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(54) METHOD OF MAKING A SUPERIOR JET FUEL

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1	The present invention	relates to a method of	
2	making a superior jet fuel for s	supersonic aircraft from a	
3	paraffinic, virgin hydrocarbon stream.		
4	Specifications for sup	personic aircraft fuel may	
5	be set by aircraft and jet engin		
6	governmental branches. These sp		
. 7	since the designs of alreraft engines and air frames are		
8	based upon the specific performance characteristics which		
9	have been set with respect to the fuel specification. For		
10	a particular example of superior jet fuel for use at Mach-3		
11	speeds, the following specifications have been set.		
12	TABLE I		
13	Specifications for Mach-3 Jet Fuel		
14	Luminometer Number	100 Minimum	
15	Freeze Point	-40°F. Maximum	
16	Viscosity at -30°F.	15 cs. Maximum	
17 18	Heat of Combustion	18,900 Btu/lb., net, Min.	
19	20% Distilled	408°F. Minimum	
51 50	Thermal Stability at 300/500/600 (Research Coker)	Must Pass	
55	Many difficulties prese	nt themselves when a virgin	
23	petroleum hydrocarbon stream such	•	
24	condensate is proposed as the bas	-	
25	these specifications. Specificat		
26	number, freeze point, and heat of combustion in particular		
27	are extremely hard to satisfy when the source of the fuel is		
28	such a virgin hydrocarbon stream. Various attempts to		
29	manufacture fuels meeting the above specifications by		
30	extraction of virgin distillates or by blending various		
31	extracted kerosenes have been unsuccessful. However, by the		
32	practice of the present invention	•	

processing steps is used and wherein the Engler distillation 1 20% point of the jet fuel product is maintained within a critical range, a Mach-3 jet fuel can be produced which 3 4 meets the desired rigorous specifications when starting from a virgin crude oil or from a heavy, natural condensate. 5 б The present invention involves the processing sequence of a first distillation to obtain a charge stock 7 8 boiling within the range of 355°F. to 510°F., followed by 9 desulfurization, reforming, extraction, hydrofining, and 10 adsorption, with the associated distillation steps which are 11 required in order to obtain a final product boiling within the range of 375°F. to 495°F. and having an Engler 12 distillation 20% point of 408°F. minimum and preferably 13 14 within the range of about 410°F, to 414°F. 15 Referring now to the drawing, a preferred mode of 16 the present invention is disclosed as comprising a first step 17 of fractionating a paraffinic feedstock such as crude oil or 18 heavy condensates (obtained from distillate wells and the like) introduced into fractionating tower 101 by way of 19 20 line 102. A low-boiling overhead stream is withdrawn by way 21 of line 103, and a high-boiling stream is removed by way of line 104, with a heart cut fraction boiling within the range 22 23 of 355°F. to 510°F. being obtained by way of line 105. The 24 heart cut is the charge stream to the processing sequence of 25 the present invention. 26 The charge stream is admixed with hydrogen supplied 27 by way of line 106 and is contacted with a hydrodesulfurization 28 catalyst in the hydrodesulfurization zone 108. A 29 hydrodesulfurized product is withdrawn by way of line 110 30 (and may be distilled in a fractionator to obtain a dry point 31 of the overhead product of 490°F. to 500°F.), admixed with 32 hydrogen supplied by way of line 112, and subjected to

catalytic reforming in the catalytic reforming zone 114. 1 2 A reformed product stream is removed by way of line 116, preferably through a stabilizer tower (not shown), 3 4 where the light ends are removed to give an IBP (initial boiling point) minimum of 300°F., and is then contacted 5 6 preferably with sulfur dioxide as a solvent for aromatic hydrocarbon in the zone 118. In this step the SO2 is 7 8 introduced by way of line 120 and an aromatic extract is removed by way of line 122, the paraffinic raffinate being 9 10 withdrawn by way of line 124. The raffinate is then fractionated in order to obtain an overhead fraction boiling 11 below about 500°F., for example having an Engler FBP (final 12 boiling point) of 480°F. to 490°F. In this distillation step 13 14 the heavy polymers which are produced in the reformer 114 15 and/or the hydrodesulfurizer 108 are removed in the high-16 boiling material. For this purpose the raffinate is 17 introduced into distillation column 130, and a high-boiling 18 bottoms stream, withdrawn by way of line 132 while the 19 overhead stream, having an FBP of about 480°F. to 490°F., is 20 withdrawn by way of line 134. 21 The overhead stream 134 is admixed with hydrogen supplied by line 136 and introduced into a hydrofining zone 22 23 138, wherein the distilled raffinate is hydrofined. The 24 hydrofined material may be fractionated to adjust the initial boiling point to about 380°F. and is then introduced into 25 26 adsorption zone 142 as a final finishing step. In the 27 adsorption zone 142, the residual aromatic compounds are 28 removed and thermal stability of the fuel improved. The Mach-3 fuel is withdrawn by way of line 144 and may be caustic 29 30 washed before or after adsorption if desired. The various 31 distillation steps after the reforming operation must produce 32 a final product having an Engler 20% point within the range of

- 1 410°F. to 414°F. Obviously, this may be accomplished by
- 2 distillation at any of many points in the processing sequence,
- 3 but preferably the 20% point is adjusted in the final
- 4 distillation zone 130 or in a distillation zone (not shown)
- 5 after hydrofining in zone 138. By raising the Engler 20%
- 6 point to 408°F. minimum, preferably 410°F. to 414°F., the
- 7 luminometer number and heat of combustion are maximized while
- 8 the freeze point is minimized, contrary to what would be
- 9 expected. It normally would be expected that these values
- 10 would degrade with an increasing 20% point. However, it has
- 11 been found that with the 20% point within the prescribed 4°F.
- 12 range, the luminometer number, freeze point, and Btu ratings
- 13 of the fuel are at their most desirable values, less desirable
- 14 values for each being obtained at a 20% point either higher or
- 15 lower than the critical range.
- 16 This is seen from the following Table II, wherein
- 17 the changes of the luminometer number, the freeze point, and
- 18 the heat of combustion are shown.

19 TABLE II

20		402°F.	410°F.
21	Luminometer Number	123.0	126.9
22	Freeze Point	-42°F.	-43°F.
23	Heat of Combustion, Btu/lb.	18,936	18,948

- Thus, it is seen that in order to make the rigorous
- 25 specifications, the Engler 20% point of the product must be
- 26 maintained within the narrowly prescribed range.
- 27 In general, the processing scheme has been shown to
- 28 include a prefractionation step followed by hydrodesulfurization,
- 29 catalytic reforming, extraction, fractionation, hydrofining,
- 30 and adsorption in order to produce a superior quality Mach-3
- 31 jet fuel. The individual steps, including the variables
- 32 involved therein, are discussed in sequence below.

1	Feedstock
2	The oil to be used as an ultimate feedstock is
3	obtained from a paraffinic crude such as Panhandle crude or
4	from field condensates. The preferred crude cil will contain
5	at least about 40% paraffins and less than about 60%
6	aromatics plus naphthenes.
7	Distillation
8	The selected crude oil is then distilled in order to
9	obtain a distillate having a boiling range from about 355°F.
10	to about 510°F. containing from about 30 to about 50% normal
11	paraffins, from about 10 to about 40% isoparaffins, from about
12	40 to about 60% aromatics plus naphthenes, and from about 0
13	to about 5% olefins. The fractionation step removes high
14	boiling, unsuitable materials from the fuel.
15	Desulfurization
16	The distilled fraction is submitted to
17	hydrodesulfurization, wherein the fraction is contacted with a
18	hydrodesulfurizing catalyst such as extruded or pilled cobalt
19	molybdate. The temperature in the hydrodesulfurization zone
20	suitably may range from about 600°F. to about 690°F., the
21	pressure from 150 psig to 300 psig, and a hydrogen feed rate
22	from 200 scf/bbl. to 600 scf/bbl. of feed, based on pure
23	hydrogen. The hydrogen recycle stream containing from about
24	60% to about 95% hydrogen may be used, in which case the
25	amount of hydrogen recycle stream per barrel of feed will be
26	larger than would be the case if pure hydrogen were being used.
27	The charge to the hydrodesulfurizing zone shall
28	suitably involve a space velocity of from about 2.0 to about
29	4.0 volumes of hydrocarbon feed per volume of catalyst per hour.
30	In the hydrodesulfurizing step, the sulfur is removed
31	and the sulfur content of the material is reduced from about

0.10 to 0.25% in the feed to about 0.002 to about 0.00% sulfur

- 1 in the effluent from the hydrodesulfurization unit. The HoS
- 2 and low-boiling materials produced by hydrodesulfurization
- 3 may be removed by steam stripping in order to provide a more
- 4 suitable feedstock for the following reforming step.

5 Catalytic Reforming

- 6 The hydrodesulfurized stream is then charged into
- 7 the reforming zone and contacted with hydrogen in the presence
- 8 of a reforming catalyst such as platinum on alumina, which has
- 9 both aromatizing and isomerizing activity. The temperature
- 10 may range from about 825°F. to about 900°F.; the pressure from
- 11 150 psig to 450 psig; the amount of hydrogen to hydrocarbon
- 12 feed from 2000 scf per barrel to 7000 scf per barrel (based
- on pure hydrogen); and the space velocity may be from 0.2 to
- 14 2.0 V/V/Hr.
- In the reforming zone, the naphthenic hydrocarbons
- 16 are preferentially converted into aromatic hydrocarbons, which
- 17 makes them susceptible to extraction. It is essential to
- 18 operate the reformer-at a minimum temperature of 825°F. to
- 19 permit isomerization of normal to isoparaffins. While the
- 20 ratio of iso- to normal paraffins in the feed may be as low
- 21 as 0.2 to 0.5, the ratio in the finished product must be in
- 22 the order of 2.0. The preceding hydrodesulfurizing step
- 23 prepares the material for reforming by removing sulfur, which
- 24 would interfere with the activity of the catalyst in this
- 25 preferential isomerization-aromatization reaction.

26 Extraction

- 27 The reformed stream is then passed into preferably
- 28 an SO₂ extraction zone wherein the extraction is carried out.
- 29 For SO2, the extraction must be carried out under critical
- 30 conditions in order to minimize the amount of aromatics
- 31 remaining in the raffinate stream. The maximum temperature
- 32 in the extraction tower may range from about 50°F. to about

- 1 100°F., but is preferably maintained at about 65°F. A
- 2 temperature gradient in the extraction tower of from about
- 3 40°F. to about 100°F. may be used, while a solvent-to-
- 4 hydrocarbon ratio within the range from about 9:10 to about
- 5 2:1 volumes of solvent per volume of hydrogen may be used.
- 6 By practicing the extraction step with sulfur dioxide under
- 7 these conditions, a raffinate may be obtained which contains
- 8 less than 3% aromatic hydrocarbon. Ammonia and phenol may be
- 9 used instead of SO₂ extraction, but the same general
- 10 temperature limits will apply.

ll Distillation

- 12 The raffinate is then submitted to after-
- 13 fractionation in order to remove polymeric high-boiling
- 14 materials formed in the hydrodesulfurization and/or reforming
- 15 steps, that is, those hydrocarbons which boil above about
- 16 480°F. to 500°F. This high-boiling material is deleterious
- 17 because directionally it has an adverse effect on freeze
- 18 point, heat content and luminometer number. It is also
- 19 likely to reduce thermal stability.
- 20 The material boiling below 480°F. to 500°F. is then
- 21 removed as a product suitable for the finishing steps.

22 Hydrofining

- 23 The fractionated hydrocarbon is then passed to a
- 24 hydrofining unit where it is contacted with hydrogen and a
- 25 hydrofining catalyst such as extruded or pilled cobalt
- 26 molybdate. The hydrofining step reduces the olefinic content
- 27 and removes sulfur, nitrogen and polar compounds. This step
- 28 is carried out at a temperature within the range from about
- 29 575°F. to about 650°F., a pressure of 250 to 350 psig, a
- 30 liquid hydrocarbon space velocity of 1.0 to 4.0 volumes of
- 31 hydrocarbon per volume of catalyst per hour, and a hydrogen-
- 32 to-hydrocarbon feed ratio of 500 to 1500 scf/bbl. (based on

- 1 pure hydrogen). The hydrofined hydrocarbon is essentially
- 2 free of olefins and aromatics, but may still contain trace
- 3 amounts of each component.

4 Adsorption

- 5 The hydrofined fuel is then passed over an adsorbent
- 6 bed having an affinity for polar compounds, such as activated
- 7 charcoal, at a treat ratio of 0.5 to 4.0 V/V/Hr., whereby a
- 8 thermally stable jet fuel boiling within the range of about
- 9 400°F. to 500°F. is obtained, having a heat of combustion of
- 10 at least 18,900 Btu/lb., and otherwise meeting the
- 11 specifications above set forth. A preferred treat rate is 1.5
- 12 to 3.5 V/V/Hr. The adsorption step removes traces of polar
- 13 compounds from the fuel.
- 14 As an example of the present invention, a run
- 15 utilizing Panhandle crude was made, and the processing
- 16 sequence above set forth was generally followed. A crude
- 17 fraction boiling within the range of 355°F. to 510°F. was
- 18 submitted to hydrodesulfurization. The sulfur content was
- 19 reduced from 1400 ppm to about 50 ppm by contacting the stream
- 20 with 260 scf per barrel of hydrogen over a cobalt molybdate
- 21 catalyst at 658°F. and 255 psig.
- The hydrodesulfurized product was catalytically
- 23 reformed over a platinum-on-alumina catalyst at 825°F. and
- 24 270 psig in the presence of 3200 scf per barrel of hydrogen
- 25 and at a space velocity of 0.4 V/V/Hr.
- 26 The feed rate was 0.4 V/V/Hr. with the reactor
- 27 Inlet temperature 825°F. with an 85°F. temperature drop across
- 28 the reactor. The hydrogen rate was 3500 scf per barrel of
- 29 feed, with a pressure of 280 psig. Conversion of naphthenes
- 30 to aromatics was approximately 60% to 75% complete, with
- 31 liquid yield being over 95%. The ratio of iso- to normal
- 32 paraffins in the product exceeded 2.0.

1	The reformate was then charged to an SO2 extraction
2	unit at a feed rate of 3200 barrels per day, to provide an SO2
3	treat of 180%. The temperature at the bottom of the mixer
4	tower was 15°F. to 25°F., while that at the tower top was 65°F.
5	The aromatics content of the raffinate obtained under these
6	conditions was indicated to be essentially 0%. The accuracy
7	of the testing method indicated that less than 0.5% aromatics
8	remained.
9	The SO2 raffinate was then caustic washed with 0.02
10	pound of NaOH per barrel of raffinate, to 0.10 pound per
11	barrel. The caustic washed material was hydrofined over
12	cobalt molybdate catalyst at 615°F. to 625°F. and 300 psig
13	pressure in order to saturate the 0.5 to 1% of olefins which
14	remained in the fuel. The front end was adjusted by
15	fractionation at the hydrofining unit.
16	The hydrofined fuel was then caustic washed with
17	30° Baume NaOH, and passed over an activated charcoal bed at
18	about 3 V/V/Hr. in order to maximize thermal stability. The
19	product obtained is a fuel which boils within the range of
50	375°F. to 495°F. and has an Engler 20% point of about 410°F.
21	The fuel meets all specifications of the Mach-3 jet fuel.
22	An analysis of the product is set forth below in
22	Mahaa TIT

1.		TABLE III	
2 3	Gravity, °API Distillation, ASTM		52.5
234567890	IBP, °F. FPB		395 489
7 8	5% 10% 20%		404 406 411
9 10	30% 40%		416 421
11 12	50% 60% 70%		427 433 440
13 14 15 16	80% 90%		440 450 465 480
16 17 18	95% Luminometer Number		117
19 20	Freeze Point, °F. Heat content, Btu/lb. Aromatics, Vol. %	net	-46 18,919
22 21	Isoparaffins Normal paraffins		0.8 59.8 28.0
23	Naphthenes		11.4

It was thus seen that Applicants have provided a novel combination of processing steps which leads to the production of a highly useful supersonic jet fuel. The various steps must be accomplished in the order given, with the exception of the distillation step after extraction, which may be accomplished either before hydrofining or after hydrofining, but before adsorption over a polar adsorbent.

THE EMBODIMENTS OF THE INVENTION IN WHICH AN EXCLUSIVE PROPERTY OR PRIVILEGE IS CLAIMED ARE DEFINED AS FOLLOWS:

1. A process of making a jet fuel which comprises:
fractionating a virgin paraffinic hydrocarbon stream
to obtain a charge stream boiling within the range of 355°F. to
510°F.;

contacting said charge stream with hydrogen and a hydrodesulfurization catalyst under hydrodesulfurization conditions to obtain a desulfurized stream containing less than 0.01% sulfur by weight;

contacting said desulfurized stream with hydrogen and a reforming catalyst at a temperature from 825°F. to 900°F., a pressure from 150 psig to 450 psig, and a space velocity from 0.2 to 2.0 V/V/Hr., whereby naphthene aromatization and normal paraffin isomerization are maximized in obtaining a reformed stream;

extracting substantially all of the aromatic hydrocarbons from said reformed stream by contact with a solvent for aromatic hydrocarbons to obtain a raffinate containing less than 3% aromatic hydrocarbons;

hydrofining said raffinate by contact with hydrogen and a hydrofining catalyst under hydrofining conditions to obtain a hydrofined product; and

contacting said hydrofined product with an adsorbent for polar compounds to obtain a thermally stable jet fuel;

said jet fuel being subjected to at least one fractionation step during the course of said process whereby the Engler distillation 20% point thereof is 408°F. minimum, and the fuel boils within the range of 375°F. to 495°F.

2. A process in accordance with claim 1 wherein the hydrodesulfurization catalyst is cobalt molybdate, and said hydrodesulfurization conditions comprise a temperature within the range of 600°F. to 690°F., a pressure within the range from 12 150 psig to 300 psig, and a space velocity of 2 to 4 V/V/Hr.

- 3. A process in accordance with claim 1 wherein said reforming catalyst is platinum on alumina.
- 4 . A process in accordance with claim 1 wherein the solvent for aromatic hydrocarbons is chosen from the group consisting of SO_2 , NH_3 , and phenol.
- 5. A process in accordance with claim 1 wherein the solvent for aromatic hydrocarbons is SO₂, and the extraction is accomplished at a solvent-to-hydrocarbon ratio from 9:10 to 2:1 in an extraction zone at a maximum temperature from 50°F. to 100°F. and a maximum temperature gradient across said zone from 40°F. to 100°F.
- 6. A method in accordance with claim 2 wherein the reforming catalyst is platinum on alumina.
- 7. A method as in claim 6 wherein the solvent for aromatic hydrocarbons is SO_2 , and the extraction is accomplished at a solvent-to-hydrocarbon ratio of from 9:10 to 2:1 in an extraction zone at a maximum temperature from 50°F. to 100°F. and a maximum temperature gradient across said zone from 40°F. to 100°F.
- 8. A method as in claim 6 wherein the hydrofining catalyst is cobalt molybdate, and said hydrofining conditions comprise a temperature from 575°F. to 650°F., a pressure from 250 psig to 350 psig, a space velocity from 1.0 to 4.0 V/V/Hr., and a hydrogen-to-hydrocarbon ratio from 500 to 1500 scf/bbl.
- 9. A process of making a jet fuel which comprises:
 fractionating a hydrocarbon stream to obtain a charge
 stream boiling within the range of 355°F. to 510°F. and comprising
 from 30 to 50% normal paraffins, from 10 to 40% isoparaffins, from
 40 to 60% aromatic and naphthenic hydrocarbons, and from 0 to 5%
 olefins;

contacting said charge stream with from 200 to 600 scf of hydrogen per barrel of charge stream in the presence of a cobalt molybdate hydrodesulfurization catalyst, at a temperature from 600°F. to 690°F., a pressure of 150 psig to 300 psig, and a space velocity of 2 to 4 V/V/Hr., whereby there is obtained a desulfurized stream containing less than 0.01% sulfur;

contacting said desulfurized stream with from 2000 to 7000 scf of hydrogen per barrel of desulfurized stream in the presence of platinum on alumina reforming catalyst, at a temperature from 825°F. to 900°F., a pressure from 150 psig to 450 psig, and a space velocity from 0.2 to 2.0 V/V/Hr., whereby naphthene aromatization and normal paraffin isomerization are maximized in obtaining a reformed stream;

extracting substantially all of the aromatic hydrocarbons from said reformed stream by contact with from 0.9 to 2.0 volumes of SO₂ per volume of reformed stream in an extraction zone at a maximum temperature from 50°F. to 100°F. and a temperature gradient across said zone of 40°F. to 100°F., whereby there is obtained a raffinate containing less than 3% aromatic hydrocarbons;

fractionating said raffinate to obtain a raffinate fraction boiling below 480°F.;

hydrofining said raffinate fraction by contact with a cobalt molybdate hydrofining catalyst and from 500 to 1500 scf of hydrogen per barrel of raffinate fraction at a temperature from 575°F. to 650°F., a pressure from 250 psig to 350 psig, and a liquid hydrocarbon space velocity of 1.0 to 4.0 V/V/Hr., whereby an essentially olefin- and aromatic hydrocarbon-free hydrofined product is obtained;

fractionating said hydrofined product to obtain a hydrofined fraction boiling below 480°F. and having an Engler distillation 20% point of 410°F. to 414°F.; and

contacting said hydrofined fraction with activated charcoal at a space velocity of 0.5 to 4.0 V/V/Hr.,

whereby there is obtained a jet fuel having a luminometer number of at least 100, a heat of combustion of at least 18,900 Btu/lb., a maximum viscosity at -30°F. of 15 centistokes, and a maximum freezing point of -40°F.

