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MEMORANDUM

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2 0 2 0 .LOW PRESSURE SCRUBBING OF CO₂ FROM RAW HYDROGENBY MEANS OF GIBOTOL AND DIAMINOPROPANE SOLVENTS.Introduction

This memorandum presents results of test work on an experimental CO₂ scrubbing system to determine the maximum capacity of a plant size system using packed towers, and to compare the relative advantages of Gibotol and diaminopropene as scrubbing media.

Summary

With the use of Raschig ring packing in place of the bubble plates in the present 6 ft. diameter plant CO₂ scrubbers, the maximum capacity of each will be as follows, using Gibotol (TM) as the scrubbing medium:

Size of Rings	CF Scrubbed H ₂ /day	50% T.M.	Circulated C.P.H.	Residual CO ₂ %
3/4 inch	4,000,000	156	0,0	
2 "	8,000,000	343	0,8	
3 " ()	9,300,000	400	0,0	

The results when using D.A.P. in the experimental system are as follows, translated in terms of the large scale scrubber:

Size of Rings	CF Scrubbed H ₂ / Day	50% D.A.P.	Circulated C.P.H.	Residual CO ₂ %
2 inch	8,000,000	169	0,0	
3 " "	16,000,000	300	0,0	

The capacities noted above for D.A.P. solution are not maxima. Higher rates have not been obtained due to shortage of experimental stripping capacity.

When using TM at 50% concentration, the present size stripper (6 ft. I.D.) will satisfactorily handle the amount of solution required when scrubbing about 9,000,000 CF H₂/day. When using D.A.P. at 50% concentration, the same stripper will handle the solution corresponding TM to 8,000,000 CF H₂/day.

4) Estimated.

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The relative steam consumption for T.E.A. and D.A.P. for stripping and preheating as obtained in the experimental system are as follows:

T.E.A. - 50 lbs. steam / MCF raw gas to scrubber
D.A.P. - 60 " " " " "

The higher D.A.P. steam consumption is due to the greater heat of CO₂ absorption, 900 B.t.u. vs. 600 B.t.u. per pound of CO₂ absorbed.

Discussion

1. General

A flow sheet of the experimental system is shown in Fig.1. The scrubber and actifier were designed to contain the same packed height as the large scale scrubber and actifier with bubble plates removed, so that the extrapolation from small to large scale units could readily be made on the basis of cross sectional areas alone. The small system contains in addition a flash chamber inserted the preheater and actifier to remove part of the CO₂ before the actifier. The flashed gas, together with the gas from the top of the actifier is condensed, and the condensate returned to the system at the flash chamber.

2. Operation with Triethanolamine (T.E.A.)

As first constructed, the scrubbing tower was packed with 5/4 in. Raschig rings to a height of 35 ft., the tower being 12 in. in diameter. The actifier (same diameter) was packed with 2 in. Raschig rings to a height of 26 ft. below the feed inlet, and 6 ft. above this point.

The first scrubbing operations were made with a 50% T.E.A. solution in water. Raw gas containing 18% CO₂ (s.S.O.365) was passed through the tower in increasing amounts, using the minimum quantity of liquor to completely remove the CO₂. At a throughput of 6000 CF/H. raw gas and a liquor rate of 370 GPH (35% T.E.A.) the tower flooded and liquor appeared in the gas trap. However, the CO₂ was completely removed. This throughput, translated into terms of one of the large 6 ft. I.D. towers is equivalent to 4 000 000 CF scrubbed H₂/day.

A short run was next made circulating pure water through the system to determine whether the carry-over was due to liquid viscosity or entrainment. At 13 000 CF. raw gas and as high as 700 GPH water no carry-over was experienced. It was then concluded that the T.E.A. solution viscosity was too great for the 5/4 inch rings, and that 2 inch rings would probably be large enough to handle the equivalent throughput of 9 000 000 CF H₂/D on the large scrubber for which the small system was designed.

After changing over to 2 in. rings in the scrubber, the runs were continued, increasing the gas as rapidly as possible to reach the maximum tower capacity. In table I are given the results of these runs. The maximum gas rate obtained was 12 000 CF/Hr., equivalent to 8 000 000 CF H₂/D on one large unit. It was not possible to circulate enough solution (44% TEA) to completely remove the CO₂ (because of tower flooding at increased circulation rates) but the final gas contained only 0.8% CO₂.

The effect of scrubbing temperature is shown/a comparison of runs 5 and 5 A. There were made under similar conditions except for scrubbing temperature.

Raw Gas CF/H	Liquor GPM	% T.E.A. Conc.	Scrubber Inlet	Temp. E.G.C.	% Exit CO ₂
11 000	895	46.4	149	157	0.0
11 200	895	47.0	165	172	1.2

Exit temperatures exceeding 165°F. cause the appearance of CO₂ in the exit gas unless, of course, more scrubbing liquor is used to compensate for the decreased absorption per gallon.

Flash drum operation was irregular throughout the test work. The temperature at which flashing took place varied from 192 to 207° F., but it was difficult to determine the extent to which the flashing was affected. From the inspections made, roughly half the dissolved CO₂ was released in the chamber. Operation of this apparatus would be much better if the chamber were larger in cross section to allow more disengaging area. The present apparatus is 12 in. in diameter and does not allow the gas sufficient time or area to disengage.

The actifier, operating at about 225°F. kettle stripped the liquor down to approximately 0.5 CF % CO₂/gallon solution. The top temperature was held at 180 ~ 190°F. This system operated satisfactorily, but at high ratings there was a tendency for it to boil over unless the preheat temperature was kept high. With a high kettle steam requirement at high ratings, the capacity of the packing is exceeded and the cover boils over. By putting more load on the preheater and flash chamber this tendency was overcome.

The total steam consumption in the preheater and actifier (when running a 200°F. preheat) was 50 lbs. per thousand CF raw gas. In actual plant practice, where use is made of heat exchange this figure amounts to 40 lbs./MCF gas.

3. Operation with Diaminopropane.

This substance absorbs CO₂ mole for mole, whereas TEA absorbs only half as much. With its high heat of absorption (900 B.t.u./#/CO₂) it was considered inadvisable to use a higher solution concentration than 30% since the heat effect in the scrubber would be too great.

The first runs (Table III) were made at an equivalent rate of 6,000,000 CF H₂/day. In all of these the CO₂ was completely removed, but the liquor rate required varied considerably due to the large variation in stripping efficiency. The following tabulation shows the effect of residual CO₂ in the stripped liquor upon the liquor rate, all for the same gas rate, equivalent to 6,000,000 CF H₂/day.

Run No.	Liquor GPM	D.A.P. Conc.	CF CO ₂ /Gal. Steam Used		Remarks
			Stripped #/MCF.	Raw Gas	
2	860	51.1	4.36	—	Residual CO ₂ appeared.
3	425	29.1	4.05	59	
3	425	24.7	2.90	74.5	
34	540	23.9	2.90	103	

o/o

The high steam consumption in run 434 was discussed in trying to lower the CO₂ content below 2,9 CF/gal.

Further runs were made at equivalent gas rates of 12 to 16 mill. CF H₂/day (Table III). At these rates the scrubber operated satisfactorily with no entrainment nor flooding difficulties. The exit CO₂ at equivalent rates as high as 13 000 000 CF H₂/day was 0,0%, and amounted at times to as much as 0,6% depending on the stripping efficiency.

At these rates it became apparent that the actifier was operating very close to the flood point so that slight variations in kettle steam resulted in flooding. This is best shown by two runs noted below:

Ras Gas CFH	D.A.P. CFH	Preheat Temp. °F	Kettle Temp. °F	Actifizer Steam CFH	% D.A.P. in Condensate.
18 650	525	202	225	16 960	0,167
19 650	525	190	232	17 000	19,8%

By increasing the kettle steam only 1000 CF/H. the D.A.P. content in the condensate from the vent gases rose from 0,167% to 19,8%, indicating boiling over.

At 16 000 000 CF H₂ ratings the exit gas contained about 3% CO₂ due to the fact that the liquor rate was not increased because of lack of stripping capacity. The runs indicated (Tables III and IV) however, that the scrubbing tower was physically capable of handling the materials successfully without flooding.

Subsequent investigations (Table IV) were made on the actifier, noting the effect of varying preheat temperatures. These are summarized below:

Equiv. Gas Rates MM CF H ₂ /D.	Equiv. Liquor GPM	Preheat Temp. °F	% of absorbed CO ₂ Flashed.
15,5	515	220	27
14,6	530	222	73
16,1	515	215	42
14,6	530	215	36,5
16,2	515	216	26,6

These results indicate about 26% flashing of the absorbed CO₂ at 216°F 35% at 215 and over 50% at 220°F. The data at 220°F are not in agreement, but it is felt that the average of the two results is not far wrong. It will be noted, however, from Table IV that the steam consumption was greatly increased by the use of flashing.

In the runs made from November 20 to 23, the stripping steam was decreased to note the effect of increased CO₂ in the liquor upon scrubbing efficiency. Although the liquor analyses were inconsistent the results were nevertheless interesting.

Liquor Rate GPH	% Conc.	Gas Rate CFH	Actifizer Steam CFH	Stripped CO ₂ Liquor CF CO ₂ /Gal.	% CO ₂ in Scrubbed Gas.
525	28,1	20 000	16 068	5,27	0,0
525	26,6	20 000	15 936	2,73	1,0
600	29,1	20 000	14 000	5,96	0,0
600	53,0	19 800	12 000	6,3	1,4
525	20,6	20 000	14 000	2,28	1,0

At 525 GPM, lowering the kettle steam from 16 occ to 14,000 CF/H resulted in a rise of exit CO₂ from 0.6 to 1.0%. The stripped liquors analyzed, however, showed a drop in CO₂ from 5.27 (28.1% sol'n) to 2.73 (26.6% Sol'n). Raising the circulation rate to 600 GPM and keeping the kettle steam constant brought the exit CO₂ back to 0.6%, with a stripped liquor content of 6.6 CF CO₂/gal. These data indicate that fairly high contents of CO₂ in the circulating liquor may be used while completely scrubbing the raw gas of CO₂.

Dams were made at 173 - 180 °F. scrubber outlet temperature (estimated for plant scale conditions using D.A.P.) to determine whether such a high temperature would require much of an increase in circulating rate to completely scrub the gas. At 173 °F. scrubber inlet, 525 GPM circulation and 19400 CF/H raw Gas, the residual CO₂ varied from 2 to 4%. Increasing the circulation to 600 GPM (and at a scrubber outlet of 180°F) the exit CO₂ was 1.4%. Since operation of the plant scrubber with 30% solution would result in temperatures above 180°F. when using theoretical liquor rates with a resulting increase in exit CO₂ it will be necessary to use greater circulation to reduce the temperature rise to about 165 °F.

4. Extrapolation to Plant Scale Equipment.

For operation at the Baton Rouge Hydro Plant up to 12,500 CF/H per day of pure hydrogen, the gas could be scrubbed free of CO₂ in the present plant system (two scrubbers and two actifiers) with TMA with no spare apparatus. It is realized that the experimental data were obtained in packed towers whereas the plant towers are equipped with bubble plates, but even these will be at capacity when using TMA.

With D.A.P. solution, all the gas can readily be handled with one scrubber, leaving one as a spare. For actifying this solution (about 400 GPM) both the present actifiers will be necessary, requiring the addition of another one as a spare, of 6-1/2 to 9 Ft. diameter. In addition, the use of a flash drum is indicated, since the actifiers will be operating fairly close to the hang-up point (for packing or entrainment point for bubble plates) and hence a flash chamber where some gas can separate and decrease the tendency of the liquid to form light droplets will be advantageous.

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AGS.

EXPERIMENTAL CO₂ SCRUBBER OPERATION ON RAW HYDROGEN GAS USING TRICHLOROACETIC AS SCRUBBING MEDIUM.

Date	Run No.	Experim. Scrubber		Temperatures		CO ₂ P. Gal. in Liquor		Steam		CO ₂ %	
		Scrubbed E ₂ - C ₂ H ₅	% Cone.	Raw Gas	T. E. A. °C/H	Cooler Cone.	Outlet	After Stripping	Pre-heater	On top	Scrubber
8/26/34	4	4 000 000	340	45.1	6 000	400	45.1	144	199	221	0,000
8/27/34	3	4 000 000	222	35.0	6 000	370	35.0	142	252	227	0,000
9/6/34	1	4 000 000	250	19.7	6 000	400	19.7	154	204	226	3,1
9/6/34	1A	4 000 000	240	40.1	6 000	400	40.1	155	204	225	0,000
9/7/34	2	5 180 000	265	42.0	7 800	440	42.0	142	152	216	0,446
9/8/34	2A	5 180 000	295	37.4	7 800	491	37.4	147	155	207	1,23
9/9/34	3	6 060 000	500	41.5	9 100	500	41.5	140	142	202	2,22
9/10/34	4	5 260 000	300	43.4	9 400	500	43.4	152	159	201	2,55
9/10/34	4A	6 260 000	556	47.7	9 400	560	47.7	147	150	192	226
9/11/34	5	7 300 000	357	49.4	11 000	595	46.4	149	157	196	225
9/11/34	5A	7 450 000	557	47.0	11 200	595	47.0	165	172	200	229
9/22/34	6	8 000 000	396	44.0	12 000	650	44.0	153	157	202	15 000
											227
											0,3
											15 000
											227
											0,3
											15 000
											227

Scrubber filled with 3/4" Rasching rings.
Liquor came over.

Filled scrubber with 2" Rasching rings.
Concentr. probably too low.

TABLE XI

EXPERIMENTAL CO₂ SCRUBBER OPERATION ON RAW HYDROGEN GAS USING DIAMINOPROPANOL AS SCRUBBING MEDIUM

TOWER PACKED WITH 1/4 INCH RASCHIG RINGS.

Date	Run No.	Equivalent Rates on Plant Scrubber:	Rates on Export Scrubber				Temperatures				CF CO ₂ /Cell. Liquid:				% DAP in Steam GP/H				Exit CO ₂ %		
			D. A. P.		D. A. P.		Raw Gas		Cooler Outlet		Pre-tester Outlet		Act Pier Kettle		After Scrubber		Stripped Liquid		Total		
			Scrubbed E2 - GP/D.	GPM Conc.	Scrubbed E2 - GP/D.	GPM Conc.	CO/H	GPE Conc.	CO/H	GPE Conc.	CO/H	GPE Conc.	CO/H	GPE Conc.	CO/H	GPE Conc.	CO/H	GPE Conc.	CO/H	GPE Conc.	
9/14/34	1	8 000 000	219	25.0	12 000	664	25.0	86	150	201	224	2.50	-	0.0	0.0	0.0	0.0	0.0	0.0	0.0	
9/15/34	2	8 000 000	204	51.1	12 000	340	31.1	85	155	202	220	4.55	2.4	-	0.2	0.0	0.0	0.0	0.0	0.0	
9/16/34	3	8 000 000	255	29.4	12 000	425	29.4	5	151	200	220	4.05	-	15 000	0.0	0.0	0.0	0.0	0.0	0.0	
9/18/34	5	8 000 000	255	24.7	12 000	425	24.7	91	151	205	221	5.90	-	19 000	0.0	0.0	0.0	0.0	0.0	0.0	
9/18/34	3A	8 000 000	204	24.9	12 000	540	24.9	85	149	209	229	7.05	2.91	1,54	26 000	0.0	0.0	0.0	0.0	0.0	0.0

A. MUDRING, SCHÜTZING UND SCHÜTZING

Reactions on Experimental Plant Scrubber		Temperature						CP - CO ₂ / Gallon : % D.A.P.	
Date	Time	Pure CO ₂ GPM	GPM Conc.	Scrubber Gas CO ₂ /H ₂	% Conc.	Inlet Outlet	Actizier Pre-heat	Steam C.E./ Heat CO ₂	Scrubber Flash. Scripp. Mesh. Condor Exit Liquor. Liquor. Sett
11-7-34	6:00 AM	12	650 000	315	29.5	19 000	525	29.5	9.9 26.4 0.956
11-7-34	12:00	12	200 000	315	21.2	18 400	525	21.2	7.2 8.7 4.00
11-8-34	2:00 AM	15	500 000	315	62.0	20 000	525	62.0	6.2 27.7 9.8
11-8-34	6:00 PM	13	600 000	315	30.5	19 550	525	30.5	10.1 9.95 5.04
11-9-34	6:00 PM	12	850 000	315	30.5	19 500	525	30.5	10.1 9.95 5.04
11-10-34	6:00 PM	12	850 000	315	30.5	19 500	525	30.5	10.1 9.95 5.04
11-11-34	6:00 PM	12	440 000	315	31.8	18 650	525	31.8	10.2 38.9 15.1
11-11-34	5:00 PM	13	200 000	315	—	19 650	525	—	10.5 39.5 14.6
11-12-34	9:30 AM	15	300 000	315	26.7	19 800	525	26.7	10.6 41.1 15.7
11-12-34	9:30 AM	15	300 000	315	26.7	19 800	525	26.7	10.6 41.1 15.7
11-14-34	9:00 PM	16	600 000	315	27.0	24 800	525	27.0	10.8 42.8 15.2
11-14-34	9:00 PM	16	600 000	315	27.0	24 800	525	27.0	10.8 42.8 15.2

TABLE IV.

EXPERIMENTAL CO₂ SCOURING UNIT

