V. D.H.D. (DEHYDRIERUNG-HOCH-DRUCK) PROCESS.

(i) Description of Process and Leuna operating results.

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The DHD process was developed in Germany by the I.G. Farbenindustie A.G. for the catalytic dehydrogenation of naphthenes to arcmatics and was intended primarily for the use on hydrogasolines. The catalyst employed is 10% molybdic oxide on alumina and the reaction is carried out with recycle hydrogen under a pressure of 30-40 ats at a temperature slightly above 500°C.

Two DHD units had been completed at Leuna and one of them had operated for 3 months. Two further units were in course of erection.

The method of operating the process is shewn in the simplified flow diagrams in FigsXII. and XIII.

Full boiling range (IBP-185°C) hydrogasoline is first fractionated to remove, as an overhead cut, the lower boiling material boiling up to 85°C. The bottoms from this prefractionator are mixed with recycle gas containing 55-65% hydrogen. 1 cu.m. of recycle gas is used per kg. of liquid feed.

The mixture of gasoline and recycle gas is first preheated by heat exchange with the product from the end reactor; then by heat exchange with the material from the fourth reactor, and finally by convection heating to a four-chamber convection heater. The pressure at the inlet of the first exchanger is 42 ats, and at the inlet of the first reactor it is 37 ats.

The feed enters the first reactor at 500°C and passes downward through the catalyst bed and leaves at 450°C. The mixture is then reheated in one section of the heater to 510°C. and enters the second reactor under 34 ats pressure. The material leaves the bottom of the second reactor at 490°C, is reheated in a third section of the heater and enters the third reactor at a temperature of 520°C and under a pressure of The outlet from this reactor at 510°C is again reheated in the fourth section of the heater and enters the fourth reactor at 530°C under a pressure of 30 ats. Each section of the preheater has its own burner and independent fuel gas, air and recycle gas supply. At the outlet of the fourth reactor, the dehydrogenation reaction is completed. The effluent from this reactor at 530°C. is cooled to 300°C by heat exchange with the fresh feed and passes to the fifth reactor for hydrogenation of the olefins. The effluent from this fifth reactor exchanges heat with the fresh feed and passes through a cooler to a

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receiver for the separation of the gas from the liquid. The gas is principally methane and hydrogen with some ethane and very little C_2 - C_1 . At the beginning of the cycle, the gas contains about 65% hydrogen, and at the end of the cycle about 55% hydrogen. The density of the recycle gas at 15°0 and 735 mm. is 0.38-0.45 kg per cubic metre. A part of this gas is recycled with the from feed, as explained above, and the remainder removed from the system as tail gas.

The liquid product from the receiver is let down, fractionated to remove the heavy materials and then sent to a stabilizer along with the LEP-85°C out from the initial charge. In the stabilizer, the athene to butane product is taken overhead and the DED gasoline is made as bottoms.

The total hydronarbon gas made is purged from the circulating gas and evolved on let down of the liquid product and on its subsequent stabilization amounts to 19-20% by weight of the original hydropetrol, i.e. about 25% by weight of the naphtha fed to the process. The gas is comprised of roughly equal proportions by weight of methane, ethane, propane and butane.

The catalyst used in all five reactors is active alumina on which has been deposited 8-10% MoCz. It is prepared by the impregnation of activated alumina with ammonium molybdate solution. The catalyst is used in the form of 8-12 mm, cubes and is placed in the reactors in a single bed. It was originally planned to operate the DED unit with a liquid space velocity of 0.38 kgs of liquid per litre of catalyst. In actual operation, space velocities of 0.25-0.52 were the highest obtainable. The length of the on-stress cycle on the catalyst depends upon the charge stock and varies from 120-240 hours. The regeneration period with the necessary parging operation requires about 24 hours.

At the end of the on-stream cycle, the reactors are first depressured and filled with nitrogen to 10 atmospheres pressure, depressured again and filled with nitrogen to 70 ata pressure. The recycle compressor is then started and air is admitted to the recycle gas surems until the cargen content at the inlet of each reactor reaches 1%, 20,000 cu.m. of gas are recycled per hour through each reactor. The normal time of regeneration is about 12 hours. Oxygen concentration in the recycle gas is controlled to hold the regeneration temperature to slightly less than 540°C. During regeneration, 1 cu.m. of water is added per hour, either at the second heat exchanger or at the cooler, for removal of acidic compounds produced during regeneration. In some cases, it has been necessary to use dilute caustic at this point. It is necessary to remove 802 sulphur dioxide completely from the recycle gas, but the cerbon dioxide concentration can go up to 8-13% without harmful effects. At the

TABLE Y.

LEUNA D.H.D. PLANT

PROPERTIES OF FEED AND PRODUCTS.

	0-85°C Light Petrol	85°-180°C Naphtha feed to D.H.D. Process.	Stabilised D.H.D. Product.	Final DHD Petrol Blend
Aniline Point	52	45	1.6	6.9
Aniline Point after treatment with sulphuric acid.	55•4	<u>-</u>	63.2	61.0.
Wt.% aromatics	4.5	9•5	66.0	52.0
Wt.% Naphthenes	46.2	51.2	7.9	14.4
Wt.% Paraffins	48.7	38.1	25.9	33.0
Wt.% Olefines	0.6	1,2	0.6	0.6
Motor Method Octane No.	73•5	52.0	82.5	80.5
Motor Method Octane No. with O.12% by vol.TEL.	-	-	-	91.5

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TABLE VI

CALCULATED COSTS FOR D.H.D. PETROL

(Based on Works Costs for 1st Qr., 1944)

i	Unit	Cost	Per Hour		Per The II-D Per	
····		Rei/Unit	Quantit		Quantity	
A. Materials		}		1		<u> </u>
1) Petrol	Torns	170.141	13.35	2,271.37	1.379	234.
2) Catalysts & Chemicals.			,	20.01		
7\ 7\ 7\ - 1				29.04		3.
Circ.Gas Purge	Cal.	7.00	14.2	99 -4 0	1.467	1 <u>0</u> ,
Rich Ges	#	11.352)	20.5	232,68	2.118	•
Redistilln Resilv	Found	100.00	0.47	47.00	i	24. 4
B. Running Costs (3) (See Table VI)					·	
Production Cost			-2 (3	594.26	<u> </u>	61,
	_ 1			2,515.59	1,000	259.
C. Loading & Evaporation	Turne	3.00	9.68	29,04	1,000	3,
D. Oncosts				262.62	i	27.
Works Coat (without conversion	- 1	·	9.68	2,807.25	1,000	250.
and oil taxes and	ì	ſ	i	1	- 1	
other special costs)			!	ŀ		
1	- 1	- 1	}	j	1	

- (1) This cost has been taken from Table I.
- (2) Butane and Propage in Rich Gas valued at Motor Spirit valuation less cost of separation.
- (3) It is estimated that for 4 DHD Stalls, the running costs would be 42 RM/The DHD Petrol and the Works cost, 270 RM/The DFD Petrol.

TABLE VII
BUILD-UP OF HUNNING COSTS FOR DHD PETROL.

		Cost per hour		Per Tne DHD Petrol.		
	Quantity		Quantity	RM		
Labour Costs Wages h Salaries Social Insurance	21.76	35.94 26.56 7.07 2.31	2,25	3.71 2.74 0.73 0.24		
Energy Costs Water M ³ H.P.Steam Tne I.P.Steam Tne L.P.Steam Tne L.P.Steam KWF L.T.Electricity KWF Fuel Gas 10 ³ Tne.Cs Other forms.	9.76 5.63 1. 1,881.31	67.44 20.10 13.41	83.20 2.21 1.01 0.58 194.35 27.64 667.46	16.91 1.38 6.97 2.08 1.39 2.73 0.54 6.01 0.03		
Repair Costs Wages hrs Material Workshop and Material on cost.	13.19	49.17 15.62 6.98 26.57	1.36	5.08 1.61 0.72 2.75		
Working Materials		1.71		0.18		
Traffic Charges		0.25		0.03		
Works General Charges		20.04		2.07		
Capital Charges Writing off Obsolescence		297.29 192.78 104.51		30.70 19.90 10.80		
Taxes		20.88		2.16		
Various Costs		6.01		0.62		
Credits		<u>0.68</u>		0.07		
TOTAL:		594.26		61.39		

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end of the regeneration syule, the inert gas is stored in cylinders for re-use on the next regeneration. The catalyst chambers are depressured and then filled with the regular recycle gas from storage for the subsequent on-stream cycle.

The reactors are the usual high pressure reaction vessels having a flanged top head. The inside diameter of the reactor is 1520 mm. The first reactor is 11 metres in length and the remaining four reactors are 13 metres in length. The reactors are lined with firebrick in order that the outer reactor wall temperature does not exceed 200°C. The exchangers are conventional tube and shell units with the fresh feed in the shell in all cases. The convection heater employs flue gas recirculation and all heating is done by convection heat rather than by radiant heat. The fractionating system is the conventional bubble plate tower type.

The finished debutarized DHD gasoline, which includes front ends from the gasoline charge, contains 50-52% by weight of arcmatics. The octane number is 80.5 CFR motor method. It appears that some paraffins are converted to aromatics in the process. Other properties of D.H.D. plant food materials and products are given in Table Y.

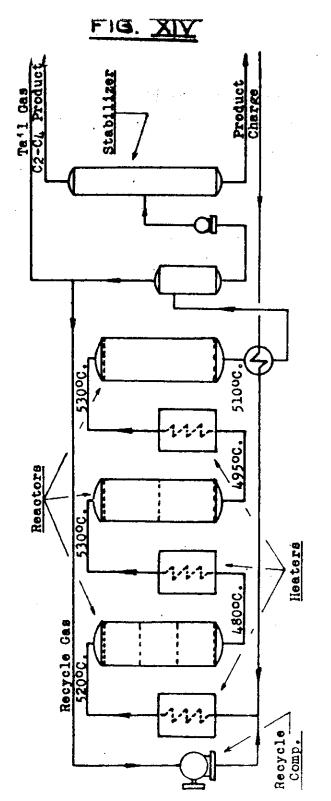
(11) Costs.

Dr. Pichler provided the following cost data for the DHD process.

The basic flowsheet assumed for the cost calculation is shewn in Fig. XII .

Table VI analyses the cost of manufacture of DHD gasoline in terms of raw material charges and operating costs and shows the credit for hydrocarbon gas.

Table VII gives a breakdown of the operating coats and provides the data on labour requirements, utilities consumption, etc, for calculation of the costs of the process if operated in U.S.A. or Britain.



FLOW DIAGRAM - HYDROFORMING PROCESS