

# PATENT SPECIFICATION

*Inventor:* MAURICE ARTHUR ERIC HODGSON

**680,865**



*Date of filing Complete Specification July 20, 1951.*

*Application Date Aug. 4, 1950.*

*No. 19518/50.*

*Complete Specification Published Oct. 15, 1952.*

Index at acceptance:—Classes 1(i), F(3a2: 4c); and 2(iii), C3a13a3(alc: e).

## COMPLETE SPECIFICATION

### Improvements in and relating to Reactors and Reactions Carried out therein

We, IMPERIAL CHEMICAL INDUSTRIES LIMITED, a British Company, of Imperial Chemical House, Millbank, London, S.W.1, do hereby declare the invention, for which we pray that a patent may be granted to us, and the method by which it is to be performed, to be particularly described in and by the following statement:—

10 Various reactions, such as the hydrogenation of high-boiling liquids and the direct hydration of olefines, require a component or components predominantly in the liquid phase and a component  
15 or components predominantly in the vapour phase, the two components having to be contacted with each other over a solid catalyst which may be in the form of pellets. The liquid portion of the material fed to the converter  
20 or part of this portion may be required for the reaction itself, and/or it may be required for the purpose of scrubbing out the reaction product. In either case the  
25 maximum output of desired product requires intimate contact between the liquid and vapour phases and between one or both of these phases and the catalyst.

30 The simplest form of converter would consist of a cylindrical vessel of suitable dimensions packed with catalyst pellets or other suitable form of solid catalyst, the material which is predominantly liquid  
35 under the conditions of reaction being fed to the top of the converter and distributed uniformly over the catalysts bed, the predominantly vapour phase reactant being caused to flow co-currently or  
40 counter-currently as desired.

This type of converter suffers from the following disadvantages:—

45 (1) It is difficult to maintain good distribution of the liquid phase over the catalyst throughout the converter as is necessary to obtain a good output of product in the case when the liquid takes part

in the reaction. This difficulty is particularly serious if the catalyst consists of pellets small compared with the diameter of the converter and/or of material which is not readily wetted by the liquid. The liquid then tends to form a limited number of relatively large channels. 55

(2) At constant space velocity, i.e. constant volume fed per hour per unit of bulk volume of catalyst, the liquid loading i.e. the quantity of liquid fed per hour to unit cross section of the catalyst bed, is proportional to the depth of the bed, which may be 30 feet or more in commercial scale converters. The large liquid loading under these conditions can lead to flooding of the bed, with consequent loss of output in cases where the reaction takes place predominantly in the vapour phase; it can also under certain circumstances, particularly in the case of high pressure reactions, cause serious catalyst erosion, thereby shortening the life of the catalyst and contaminating the product with catalyst material. 70

It is an object of this invention to provide a reactor and reaction conditions by which these difficulties can be overcome. 75

According to the invention, there is provided a converter for reactions in which at least one component in the liquid phase such as a reactant or solvent is contacted with at least one further component which may be in the liquid or gaseous phase and in which there is present at least one solid component such as a catalyst or heat reservoir, which converter contains at least one sheet of gauze positioned to ensure that any part of said solid component or components is within a desired maximum horizontal distance from such sheet, the gauze consisting of a material which under reaction conditions is wetted by said liquid component or components and having a 85 90

weave in the form of galloon netting, and means for feeding said liquid phase component or components on to the said gauze, said sheet of gauze being positioned to ensure that liquid flows from its upper to its lower part by gravity.

Preferably the sheets are disposed vertically. Also preferably the weft threads, which in galloon netting are arranged close to one another, are disposed horizontally. The warp threads, which in galloon netting are substantially straight, are then preferably disposed vertically.

The gauze, owing to its galloon netting weave, is self-wetting i.e. liquid flowing within its interstices will spread to wet a substantial area without the need of pre-flooding.

In this specification the term "gaseous" is intended to refer to those substances which are usually gaseous and those which are in the vapour form under the conditions of operation.

It is to be understood that the material of the gauze will be so chosen that it is not attacked under reaction conditions.

The gauze may be arranged in any of several forms, e.g. with a continuously spiral plan view, as a series of coaxial cylinders or as a series of parallel sheets. It should preferably be so arranged that no part of the catalyst bed is more than about 2 inches or, more preferably, about 1 inch from the nearest gauze surface.

The reactor design according to this invention is applicable to a large variety of reactions, particularly, though not exclusively, to those which take place partly in the liquid and partly in the gaseous phase, and particularly, though not exclusively, to catalytic reactions, especially when a solid catalyst is used. Examples of such reactions are the carbonylation of olefines by carbon monoxide and hydrogen to aldehydes or alcohols using solid catalysts which will hereinafter be briefly referred to as "carbonylation" and the hydrogenation of cresote or fuel oil using solid catalysts. A particularly important example of the application of the invention is that to the direct catalytic hydration of olefines such as ethylene and propylene at high pressure, say 250 atmospheres, and high temperatures, say 280° C. to 300° C.

A converter constructed according to the invention is illustrated in perspective, partly cut away, and in plan, in the single figure of the drawing accompanying the provisional specification. In this figure, 1 is the converter shell and 2 a catalyst basket extending along most of the length of the converter. The basket contains three coaxial cylinders of galloon netting indicated at 3, 4 and 5,

and is filled in operation with catalyst material, preferably substantially up to the top ends of cylinders 3, 4 and 5. The liquid material flows down the converter through a distributor plate 6 having three circular rows of openings corresponding to the cylinders 3, 4 and 5, a tube 7 being connected to each of these openings, so as to direct the liquid on to the cylinders. The gaseous or vaporous material may be fed co-currently or countercurrently to the liquid flow as desired.

#### EXAMPLE.

A cylindrical converter for the direct catalytic hydration of propylene to isopropanol over a solid pelleted catalyst essentially consisting of a blue oxide of tungsten had an overall height of 70 feet obtained by operating three shorter converters in series. These three vessels will hereinafter be referred to as one single converter. Its diameter was 3 feet 8 inches. It was operated at a pressure of 250 atmospheres and a temperature of 270° C. by introducing into it, at a rate of 3 litres per litre of catalyst bulk space per hour, 7 parts by weight of water and 1 part of propylene. It was found that the output of isopropanol per litre of catalyst bulk space per hour was 0.14 kgs. The converter in this form was used for comparison purposes only and is not claimed. The converter was then modified by fitting it with 22 cylinders of stainless steel galloon netting all coaxial with the converter and spaced at regular intervals of 2 inches so that no point in the converter was more than one inch distant from the nearest wire cylinder. Upon operating the modified converter as before it was found that the isopropanol output has risen to 0.20 kgs. per litre of catalyst bulk space per hour.

What we claim is:—

1. A converter for reactions in which at least one component in the liquid phase such as a reactant or solvent is contacted with at least one further component which may be in the liquid or gaseous phase and in which there is present at least one solid component such as a catalyst or heat reservoir, this converter containing at least one sheet of gauze positioned to ensure that any part of said solid component or components is within a desired maximum horizontal distance from such sheet, the gauze consisting of a material which under reaction conditions is wetted by said liquid component or components and having a weave in the form of galloon netting, and means for feeding said liquid phase component or components on to the said gauze, said sheet of gauze being posi-

tioned to ensure that liquid flows from its upper to its lower part by gravity.

2. A converter as claimed in Claim 1 in which said sheet of gauze is disposed vertically within the converter.

3. A converter as claimed in Claim 1 or 2 in which the weft threads of the galloon netting are disposed horizontally.

4. A converter as claimed in any of the preceding claims in which a sheet or sheets of galloon netting is or are arranged to ensure that no part of said solid material is more than 2 inches distant horizontally from the nearest gauze surface.

5. A converter as claimed in Claim 4 in which a sheet or sheets of galloon netting is or are arranged to ensure that no part of said solid material is more than 1 inch from the nearest gauze surface.

6. A converter as claimed in any of the preceding claims containing a sheet of galloon netting formed to have a continuously spiral cross-section.

7. A converter as claimed in any of

Claims 1 to 5 containing a cylindrical sheet of galloon netting or a series of such cylindrical sheets preferably disposed coaxially.

8. A converter as claimed in any of Claims 1 to 5 containing a plane sheet of galloon netting or a series of such plane sheets preferably disposed parallel to each other.

9. A converter whenever constructed substantially as described by reference to the drawing accompanying the provisional specification.

10. A converter whenever constructed substantially as described in the foregoing example.

11. A method of direct catalytic hydration of olefine, of carbonylation of an olefine with carbon monoxide or hydrogen to an aldehyde or alcohol or of hydrogenating cresote or a fuel oil whenever carried out in a converter as claimed in any of the preceding claims.

J. W. RIDSDALE,

Solicitor for the Applicants.

#### PROVISIONAL SPECIFICATION

#### Improvements in and relating to Reactors and Reactions Carried out therein

We, IMPERIAL CHEMICAL INDUSTRIES LIMITED, a British Company, of Imperial Chemical House, Millbank, London, S.W.1, do hereby declare this invention to be described in the following statement:—

Various reactions, such as the hydrogenation of high-boiling liquids and the direct hydration of olefines, require a component or components predominantly in the liquid phase and a component or components predominantly in the vapour phase, the two components having to be contacted with each other over a solid catalyst which may be in the form of pellets. The liquid portion of the material fed to the converter or part of this portion may be required for the reaction itself, and/or it may be required for the purpose of scrubbing out the reaction product. In either case the maximum output of desired product requires intimate contact between the liquid and vapour phases and between one or both of these phases and the catalyst.

The simplest form of converter would consist of a cylindrical vessel of suitable dimensions packed with catalyst pellets or other suitable form of solid catalyst, the material which is predominantly liquid under the conditions of reaction being fed to the top of the converter and distributed uniformly over the catalyst bed, the predominantly vapour phase

reactant being caused to flow concurrently or countercurrently as desired.

This type of converter suffers from the following disadvantages:—

(1) It is difficult to maintain good distribution of the liquid phase over the catalyst throughout the converter as is necessary to obtain a good output of product in the case when the liquid takes part in the reaction. This difficulty is particularly serious if the catalyst consists of pellets small compared with the diameter of the converter and/or of material which is not readily wetted by the liquid. The liquid then tends to form a limited number of relatively large channels.

(2) At constant space velocity, i.e. constant volume fed per hour per unit of bulk volume of catalyst, the liquid loading, i.e. the quantity of liquid fed per hour to unit cross section of the catalyst bed, is proportional to the depth of the bed, which may be 30 feet or more in commercial scale converters. The large liquid loading under these conditions can lead to flooding of the bed, with consequent loss of output in cases where the reaction takes place predominantly in the vapour phase; it can also under certain circumstances, particularly in the case of high pressure reactions, cause serious catalyst erosion, thereby shortening the life of the catalyst and contaminating

the product with catalyst material.

It is an object of this invention to provide a reactor and reaction conditions by which these difficulties can be overcome.

- 5 According to the invention, vertical sheets of gauze are suitably disposed throughout the bed, the material of the gauze being such as to be wetted by the liquid phase, and the weave being in the form of galloon netting, with the warp threads, which in this type of netting are substantially straight, preferably disposed vertically while the weft threads, which in this type of netting are arranged closely against one another, preferably disposed horizontally. The gauze will then be self-wetting, i.e. liquid flowing within its interstices will spread to wet a substantial area without the need of pre-flooding.

- 20 The gauze may be arranged in any of several forms, e.g. with a continuously spiral plan view, as a series of coaxial cylinders or as a series of parallel sheets. 25 It should preferably be so arranged that no part of the catalyst bed is more than about 2 inches from the nearest gauze surface.

- 30 The reactor design according to this invention is applicable to a large variety of reactions, particularly, though not exclusively, to those which take place partly in the liquid and partly in the

gaseous phase, and particularly, though not exclusively, to catalytic reactions, especially when a solid catalyst is used. An important example of the application of the invention is that to the direct catalytic hydration of olefines such as ethylene and propylene at high pressure, say 250 atmospheres, and high temperatures, say 280 to 300° C.

A converter constructed according to the invention is illustrated in perspective, partly cut away, and in plan, in the single figure of the accompanying drawing. In this figure, 1 is the converter shell and 2 a catalyst basket extending along most of the length of the converter. The basket contains three coaxial cylinders of galloon netting indicated at 3, 4 and 5, and is filled in operation with catalyst material, preferably substantially up to the top ends of cylinders 3, 4 and 5. The liquid material flows down the converter through a distributor plate 6 having three circular rows of openings corresponding to the cylinders 3, 4 and 5, a tube 7 being connected to each of these openings, so as to direct the liquid on to the cylinders. The gaseous or vaporous material may be fed co-currently or countercurrently to the liquid flow as desired.

J. W. RIDSDALE,  
Solicitor for the Applicants.

