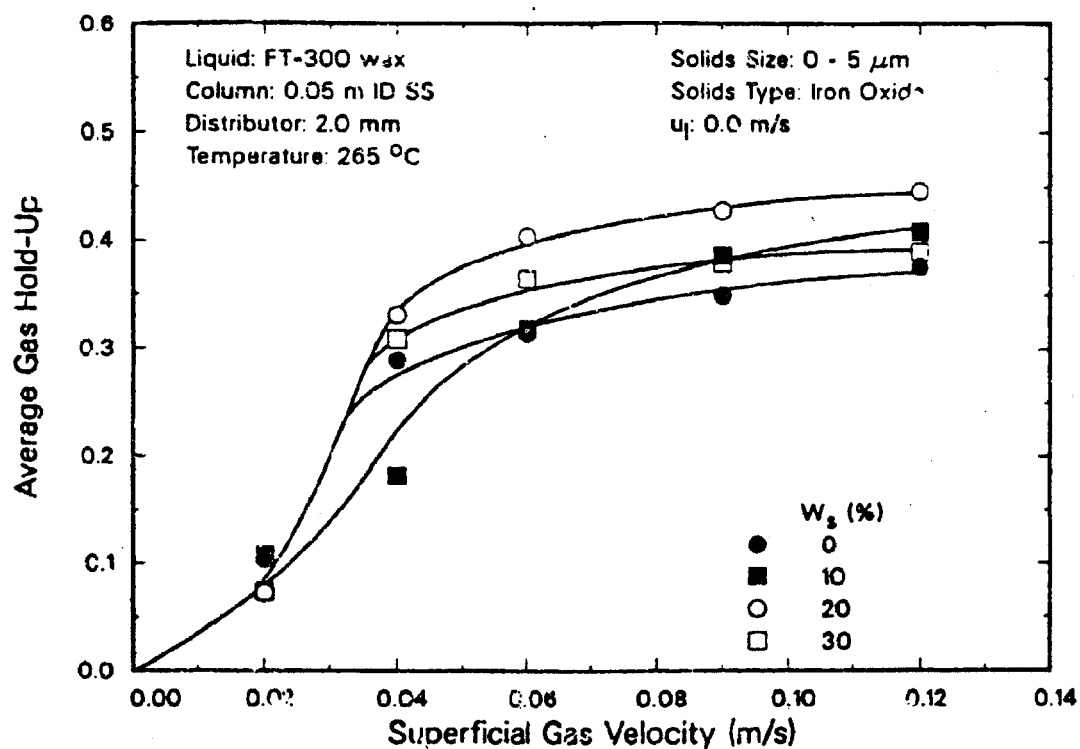


Figure 6. Effect of solids concentration on average gas holdup during batch mode of operation.

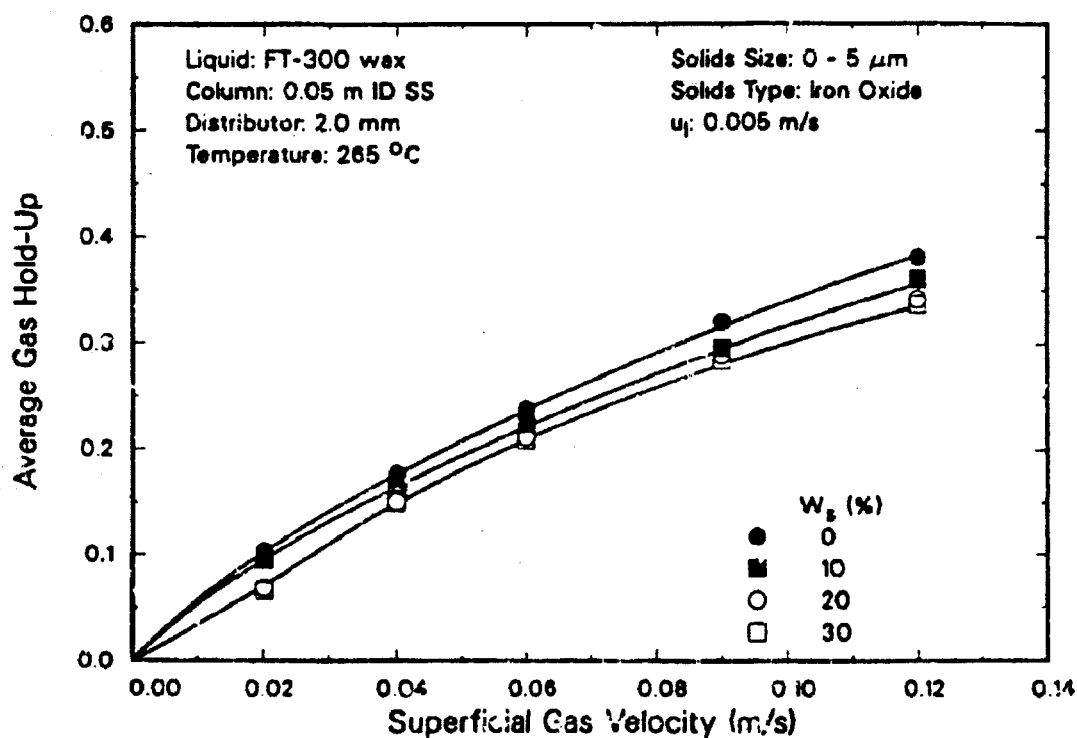


Figure 7. Effect of solids concentration on average gas holdup during continuous mode of operation.

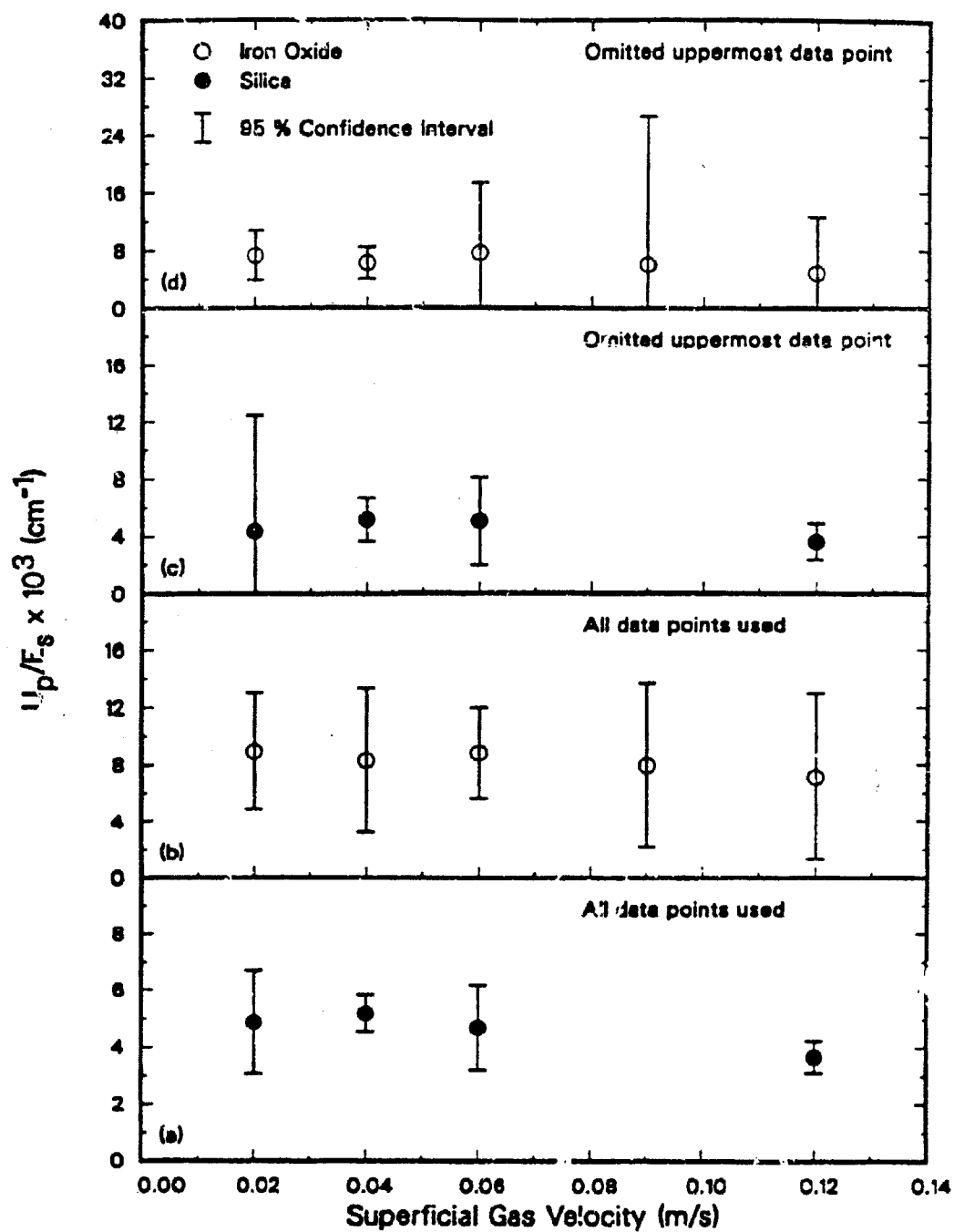


Figure 8. Effect of superficial gas velocity on U_p/E_s for iron oxide and silica (FT-300, 265°C, batch mode, 20 wt.% of 20-44 μm particles).

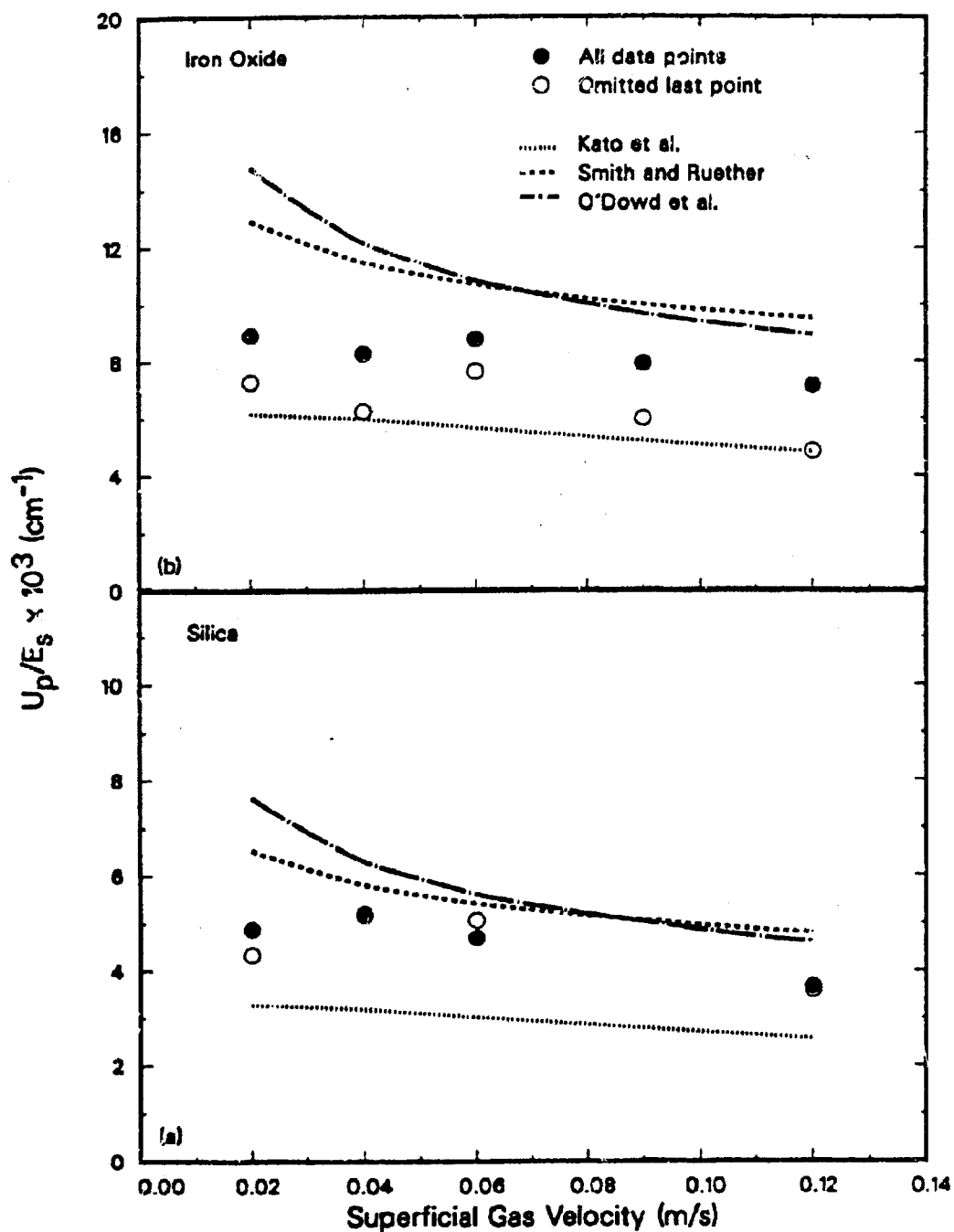


Figure 9. Comparison of U_p/E_s with literature for iron oxide and silica (FT-300, 265°C, batch mode, 20 wt.% of 20-44 μm particles).

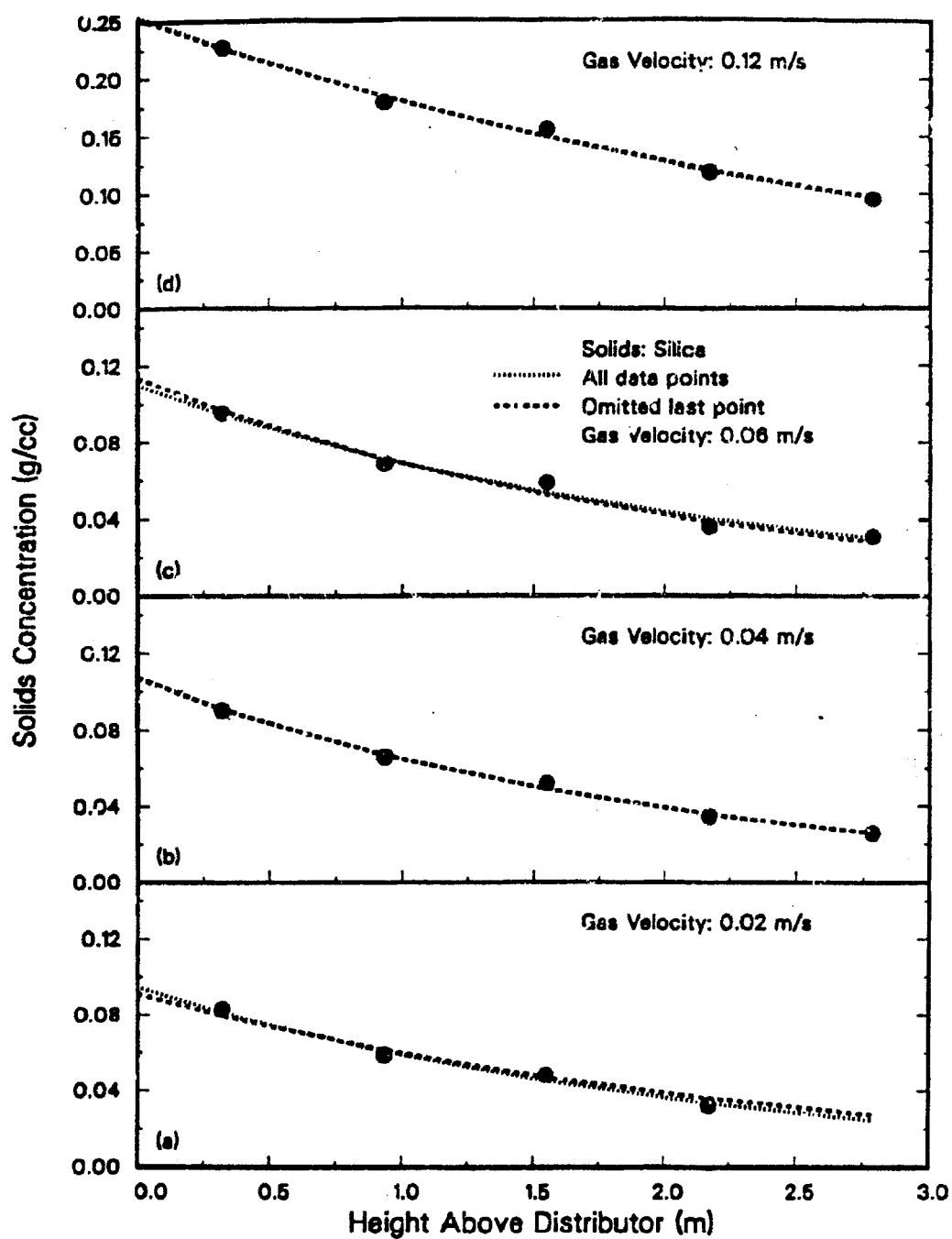


Figure 10. Comparison of measured axial solids concentration profiles for silica with values predicted by the semi-infinite dispersion model (FT-300, 265°C, batch mode, 20 wt.% of 20-44 μ m particles).

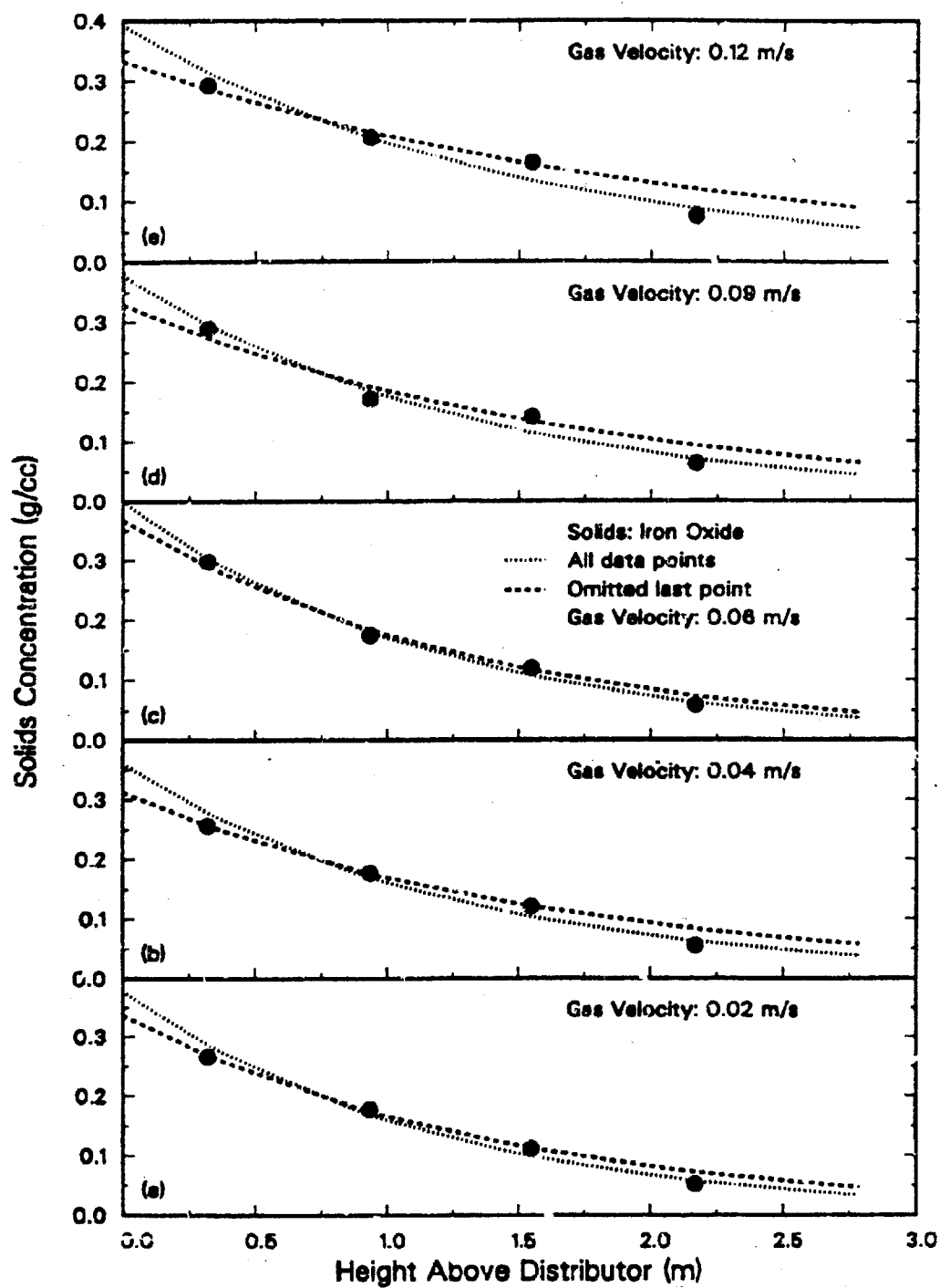


Figure 11. Comparison of measured axial solids concentration profiles for iron oxide with values predicted by the semi-infinite dispersion model (FT-300, 265°C, batch mode, 20 wt.% of 20-44 μ m particles).

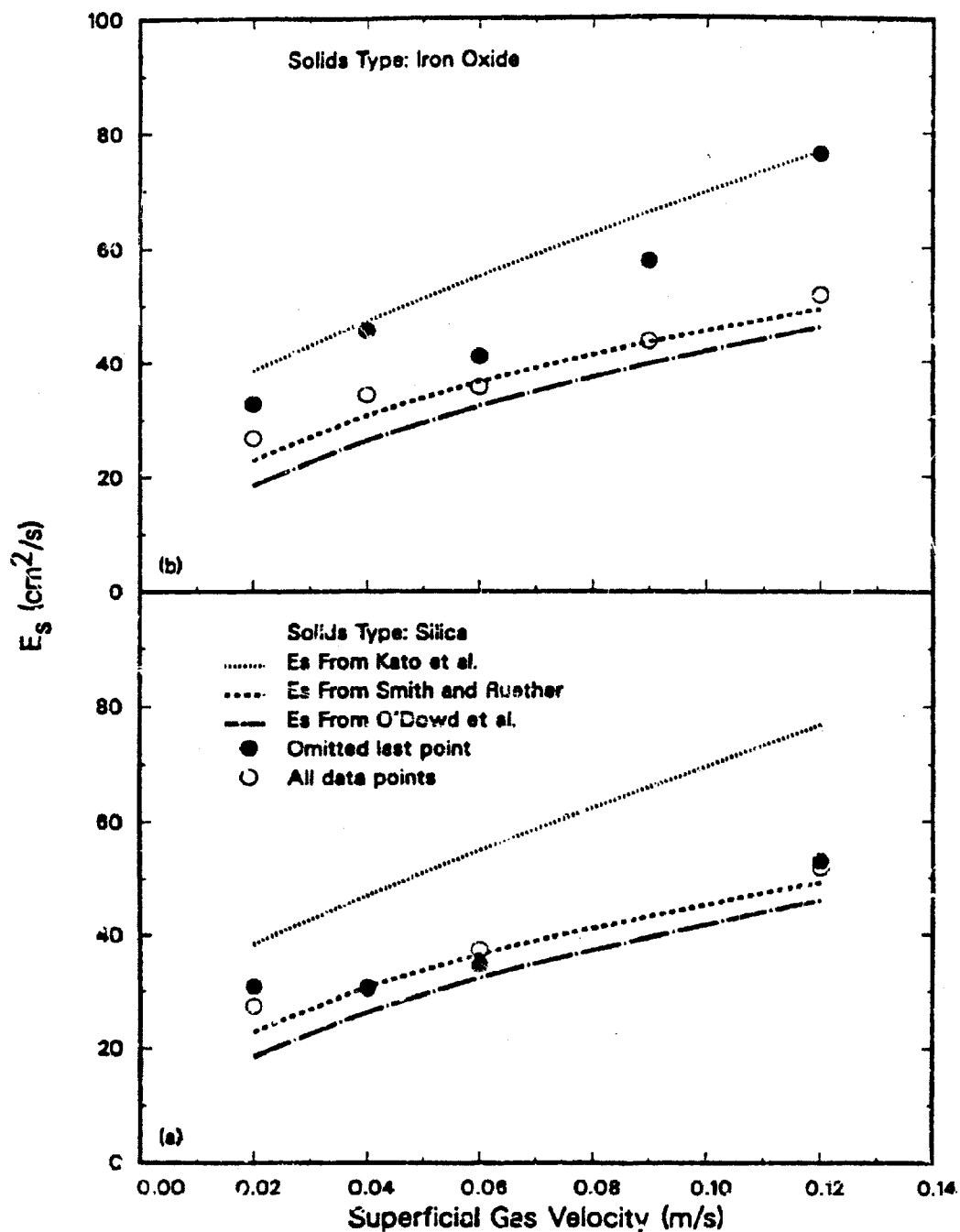


Figure 12. Effect of superficial gas velocity on axial solids dispersion coefficient using Kato et al.'s correlation for hindered settling velocity (FT-300, 265°C, batch mode, 20 wt.% of 20-44 μ m particles).

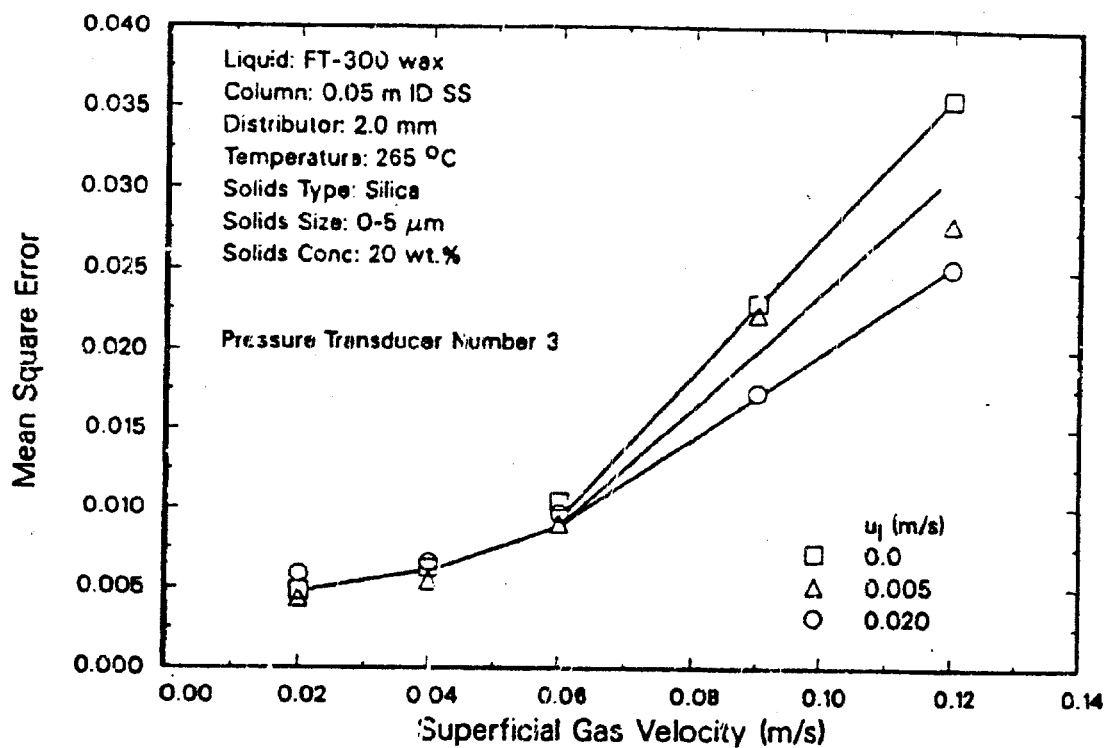


Figure 13 Effect of liquid circulation rate on the mean square error of pressure fluctuations at the wall.

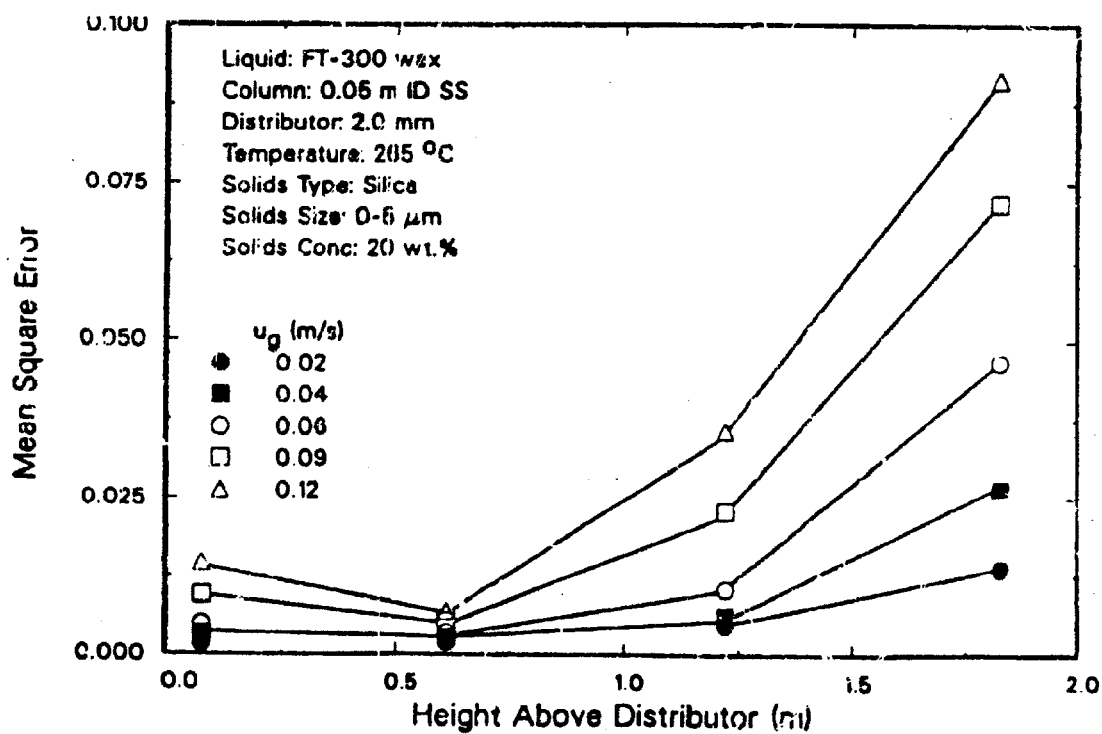


Figure 14 Effect of height above the distributor on the mean square error of pressure fluctuations at the wall

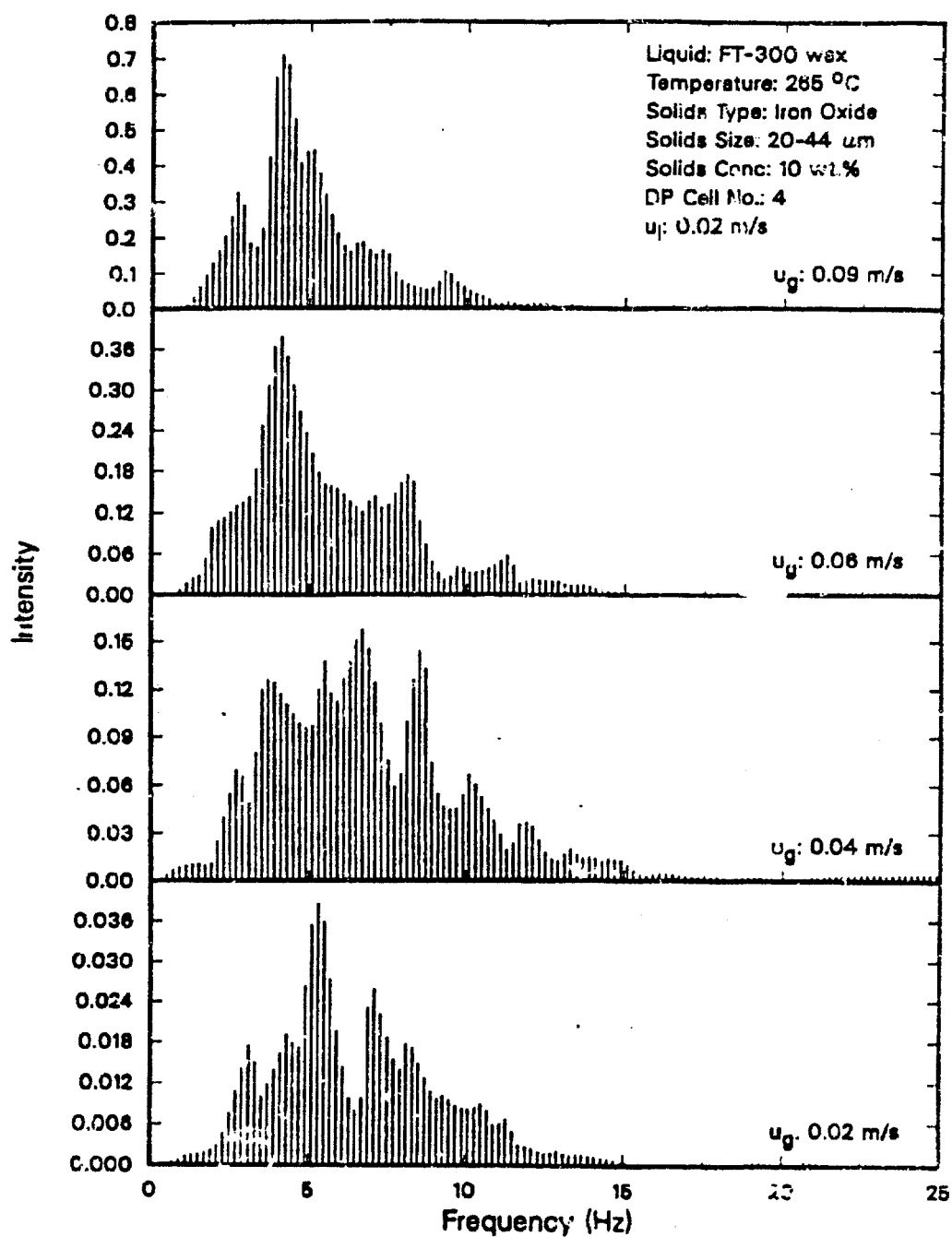


Figure 15. Effect of superficial gas velocity on the power spectral density function for pressure fluctuations at the wall.

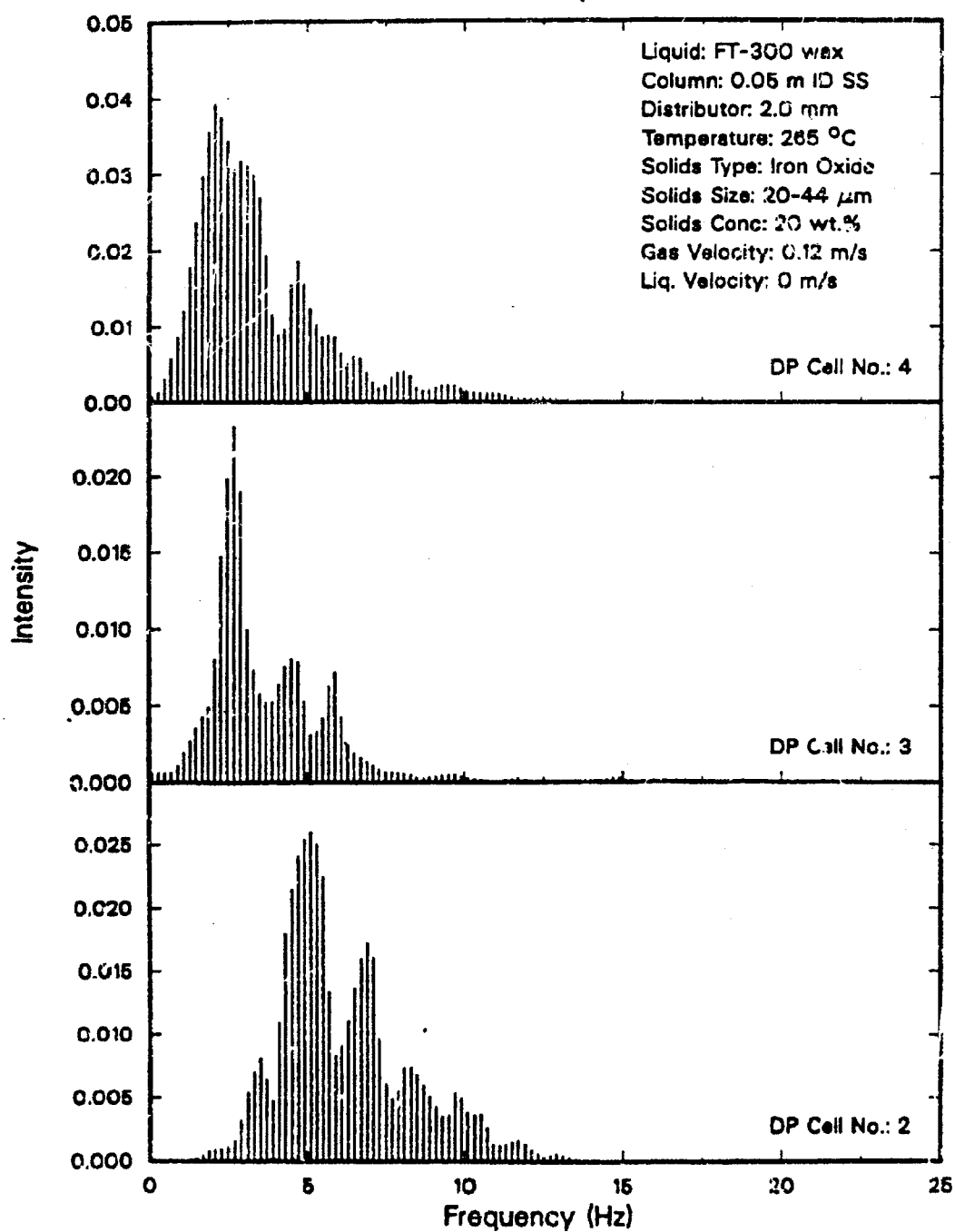


Figure 16. Effect of height above the distributor on the power spectral density for pressure fluctuations at the wall.

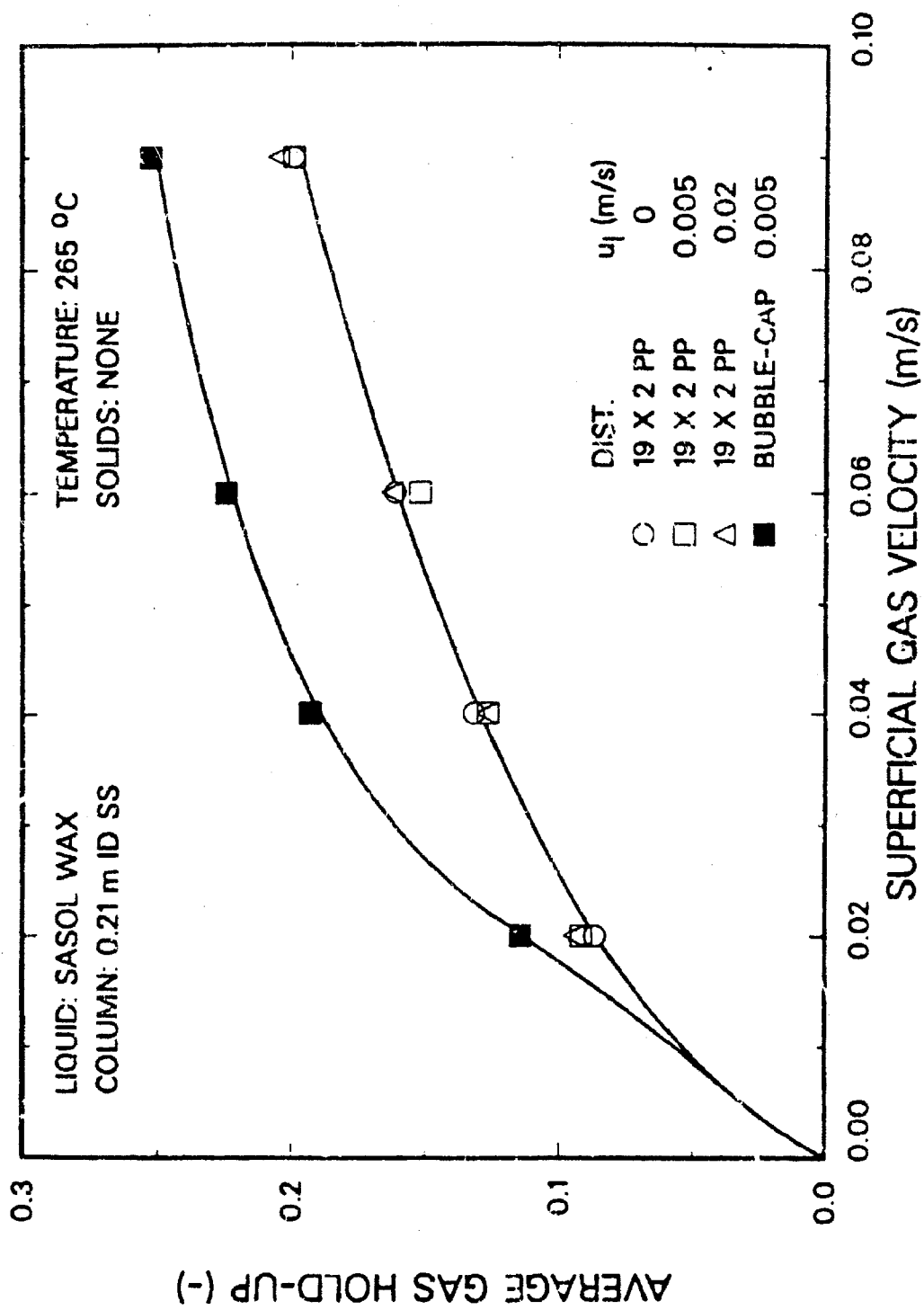


Figure 17. Effect of superficial gas velocity and distributor type on average gas holdup.

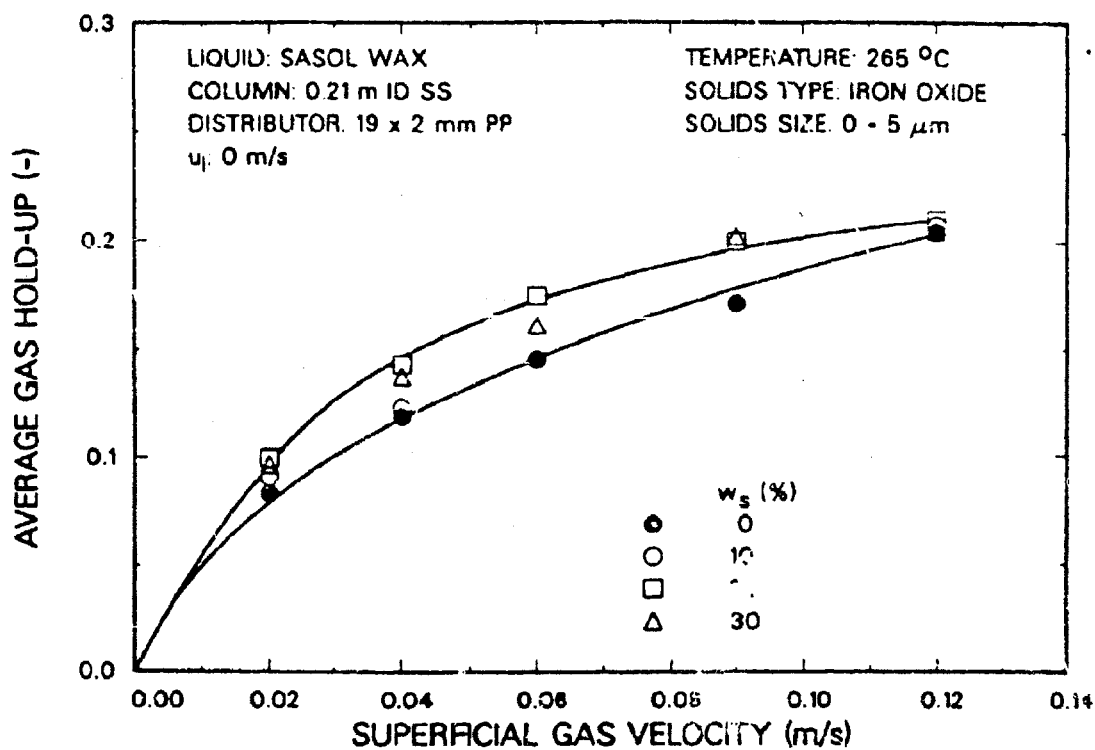


Figure 18. Effect of superficial gas velocity and solids concentration on average gas holdup (batch mode of operation).

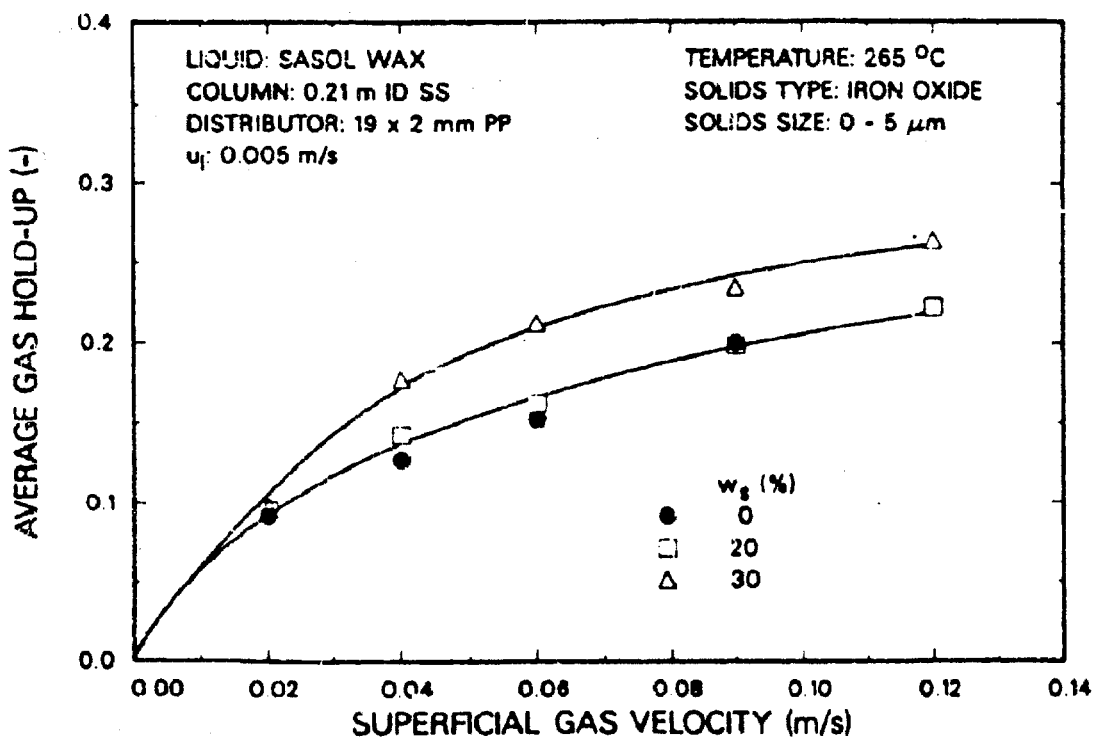


Figure 19. Effect of superficial gas velocity and solids concentration on average gas holdup (continuous mode of operation).

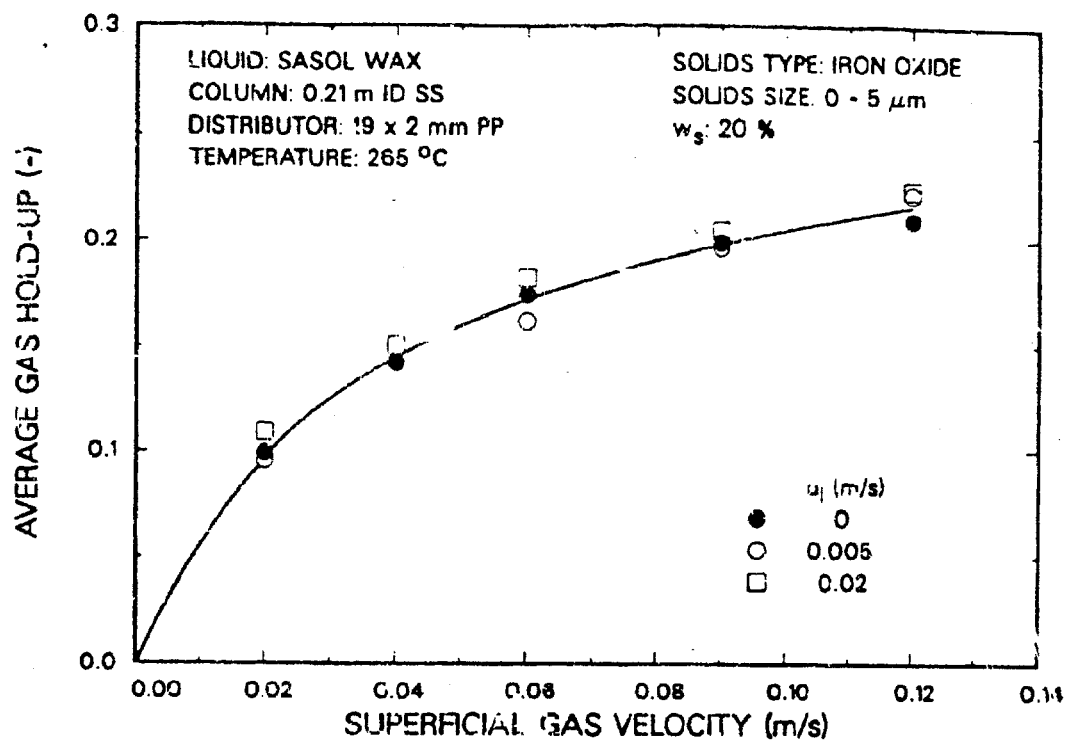


Figure 20 Effect of superficial gas velocity and liquid circulation on average gas holdup (20% solids present).

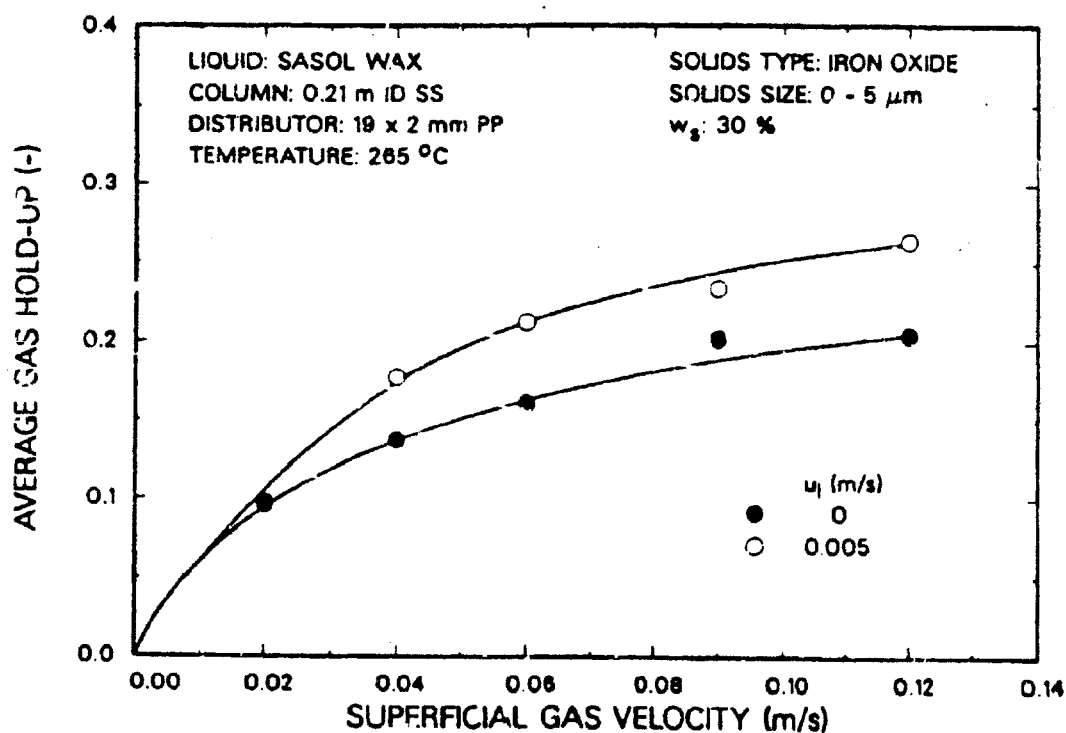


Figure 21 Effect of superficial gas velocity and liquid circulation on average gas holdup (30% solids present).

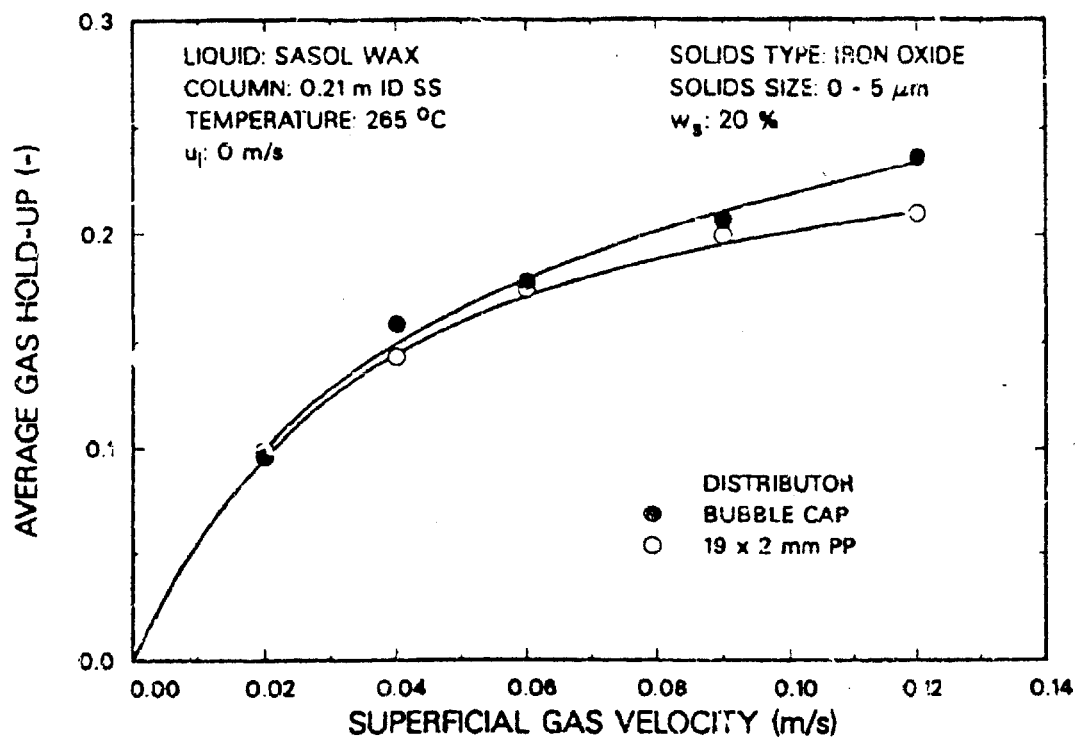


Figure 22 Effect of superficial gas velocity and distributor type on average gas holdup (batch mode).

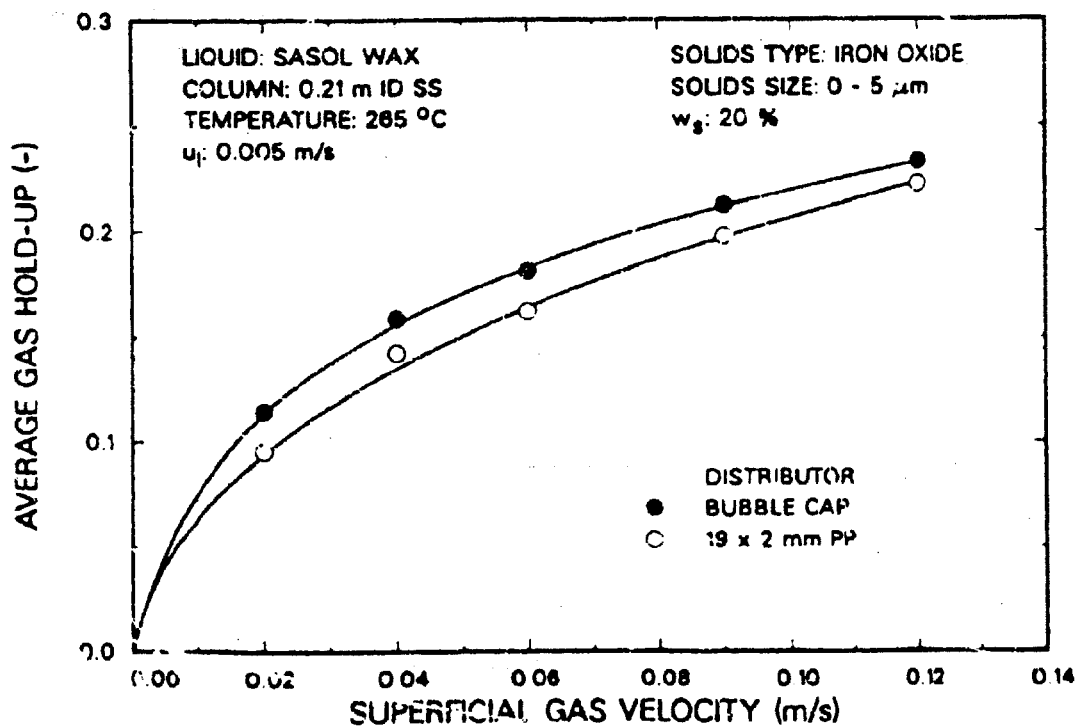


Figure 23 Effect of superficial gas velocity and distributor type on average gas holdup (continuous mode)

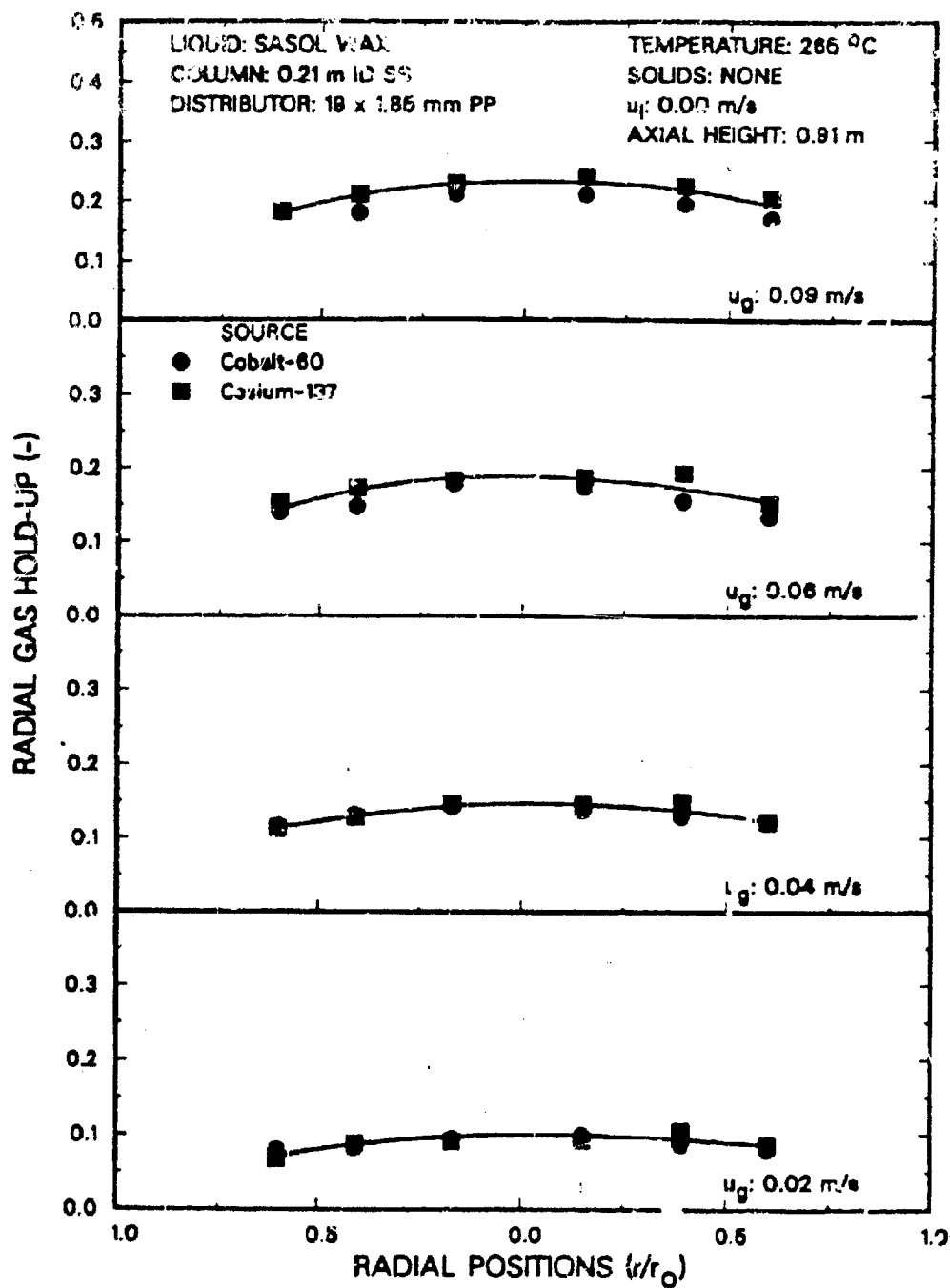


Figure 24. Effect of superficial gas velocity on radial gas holdup

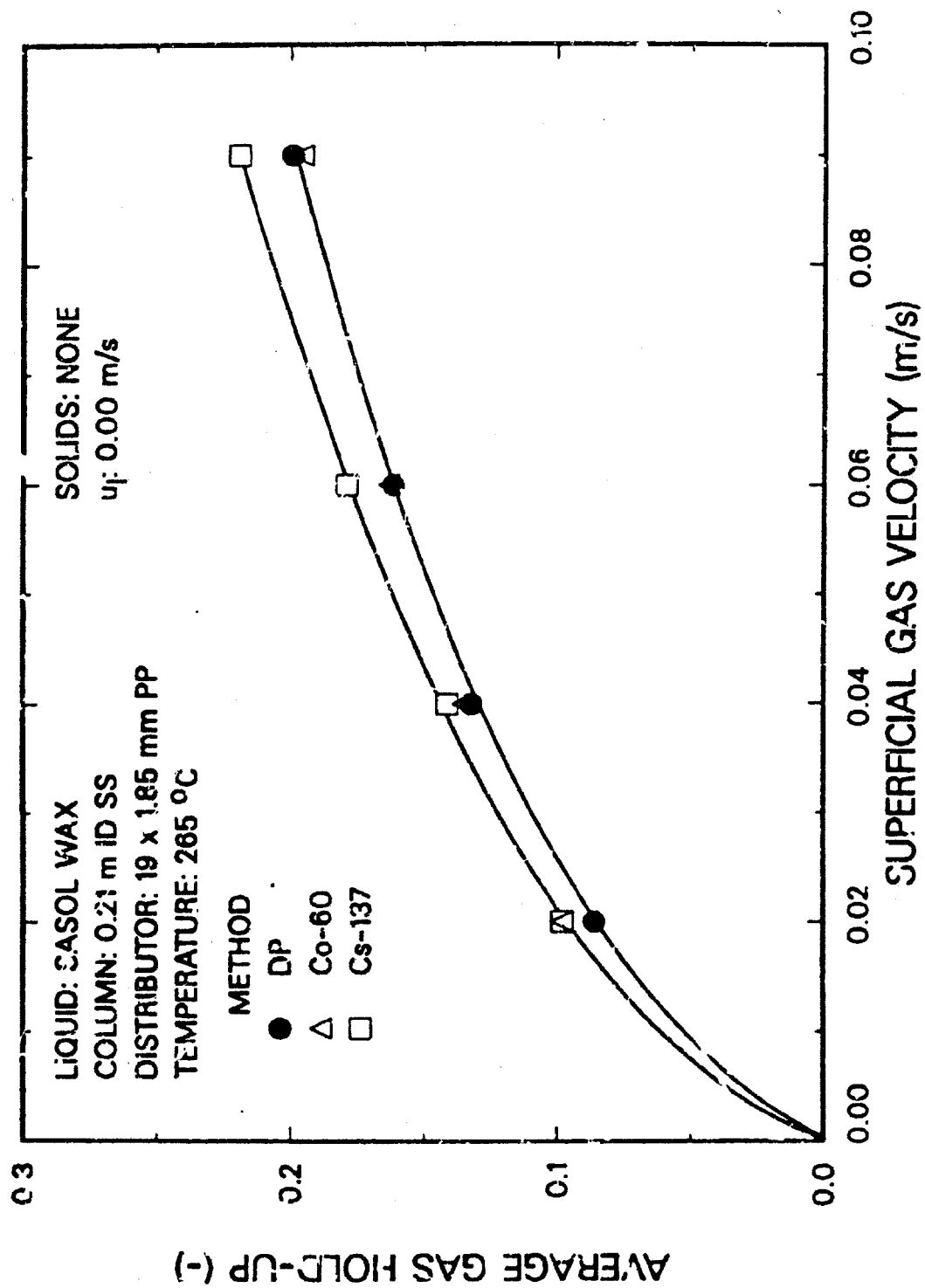


Figure 2.5. Comparison of average holdups from different techniques