

Mr Wiley

U. S. BUREAU OF MINES T-370
HYDRO. DEMON. PLANT DIV. 959

KCBraum
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Aromatics - Naphthene Equilibrium
By Dr. Peters, Ludwigshafen, 22 May 1943

Some years ago Dr. V. Miffling calculated the equilibrium for benzol-cyclo-hexane and toluol-methylcyclohexane, shown in Fig. 1.

These data are graphically represented in Fig. 2 and Fig. 3 in another form.

- a) Fig. 2 shows the dependence of the aromatics concentration on H₂ partial pressure in different temperature ranges.
- b) Fig. 3 shows the pressures and temperatures required in order to obtain the thermo-dynamic equilibrium of certain aromatic concentrations.

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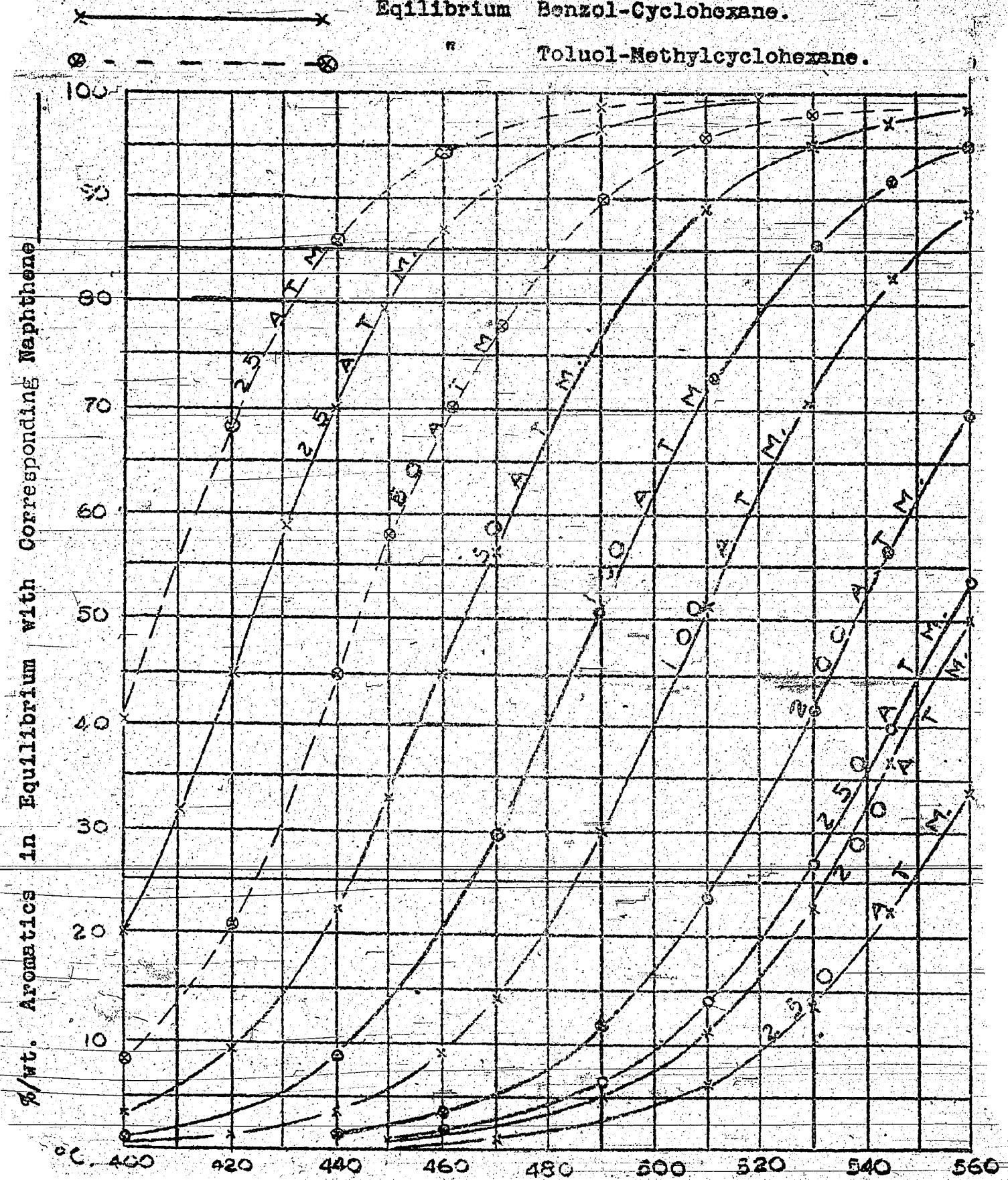
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Fig.

Aromatics-Naphthalene Equilibrium For Different H_2 Partial Pressures

Eqilibrium Benzol-Cyclohexane.

Toluol-Methylcyclohexane.



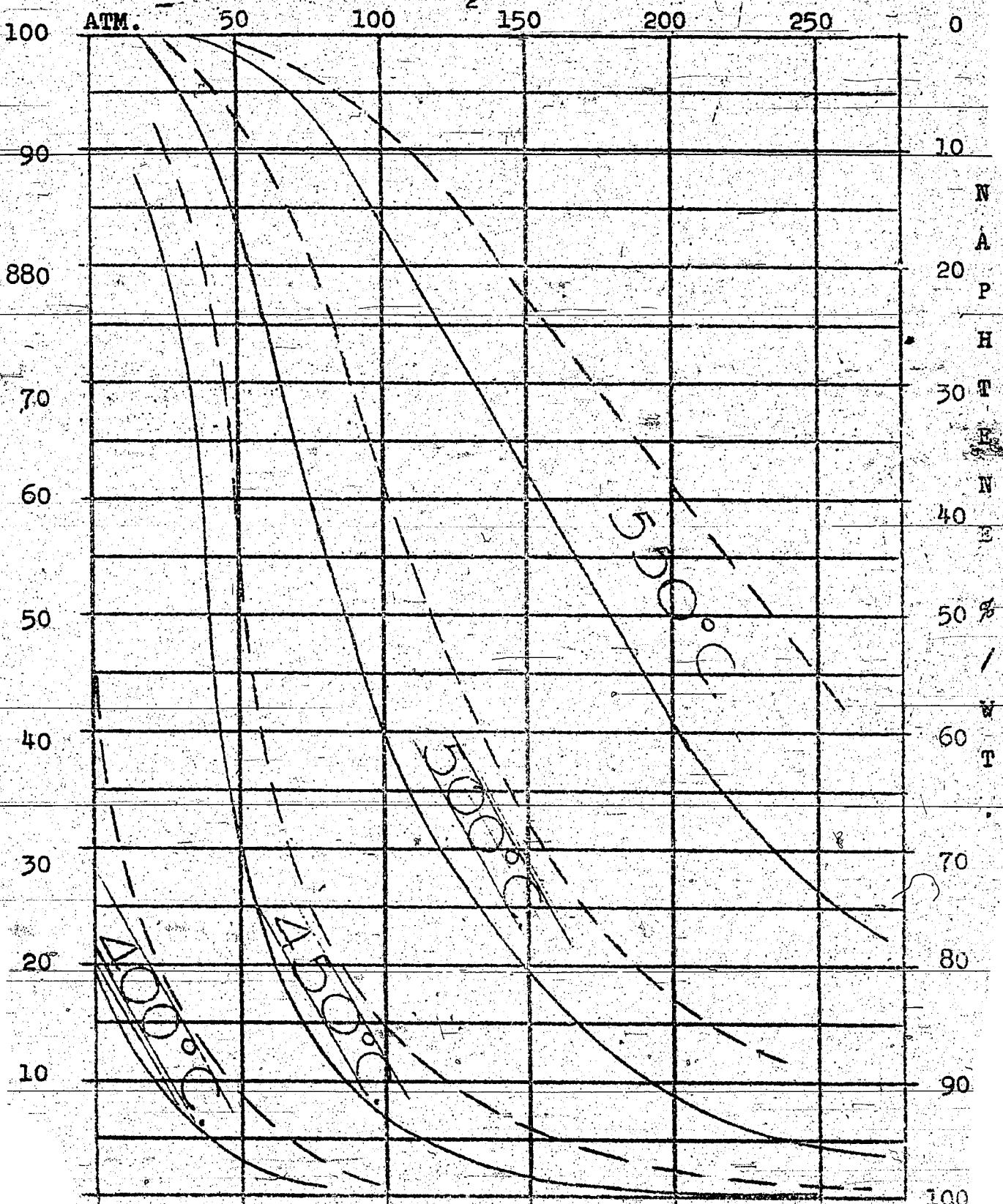
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Fig. 2.

AROMATIC-NAPHTENE-EQUILIBRIUM AT DIFFERENT TEMPERATURES.

BENZOL-CYCLOHEXANE
TOLUOL-METHYLCYCLOHEXAN

 H_2 PARTIAL PRESSURES

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FIG. 3

CONDITIONS OF EQUILIBRIUM FOR
AROMATIC CONCENTRATIONS

ATM.

250

200

150

100

50

25

400 420 440 460 480 500 520 540 560

BENZOL - CYCLOHEXANE

— — — — TOLUOL - METHYL CYCLOHEXANE

