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APPLICATION VOID.

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PATENT SPECIFICATION

Convention Date (Italy): Jan. 27, 1936.

493,531

Application Date (in United Kingdom): Dec. 23, 1936. No. 35268/36.

Specification not accepted

3367

COMPLETE SPECIFICATION

Process for the Preparation of Methyl Alcohol from Marble and the Production of Fuel therefrom

We, A.P.I.S. Soc. An. Proporting Italiania Sinteriol, a budy corporate organised under the laws of Italy, of 4, Via Vincenzo, Monti, Milan, Italy, do bereby declare the nature of this invention and in what manner the same is to be performed, to be particularly described and ascertained in and by the following statement:—

The present invention which refers to a novel process for the production of methyl alcohol and other organic compounds derived and derivable therefrom is characterised essentially in that the basic sub-

15 stance is carbonic acid obtained by the reaction of acids (preferably sulphuric acid) on marble powder. The carbonic acid thus obtained is reduced successively into exide of carbon which is combined with a 20 double volume of hydrogen and com-

double volume of hydrogen and compressed to 200 atm. which mixture when treated in a catalysing apparatus produces methyl alcohol.

The process is carried out most favour-25 ably in the manner described below, using apparatus which is already known and in common use in the chemical industry.

There is introduced into a receptacle of the desired capacity, provided with an agitator, a definite quantity of marble in powder form on which there is pouted gradually a 10% sulphuric acid solution. The sulphuric acid in contact with the marble produces carbonic acid gas which by means of piping is directed into a purifying column. The possible residues are discharged at the bottom of the receptacle.

discharged at the bottom of the receptacle.

The purifying column is filled with layers of wood charcoal. The carbonic 40 acid gas is purified, passing through the layers of wood charcoal from the bottom towards the top, from any traces of water which may be carried along by it.

The purified carbonic acid continues its
45 course towards a reducing apparatus,
heated by means of steam. In this
[Price 1]-]

reducer there is effected in the presence of wood charcoal and scraps of iron at a definite temperature, the reduction of the earbonic acid into carbon monoxide, which 50 passes from the reducer to a gasometer.

From this gasometer, containing carbon monoxide and a second gasometer containing hydrogen, there are taken by means of a suitable compressor, two volumes of 55 carbon monoxide and four volumes of hydrogen.

The mixture of these aspirated gases is compressed to 200 atm. and directed to a catalyser apparatus constituted for 60 example by an autoclave containing scraps of iron, caustic potash and oxide of zine which have been previously reduced and heated to 410° C.

The compressed mixture on coming into 65 contact with the catalyser produces methyl alcohol which on leaving the apparatus is passed into a heat exchanger which by reducing the pressure of the gases permits the methanol or methyl 70 alcohol to be condensed and at the same time the steam to be used either for the reducing apparatus or for the rectification. The crude methanol thus produced is then passed into a rectification and distillation apparatus. The distilled methanol is then condensed and collected in barrels ready to be used in commerce.

From the methyl alcohol thus obtained, the preparation of other organically 80 derived compounds can be proceeded with, by means of the conversion known in the chemical industry.

chemical industry.

One of the principal applications of the methanol is the preparation of an excel- 85 lent synthetic fuel.

The composition of this fuel is as

 $\{(x_i,y_i,x_i,x_i,x_i)\}_{i=1}^n$

Methyl alcohol - - 94.530% Synthetic camphor - - 0.010% 90 Acetone - - - 0.165%

		
	Ethyl ether 1.000% Terebenthine essence - 0.010% Benzene 4.285% Mirbene essence in drops -	coal and iron scrap, the hydrogenation of the carbon monoxide being then carried out under high pressure and high tem- perature, methyl alcohol being obtained therefrom.
ħ	To obtain the above-mentioned fuel the	2. Process according to claim 1, charac-
_	following procedure must be adopted:	terised by the fact that the marble powder
	There is introduced into a metal plate	is treated in an agitating apparatus a
	receptacle first the percentage shown of	"10% solution of sulphuric acid being
	synthetic camphor and then the percen-	poured gradually and in a continuous
10	tage of methanol or methyl alcohol.	manner on this powder.
*-	Using a wooden spatular the mixture is	3. Application of the methyl alcohol
	stirred until there is obtained a perfectly	obtained according to claims 1 and 2 as a
	limpid and clear solution of the two com-	fuel, characterised by this being mixed
	pounds: there is then added the percen-	in the following manner:
15	tage of acetone, of ether, of benzene and	20.00 2 2 2 3 3 4 5 1 3
	of trementhene, the mixture being stirred	Methyl alcohol obtained from marble 94.530%
	until there is again obtained perfect	
	limpidity of the fuel thus produced.	Synthetic camphor 0.010%
	This fuel has given excellent results in	Acetone 0.165% Ethyl ether 1.000%
20	tests made, in spite of the complete	Trementhine essence - 0.010%
	absence of natural spirit.	Benzene 4.285%
	Having now particularly described and	Mirbane essence in drops —
	ascertained the nature of our said inven-	Milipane essence in arops
	tion and in what manner the same is to	4. Process for the preparation of methyl
Zū	be performed we declare that what we claim is:	alcohol from marble and the production
	I. Process for the preparation of methyl	of fuel therefrom substantially as
	alcohol and other organic compounds	described.
	derived or derivable therefrom, character-	1. Practice 1
QΛ	ised in that carbonic acid obtained from	Dated this 23rd day of December, 1936.
DO.	the reaction of sulphuric acid on marble	WHEATLEY & MACKENNIE,
	in powder form is reduced to carbon	40. Chancery Lane, London W.C.2.
	monoxide in the presence of wood char-	Agents.
		- 19a, 10 g

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