

## PATENT SPECIFICATION

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## COMPLETE SPECIFICATION.



## Improvements in Catalysts of High Mechanical Strength.

We, I. G. FARBENINDUSTRIE AKTIEN-GESELLSCHAFT, of Frankfurt-on-Main, Germany, a joint stock company organized under the laws of Germany, do hereby  
5 declare the nature of this invention and in what manner the same is to be performed, to be particularly described and ascertained in and by the following statement:—

10 Catalysts, especially those of oxide character, when employed in catalytic processes frequently cause trouble in working since, owing to their relatively  
15 low mechanical strength, dusty particles are carried away by the gases and vapours passing over or through the said catalysts, which choke up the apparatus, con-  
20 terminate the reaction product, and cause other difficulties.

25 In order to avoid these difficulties it has been proposed to increase the stability of the catalysts by incorporating therewith water glass or calcium com-  
30 pounds as binding agents. Also catalysts which comprise metallic halides or mag-  
35 nesium salts or magnesium oxide are well known in the art, which additions have been made either for the aforesaid pur-  
40 pose of stabilisation or for improving the catalytic efficiency.

45 We have now found that catalysts of high mechanical stability and free from the above mentioned defects, may be obtained by a specific method of adding  
50 water-soluble magnesium salts to the catalysts during their preparation, prior to their use. The method of carrying out the process according to our invention  
55 consists in making one of the components of the catalyst into a solid mass by means of one or more water-soluble magnesium salts, expelling the acid or acids of the  
60 magnesium salt or salts from the mass and adding thereto a solution of the other components of the catalyst. Oxide masses of great durability and especially suitable for the hydrogenation of oxides of carbon, are obtained in this manner. Additions  
65 of magnesium chloride, sulphate or nitrate, for example, are suitable for the said purposes.

The mechanical stability may be increased, to a remarkable degree, by suit-  
[Price 1/-]

ably selecting the working conditions in the process of production, without impair-  
55 ing the catalytic efficiency.

The following examples will further illustrate how the said invention may be carried into effect but the invention is not  
60 limited thereto. The parts are by weight.

## EXAMPLE 1.

Zinc oxide is made into a paste with a solution of magnesium chloride, so that the mixture contains about 10 parts of  
65 magnesium chloride to 100 parts of zinc oxide. The paste is then dried, pressed, and the hydrochloric acid contained there-  
70 in is expelled by heating in air or in a current of inert gas, the resulting product being then impregnated with chromic acid. If a mixture of 1 part of carbon  
75 monoxide and 3 parts of hydrogen be passed over this catalyst at about 400° Centigrade and under a pressure of about  
80 200 atmospheres, a liquid reaction product is obtained containing 95 per cent. of methanol, the yield and quality being  
85 the same as in the case of catalysts containing only the oxides of zinc and chromium, but the advantage being  
90 experienced according to the present invention that troubles due to catalyst dust are avoided.

## EXAMPLE 2.

The mass prepared from zinc oxide and magnesium chloride in the manner  
85 described in Example 1 is treated with a solution of alkali metal bichromate after the expulsion of the hydrochloric acid. If a mixture of 1 part of carbon monoxide  
90 and 3 parts of hydrogen be passed over this catalyst at about 450° Centigrade and under a pressure of about 200 atmospheres, a liquid product is obtained containing, in  
95 addition to methanol, about 30 per cent. of higher alcohols and other organic compounds containing oxygen. Other water-  
100 soluble magnesium salts, such as the sulphate, or nitrate, may also be employed in a similar manner.

In addition to preventing the forma-  
105 tion of dust, the advantage of the catalysts produced in this manner consists especially in the fact that the masses do not cake together in use.

Having now particularly described and

ascertained the nature of our said invention and in what manner the same is to be performed, we declare that what we claim is:—

5 1. A process for the manufacture and production of catalysts of high mechanical strength which consists in making one of the components of the catalyst into a solid mass by means of one or more water-soluble magnesium salts, expelling the  
10 acid or acids of the magnesium salt or salts from the mass, and adding thereto a solution of the other components of the catalyst.

2. The process for the manufacture and 15 production of catalysts of high mechanical stability substantially as described in each of the foregoing examples.

3. Catalysts of high mechanical 20 stability when prepared in accordance with the preceding claiming clauses.

Dated this 14th day of October, 1927.

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