## **PCT**

## WORLD INTELLECTUAL PROPERTY ORGANIZATION



## INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(51) International Patent Classification 7:
C10G 2/00, C07C 31/04, 1/04

(11) International Publication Number: WO 00/68339

(43) International Publication Date: 16 November 2000 (16.11.00)

(21) International Application Number: PCT/US00/12808

(22) International Filing Date: 8 May 2000 (08.05.00)

(30) Priority Data:

60/133,147 7 May 1999 (07.05.99) US

(71) Applicant (for all designated States except US): RENTECH, INC. [US/US]; 12401 E. 37th Avenue, Denver, CO 80239–3404 (US).

(72) Inventors; and

(75) Inventors/Applicants (for US only): SHEPARD, Richard, O. [US/US]; 1374 Kerr Gulch Road, Evergreen, CO 80439 (US). DAVIS, Robert, B. [US/US]; 1937 W. Vandalia, Broken Arrow, OK 74012 (US). LEONARD, Thomas, M. [US/US]; 4145 E. 35th Place, Tulsa, OK 74135–1741 (US).

(74) Agent: MARTIN, Rick; Patent Law Offices of Rick Martin, PC, 416 Coffman Street, Longmont, CO 80501 (US).

(81) Designated States: AE, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CR, CU, CZ, DE, DK, DM, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, US, UZ, VN, YU, ZA, ZW, ARIPO patent (GH, GM, KE, LS, MW, SD, SL, SZ, TZ, UG, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG).

#### Published

With international search report.

Before the expiration of the time limit for amending the claims and to be republished in the event of the receipt of amendments.

(54) Title: CONVERTIBLE METHANOL/FISCHER-TROPSCH PLANT AND METHOD

#### (57) Abstract

Lower molecular weight organic materials are converted to more valuable products in a variable flow, selectively operated multiproduct plant. The plant includes a reformer (12, 25) to produce syngas from the hydrocarbon feed, multi output valving (13) to direct the syngas product stream to one or both of a methanol reactor (18) and a Fischer–Tropsch (FT) products plant (19), depending on the ambient value of these chemicals. Prior to introducing the syngas into the FT plant 19, the  $CO_2$  and  $H_2$  are removed by scrubbers (12, 13) and recycled appropriately.

## FOR THE PURPOSES OF INFORMATION ONLY

Codes used to identify States party to the PCT on the front pages of pamphlets publishing international applications under the PCT.

AL	Albania	ES	Spain	LS	Lesotho	SI	Slovenia
AM	Armenia	FI	Finland	LT	Lithuania	SK	Slovakia
AT	Austria	FR	France	LU	Luxembourg	SN	Senegal
AU	Australia	GA	Gabon	LV	Latvia	SZ	Swaziland
AZ	Azerbaijan	GB	United Kingdom	MC	Monaco	TD	Chad
BA	Bosnia and Herzegovina	GE	Georgia	MD	Republic of Moldova	TG	Togo
BB	Barbados	GH	Ghana	MG	Madagascar	TJ	Tajikistan
BE	Belgium	GN	Guinea	MK	The former Yugoslav	TM	Turkmenistan
BF	Burkina Faso	GR	Greece		Republic of Macedonia	TR	Turkey
BG	Bulgaria	HU	Hungary	ML	Mali	TT	Trinidad and Tobago
ВJ	Benin	IE	Ireland	MN	Mongolia	UA	Ukraine
BR	Brazil	IL	Israel	MR	Mauritania	UG	Uganda
BY	Belarus	IS	Iceland	MW	Malawi	US	United States of Americ
CA	Canada	IT	Ĭtaly	MX	Mexico	UZ	Uzbekistan
CF	Central African Republic	JP	Japan	NE	Niger	VN	Viet Nam
CG	Congo	KE	Kenya	NL	Netherlands	YU	Yugoslavia
CH	Switzerland	KG	Kyrgyzstan	NO	Norway	$\mathbf{z}\mathbf{w}$	Zimbabwe
CI	Côte d'Ivoire	KP	Democratic People's	NZ	New Zealand		
CM	Cameroon		Republic of Korea	PL	Poland		
CN	China	KR	Republic of Korea	PT	Portugal		
CU	Cuba	KZ	Kazakstan	RO	Romania		
$\mathbf{CZ}$	Czech Republic	LC	Saint Lucia	RU	Russian Federation		
DE	Germany	LI	Liechtenstein	SD	Sudan		
DK	Denmark	LK	Sri Lanka	SE	Sweden		
EE	Estonia	LR	Liberia	SG	Singapore		

## CONVERTIBLE METHANOL/FISCHER-TROPSCH PLANT AND METHOD

## **CROSS-REFERENCE TO RELATED APPLICATION**

This application is a non-provisional application claiming the benefits of provisional application No. 60/133,147 filed May 7, 2000.

### FIELD OF THE INVENTION

This invention relates to improved plants and processes for the manufacture of organic chemicals from natural gas and other low molecular weight feedstocks.

#### BACKGROUND OF THE INVENTION

All industries go through cycles of excess capacity and not enough capacity. The chemical industry is no exception. Changing industry economics due to changes in public tastes, governmental regulation and environmental considerations all affect the industry. For example, the finding of methyl t-butyl ether (MTBE) in California ground waters is leading to legislation forbidding its use as a gasoline additive. Methanol is a raw material for the manufacture of MTBE and this legislation combined with over capacity may render some methanol plants uneconomic.

Basically, the plant of this invention is designed to ameliorate such problems for methanol plants. The plants combine a syngas-producing reformer of a methanol reactor with a Fischer-Tropsch (FT) reactor. These problems are discussed at some length by Frank C. Brown in Petroleum Technology Quarterly, Spring 2000, pp. 84-90 with regard to both the industry

and the invention herein. The plant is piped to eliminate or modulate the production of one of the hydrocarbons, alcohols and hydrogen products of the combined plant so as to produce the most profitable product or mix of products at any given time. The plants can be easily separated for upgrade or the scrapping of either the methanol or FT plant

Each of the reactors of the new plants and their operations is well known. Thus, the T.E. O'Hare et al, paper given to the 1986 International Gas Research Conference, Sept. 8-11, 1986 in Toronto Canada describes the "Liquefaction of Natural Gas to Methanol for Shipping and Storage". The specific technology was developed by the Department of Applied Science, Brookhaven National Laboratory, Associated Universities, Inc. The process described utilizes a four-part operation. Natural gas and steam are subjected to catalytic secondary reforming to obtain syngas. The syngas is sparged through a liquid catalyst solution and then cooled to recover a methanol stream at 120°C at about 194 psi absolute.

U.S. Patents 5,621,155; 5,620,670 and 5,543,437 issued to C. B. Benham, M. S. Bohn and D. L. Yakobson of Rentech, Inc. of Denver, Colorado. These patents teach a process utilizing a natural gas feedstock which is reformed to form a syngas which, after carbon dioxide removal, is fed to a Fischer-Tropsch (FT) reactor to form water, a variety of alcohols and  $C_5$  and  $C_{20}$  hydrocarbon products, e.g., liquids and waxes. These references are exemplary of an extensive body of literature.

#### SUMMARY OF THE INVENTION

A multipurpose plant package includes a methanol plant and a FT plant utilizing a shared syngas reformer. The reformer is plumbed to feed all its product to the methanol plant or to the FT plant or to both plants simultaneously.

To manufacture methanol, the reformer syngas product is compressed and fed to a methanol reactor to provide a methanol stream. To provide both methanol and FT products, the CO<sub>2</sub> and H<sub>2</sub> are removed from a portion of the syngas stream and the purified syngas is fed to the FT reactor while the remaining portion of syngas stream is fed to the methanol reactor. To obtain only FT products, the entire syngas stream is fed to the FT reactor.

### BRIEF DESCRIPTION OF THE DRAWINGS

Figure 1 is a functional diagram of one form of a three-reactor plant of this invention.

Figure 2 is a diagram of a preferred FT process.

## **DETAILED DESCRIPTION OF THE DRAWINGS**

Figure 1 depicts a combined methanol and FT reactor adaptable for the primary production of methanol and FT products. Block A identifies the existing methanol plant less the reformer.

Steam 10 and natural gas 11 are piped to the reformer 12. When the market justifies the production of only methanol, valve 13 directs all the reformer 12 output to a syngas compressor 14. The compressed syngas is fed

to the methanol reactor 15. Reactor 15 products are distilled in tower 16 from whence methanol is recovered and the high ends are flared or recycled as fuel.

To operate totally as a FT products plant, valve 13 is used to direct all of the syngas to a  $CO_2$  absorption unit 17. The recovered  $CO_2$  is then recycled to the reformer 12 while the remaining gases are treated in an  $H_2$  removal unit 18. The  $H_2$  is preferably used as fuel for the reformer or marketed. The remaining syngas is fed to the FT reactor 19. The  $H_2$ :CO ratio of the FT reactor feed is preferably about 2:1 or less. The remaining FT reactor 19 products are passed into the FT product separation unit 21. Tail gas from column 21 can be flared, used as reformer 12 fuel or used as a feedstock.

Figure 2 depicts a more preferred flowsheet based on the use of the entire output of reformer 12 to produce methanol FT producers. In the diagram of Figure 2, the methanol plant has been taken off line through the use of the valving 13 of Fig. 1. The synthesis gas and steam feedstocks utilized to produce methanol are now introduced into the FT plant of Fig. 2. The natural gas and steam are fed to the reformer 25 through the piping of line 26. Fuel for heating reformer 25 is supplied through piping 27. The hot gases from reformer 25 are cooled, then passed through compressor 28 and introduced into CO<sub>2</sub> absorption unit 29. CO<sub>2</sub> from absorber 29 is recycled through compressor 31 to reformer 25. The remaining synthesis gas is passed through H<sub>2</sub> recovery unit 30 where the H2:CO ratio is reduced to below 2:1. Then the steam produced in the FT reactor 32 can be used for power generation. The remaining products exiting reactor 32 are then separated in product separator 33. Water and hydrocarbons can be recovered while the hydrocarbon gases

are recycled for use as a fuel, as a feed to reformer 25 or as feed to the FT reactor 32 via line 33 and compressor 34.

The percentage of the CO<sub>2</sub> removed from the syngas leaving the reformer should reduce the CO<sub>2</sub> content of the H<sub>2</sub> scrubber 30 feeds to about 5% or less and preferably as low as 0.5%. The operation of the H<sub>2</sub> scrubber 30 should provide the FT reactor 32 with a feed having a H<sub>2</sub>:CO ratio of less than 2:1 and greater than 1:1. The tail gas recycle to the reformer 25 should be between 0% and 20% of the total tail gas from product separator 33. The tail gas recycled to the FT reactor 32 via CO<sub>2</sub> scrubber 29 should be 80 to 100%. Any remaining tail gas is available for use as reformer fuel.

## **GENERAL TEACHING OF THE INVENTION**

The Figures depict the skeletal backbone of the multiple product chemical plants. Computerized controls and equipment are necessary to maintain fluid and gas volume throughputs, the desired temperatures and pressures. The sensors needed to provide data for actuating the controls and the conditions of practical and optimum unit operations of each of the reformer and methanol and FT reactors are well known.

The preferred reformer process for operation with the preferred FT unit operation utilizes a nickel-based, alkali-promoted catalyst which permits a substantial CO<sub>2</sub> input, e.g., about 20 to about 40% and preferably about 30 to about 35% and a reduced steam input of about 2.2:1 steam to carbon. The pressure drop across the reformer will range from about 30 to 80 psi and preferably about 35-65 psi.

The FT reactor is operated at temperatures of 220 to 280°C and preferably 230 to 270°C and more preferably at 250 to 265°C. The FT reactor operating pressures are generally 100 to 600 psia; preferably from 150 to 550 psia and more preferably 300 to 500 psia. The term "scrubber" is defined here as any equipment utilized to remove the identified syngas component. The methanol reactor operating temperature is from 150 to 300°C, preferably from 175 to 250°C and more preferably 200 to 235°C. The methanol reactor operating pressures are 700 – 1400 psi, preferably 750 to 1300 psi and more preferably 800 – 1200 psi absolute. The FT reactor is preferably operated utilizing the preferred and most preferred conditions and catalysts of the Rentech, Inc. references cited previously. The methanol reactor catalyst can be any of copper, zinc and aluminum.

The reformer operation is often the plant bottleneck. One or more of pressure drop, burner firing rate, tube heat reflux emissions, or a combination of these three will limit the total plant capacity. For FT plants, the design objectives include maximizing the production of  $C_5+$  hydrocarbons, minimizing consumption of natural gas, and minimizing the size of the major unit operations and hence capital cost.

Iron-based FT catalysts promote the water-gas shift reaction:

$$CO + H_2O = H_2 + CO_2$$

Carbon dioxide is therefore a byproduct of the FT reaction. As such, it is important to recycle the carbon dioxide back to the reformer to improve carbon efficiency. The carbon dioxide scrubber unit shown in Figs. 1 and 2 delivers carbon dioxide to the reformer where it is partially converted back to carbon monoxide by the water-gas shift reaction in the reformer. The carbon dioxide

absorbers preferably remove carbon dioxide to low levels, preferably to approximately 0.5% on a volume basis in the exiting syngas stream, so that the maximum carbon production efficiency can be achieved.

The hydrogen removal unit is used to ensure that the FT reactor is fed with a reasonable ratio of hydrogen to carbon monoxide. For the example calculations, the ratio was set to 1:4:1.

As noted previously, the tail gas from the FT reactor, after separation of the liquid products and carbon dioxide, is split three ways. Approximately 5% is purge gas and can be used as fuel. Approximately 90% is recycled to the CO<sub>2</sub> removal unit in order that unreacted CO can be further converted in the FT reactor. The remaining tail gas is sent to the reformer where light hydrocarbons produced by the FT reactor can be reformed into carbon monoxide and hydrogen. This small recycle stream reduces the build up of light hydrocarbons in the main recycle loop.

The flowsheet in Fig. 2 was modeled to determine yield, natural gas consumption/reformer heat duty and pressure drop, and liquid hydrocarbon yield based on a plant which originally produced 75,000 gallons/day of methanol. The table below gives the important results.

Reformer heat duty (net)	137 MMBTU/hr		
Reformer pressure drop	60 psid		
Total natural gas consumption	9.94 MMSCFD		
C <sub>5</sub> -yield	932 bbl/day		
Plant efficiency	10.1 MMBTU/bbl		

All fuel MMBTU are lower heating values.

Feed gas flow of the major unit operations are given below:

Reformer	46
CO <sub>2</sub> absorber 1	82
H <sub>2</sub> removal	67
FT	60
Product separation	44

All flows are given in MMSCFD (standard temperature and pressure are  $60^{\circ}F$  and 14.69 psia). The preferred catalyst and reactor for the FT plant is precipitated iron catalyst with potassium and copper promoters in a slurry reactor. The catalyst mass concentration ratios in the slurry generally range from about 10 to about 20%, more preferably from about 10 to about 20% and most preferably about 15% (oxide basis) with a preferred catalyst size of about  $30~\mu m$ . The space velocity is about 1.5 to about 6, preferably about 2 to about 5 and most preferably 2.5 normal liters syngas per hour per gram of catalyst.

The superficial velocity ranges from about 3 to about 30 and preferably from about 7.5 to about 12 cm/sec.

For methanol plants, syngas is generally compressed to about 1100 psig. The compressor of the methanol plant can be used to advantage for the FT retrofit, especially if the methanol plant is off-line or mothballed. The compressor after the reformer in Fig. 2 is one stage of the methanol plant compressor. It boosts the syngas from the reformer outlet from about 225 psig to about 500 psig. The benefit is three-fold. First, the back end of the plant operates more efficiently at higher pressure. Second, smaller vessels are required. Third, recycle loops are facilitated. For example, minimal compression of the tail gas is needed to recycle it to the reformer or to the first carbon dioxide absorber. The second compressor shown in Fig. 2 boosts the

carbon dioxide pressure from the scrubber to the reformer inlet and can be another stage of the original methanol compressor. The FT product compressor 34, shown in Fig. 2, boots the tail gas pressure for the separator 33 to that of the CO<sub>2</sub> scrubber 29. The FT product compressor 34 can be another stage of methanol compressor 28.

1	CLAIMS
2	What is claimed is:
3	Claim 1. A reactor package for the production of methanol and/or
4	FT products and hydrogen comprising
5	a syngas reformer having steam, lower molecular weight hydrocarbon
6	and recycle CO <sub>2</sub> input piping; and output piping means for piping syngas
7	product to a multi output valve;
8	a methanol plant having a syngas compressor piped to receive syngas
9	from the multi output valve and to feed the compressed syngas into a methanol
10	reactor, the methanol reactor having piping connected to deliver methanol to a
11	distillation tower and the distillation tower having one or both of an exhaust flare
12	and/or recycle piping to return at least one component of the exhaust to the
13	syngas reformer; and
14	a FT reactor plant having at least one CO <sub>2</sub> scrubber and at least one H <sub>2</sub>
15	scrubber for purifying the syngas received from the syngas reformer and piping
16	directing the syngas stream from the multiproduct valve through the CO2 at
17	least one scrubber and the H <sub>2</sub> scrubber into the FT reactor, piping exhausting
18	the FT reactor products into a product separator means to recover the FT
19	products and further piping to conduct the FT reactor light gases to at least one
20	of a flare to the reformer and to the FT reactor.
21	Claim 2.
22	The reactor package of Claim 1 wherein the reformer contains an alkali
23	promoted nickel catalyst.

1	Claim 3.
2	The reactor package of Claim 2 wherein the catalyst usage permits a
3	CO <sub>2</sub> feedstock having a molar of about 20% to about 40%.
4	Claim 4.
5	The reactor package of Claim 2 wherein the catalyst permits a CO2
6	feedstock of about 30 to about 35%.
7	Claim 5.
8	The reactor package of Claim 2 wherein the reformer feedstock has a
9	steam- to-carbon ratio of about 2.2:1.
10	Claim 6.
11	The reactor package of Claim 2 preconfigured to operate within a
12	pressure drop range of about 50 to about 80 psi.
13	Claim 7.
14	The reactor package of Claim 6 preconfigured to operate within a
15	pressure drop range of about 55 to about 65 psi.
16	Claim 8.
17	The reactor package of Claim 1 wherein the FT reactor contains a
18	potassium and copper promoted precipitated iron catalyst of particle sizes of
19	0.5 to 1.00 microns.
20	Claim 9.
21	The reactor package of Claim 1 wherein the FT reactor contains a
22	potassium and copper promoted precipitated iron catalyst of particle sizes of 2
23	to 6 microns.
24	
25	

1	Claim 10.
2	The plant of Claim 8 wherein the FT reactor contains an amount of
3	catalyst during operation which ranges from about 10 to about 30% of the slurry
4	by mass.
5	Claim 11.
6	The reactor of Claim 6 wherein the reactor is preconfigured to operate
7	within a space velocity range of about 2.3 normal liters of syngas per hour per
8	gram of catalyst.
9	Claim 12
10	The reactor of Claim 6 wherein the reactor is preconfigured to operate
11	within a space velocity range of about 2.5 normal liters of syngas per hour per
12	gram of catalyst.
13	Claim 13.
14	The reactor of Claim 6 including controllers to maintain a superficial
15	throughput velocity ranging from about 5 to about 30 cm/sec.
16	Claim 14.
17	The reactor of Claim 6 including controllers to maintain a superficial
18	throughput velocity ranging from about 5 to about 7.5 cm/sec.
19	Claim 15.
20	A multi-purpose plant package which includes a methanol plant and a
21	FT plant utilizing a shared reformer comprising:
22	feedstock plumbing means for introducing at least steam and
23	substantially hydrocarbon feedstocks into the syngas reformer;

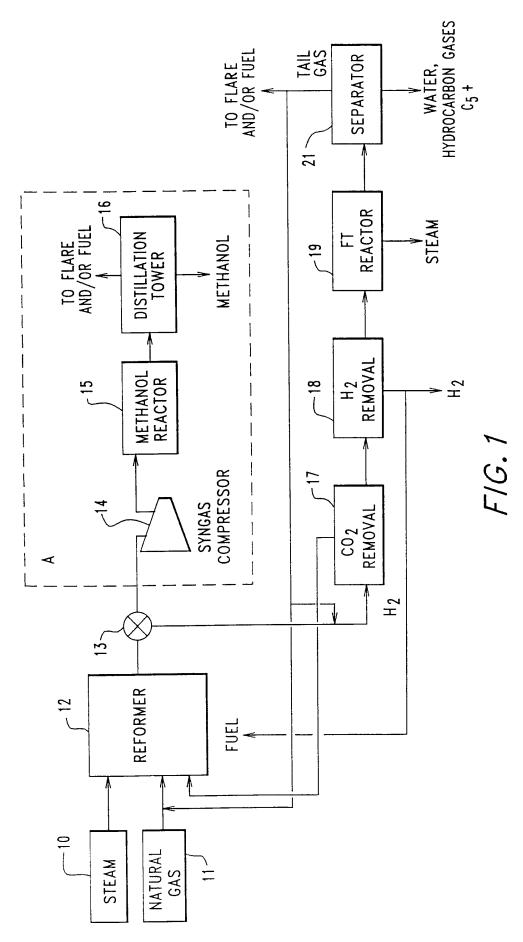
reformer means for producing syngas;

PCT/US00/12808 WO 00/68339

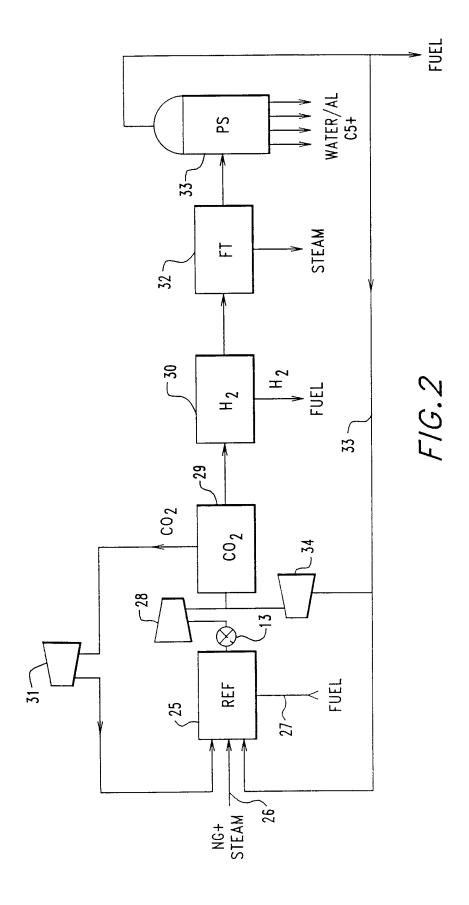
1

multi-purpose valve means; plumbed to the reformer, to a syngas 2 compressor, and to a CO<sub>2</sub> scrubber; for controlling the flow of syngas from the reformer to the methanol plant and to the FT plant; 3 a syngas compressor, plumbed to the methanol reactor means, for 4 conveying compressed syngas to the methanol reactor means, to a CO2 5 6 scrubber and to receive gaseous FT product from the FT product separator; 7 methanol reactor means, plumbed to a distillation tower, for separating 8 methanol and by-products; CO<sub>2</sub> means for removing CO<sub>2</sub> from the syngas plumbed to a hydrogen 9 10 separator means; hydrogen separator means, plumbed to FT reactor means, for removing 11 12 hydrogen from the syngas; FT reactor means, plumbed to an FT reactor product separator means, 13 14 for converting syngas to FT products; and FT product separator means for separating FT gas, liquid and solid 15 products. 16 Claim 16. 17 The multi-purpose plant package of Claim 15 further including plumbing 18 means for transferring gaseous FT products from the FT reactor product 19 separator means to a FT reactor product compressor for transferring the FT 20 product to the CO<sub>2</sub> scrubber. 21 Claim 17. 22 The multi-purpose plant package of Claim 15 further including plumbing 23 means for transferring compressed CO2 from the CO2 scrubber to the CO2 24

1	recycle compressor and further including plumbing and a CO <sub>2</sub> compressor					
2	means for returning compressed CO <sub>2</sub> to the reformer means.					
3	Claim 18.					
4	A process for operating a multiple process, multiple product chemical					
5	plant for the production of methanol and/or FT products and hydrogen					
6	comprising:					
7	introducing into a syngas reformer a sweet lower molecular hydrocarbon					
8	feedstock and water, reacting the feedstock and water utilizing a commercial					
9	nickel catalyst to produce a syngas stream containing CO2 and H2 and					
10	introducing air and the compressed syngas into a methanol reactor to form a					
11	methanol stream, and removing CO2 and hydrogen to produce higher purity					
12	syngas, CO <sub>2</sub> and H <sub>2</sub> streams;					
13	segregating the CO <sub>2</sub> for recycle to the gaseous FT product compressor					
14	means and/or for other usage;					
15	segregating the H <sub>2</sub> for recycle to the syngas reformer and/or other					
16	usage;					
17	introducing the syngas stream into a FT reactor and producing FT solids,					
18	liquids and gases and segregating the FT products.					
19	Claim 19. All inventions taught herein.					
20						
21						



SUBSTITUTE SHEET (RULE 26)



SUBSTITUTE SHEET (RULE 26)

# INTERNATIONAL SEARCH REPORT

ir. Ational Application No PCT/US 00/12808

		101/03 0	07 12000				
A. CLASSIF IPC 7	C10G2/00 C07C31/04 C07C1/04	1					
According to	International Patent Classification (IPC) or to both national classific	ation and IPC					
B. FIELDS	SEARCHED						
Minimum do IPC 7	Minimum documentation searched (classification system followed by classification symbols)						
	on searched other than minimum documentation to the extent that s						
	ata base consulted during the international search (name of data ba	se and, where practical, search terms use	id)				
C. DOCUME	ENTS CONSIDERED TO BE RELEVANT						
Category °	Citation of document, with indication, where appropriate, of the rel	evant passages	Relevant to claim No.				
А	EP 0 178 007 A (SHELL INT RESEARCE 16 April 1986 (1986-04-16) claims 1,5,12 page 5, line 23 - line 28	CH)	1,15,18,				
Funt	ner documents are listed in the continuation of box C.	X Patent family members are liste	d in annex.				
<ul> <li>Special categories of cited documents:</li> <li>"A" document defining the general state of the art which is not considered to be of particular relevance</li> <li>"E" earlier document but published on or after the international filing date</li> <li>"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)</li> <li>"O" document referring to an oral disclosure, use, exhibition or other means</li> <li>"P" document published prior to the international filing date but later than the priority date claimed</li> <li>Date of the actual completion of the international search</li> <li>"C" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention</li> <li>"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is combined with one or more other such document is combined with one or more other such document is combination being obvious to a person skilled in the art.</li> <li>"&amp;" document member of the same patent family</li> <li>Date of the actual completion of the international search report</li> <li>20 September 2000</li> </ul>			h the application but heory underlying the claimed invention of be considered to occurrent is taken alone claimed invention niventive step when the nore other such docuous to a person skilled at family				
ļ	mailing address of the ISA	Authorized officer					
	European Patent Office, P.B. 5818 Patentlaan 2 NL – 2280 HV Rijswijk Tel. (+31-70) 340–2040, Tx. 31 651 epo nl,	De Herdt, O					

# INTERNATIONAL SEARCH REPORT

Information on patent family members

Int Monal Application No PCT/US 00/12808

Patent document cited in search report	Publication date	Patent family member(s)		Publication date	
EP 0178007 A	16-04-1986	GB 216	5551 A	16-04-1986	
		AT 5	4120 T	15-07-1990	
		AU 570	0134 B	03-03-1988	
		AU 4830	6985 A	17-04-1986	
			4974 A	22-07-1986	
		CA 129	2619 A	03-12-1991	
		CN 8510	7305 A.B	09-07-1986	
		DE 3578	3411 D	02-08-1990	
		ES 547	7666 D	01-03-1986	
		ES 8604	4831 A	01-08-1986	
		IN 16	5968 A	17-02-1990	
		JP 210:	3709 C	22-10-1996	
		JP 7100	0796 B	01-11-1995	
			1292 A	09-05-1986	
			3732 A	29-03-1989	
			7734 A	28-05-1986	